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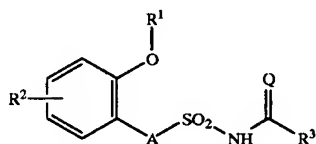
HERBICIDAL SULPHONYLAMINO(THIO)
CARBONYL COMPOUNDS

This is a Continuation of U.S. Ser. No. 09/006,686, filed on Jan. 8, 1998 (now abandoned); which is a Continuation-in-Part of International Application PCT/EP96/02826, filed on Jun. 28, 1996.

The invention relates to novel sulphonylamino(thio)carbonyl compounds, to a number of processes and to novel intermediates for their preparation, and to their use as herbicides.

It is already known that certain sulphonylaminocarbonyl compounds, such as, for example, the compounds 4,5-dimethoxy-2-(2-methoxy-phenylsulphonylamino-carbonyl)-2,4-dihydro-3H-1,2,4-triazol-3-one, 4,5-dimethoxy-2-(2,5-dimethoxy-phenylsulphonylaminocarbonyl)-2,4-dihydro-3H-1,2,4-triazol-3-one, 4,5-diethoxy-2-(2,5-dimethoxy-phenylsulphonylaminocarbonyl)-2,4-dihydro-3H-1,2,4-triazol-3-one and N-(2,5-dimethoxy-phenylsulphonyl)-1,5-dimethyl-1H-pyrazole-3-carboxamide possess herbicidal properties (cf. EP 341489, EP 422469, EP 425948, EP 431291, EP 507171, EP 534266, DE 4029753). The action of these compounds, however, is not in every respect satisfactory.

The novel sulphonylamino(thio)carbonyl compounds have now been found of the ageneral formula (I),



in which

A represents a single bond, oxygen, sulphur or the group N—R, in which R represents hydrogen, alkyl, alkenyl, alkynyl or cycloalkyl,

Q represents oxygen or sulphur,

R¹ represents hydrogen or formyl or represents in each case optionally substituted alkyl, alkenyl, alkynyl, alkylcarbonyl, alkoxy-carbonyl, alkylsulphonyl, cycloalkyl, cycloalkylcarbonyl or cycloalkylsulphonyl,

R² represents cyano or halogen or represents in each case optionally substituted alkyl, alkenyl, alkynyl, alkoxy, alkenyloxy or alkinyloxy, and

R³ represents in each case optionally substituted heterocycl having 5 ring members of which at least one is oxygen, sulphur or nitrogen and from one to three further ring members can be nitrogen,

and salts of compounds of the formula (I),

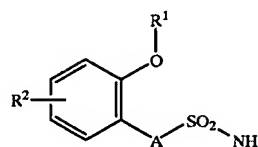
the previously known compounds 4,5-dimethoxy-2-(2,5-dimethoxy-phenylsulphonylaminocarbonyl)-2,4-dihydro-3H-1,2,4-triazol-3-one, 4,5-diethoxy-2-(2,5-dimethoxy-phenylsulphonylaminocarbonyl)-2,4-dihydro-3H-1,2,4-triazol-3-one and N-(2,5-dimethoxy-phenylsulphonyl)-1,5-dimethyl-1H-pyrazole-3-carboxamide being excluded by disclaimer.

The novel sulphonylamino(thio)carbonyl compounds of the general formula (I) are obtained if

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(a) aminosulphonyl compounds of the general formula (II)

(II)



in which

A, R¹ and R² have the meanings given above are reacted with (thio)carboxylic acid derivatives of the general formula (III)

(III)



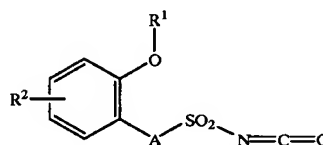
in which

Q and R³ have the meanings given above and Z represents halogen, alkoxy, aryloxy or arylalkoxy, optionally in the presence of an acid acceptor and optionally in the presence of a diluent,

or if

(b) sulphonyl iso(thio)cyanates of the general formula (IV)

(IV)



in which

A, Q, R¹ and R² have the meanings given above are reacted with heterocycles of the general formula (V)



(V)

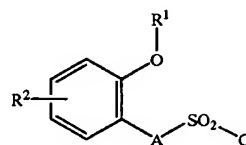
in which

R³ has the meaning given above, optionally in the presence of a reaction auxiliary and optionally in the presence of a diluent,

or if

(c) chlorosulphonyl compounds of the general formula (VI)

(VI)



in which

A, R¹ and R² have the meanings given above are reacted with heterocycles of the general formula (V)



(V)

in which

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R^3 has the meaning given above
and metal (thio)cyanates of the general formula (VII)



in which

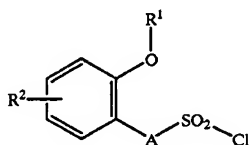
Q has the meaning given above, and

M represents an alkali metal or alkaline earth metal equivalent,

optionally in the presence of a reaction auxiliary and 10
optionally in the presence of a diluent,

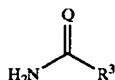
or if

(d) chlorosulphonyl compounds of the general formula (VI)



in which

A, R^1 and R^2 have the meanings given above
are reacted with (thio)carboxamides of the general formula 15
(VIII)



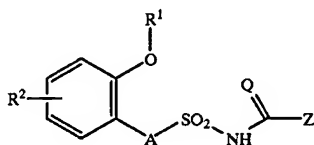
in which

Q and R^3 have the meanings given above,

optionally in the presence of an acid acceptor and option-
ally in the presence of a diluent,

or if

(e) sulphonylamino(thio)carbonyl compounds of the general 20
formula (IX)



in which

A, Q, R^1 and R^2 have the meanings given above, and

Z represents halogen, alkoxy, aryloxy or arylalkoxy,
are reacted with heterocycles of the general formula (V)



in which

R^3 has the meaning given above,

optionally in the presence of an acid acceptor and option-
ally in the presence of a diluent,

or if

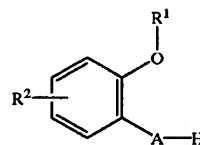
(f) heterocycles of the general formula (V)



in which

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R^3 has the meaning given above,
are reacted with chlorosulphonyl iso(thio)cyanate, option-
ally in the presence of a diluent, and the adducts formed in
this reaction are reacted in situ with benzene derivatives of
the general formula (X) 5



(X)

in which

A, R^1 and R^2 have the meanings given above,

optionally in the presence of an acid acceptor and option-
ally in the presence of a diluent,

20 and, if desired, the compounds of the formula (I) obtained
by processes (a), (b), (c), (d), (e) or (f) are converted
into salts by customary methods.

The novel sulphonylamino(thio)carbonyl compounds of
the general formula (I) are distinguished by a strong herbi-
cidal activity. 25

The invention relates preferably to compounds of the
formula (I) in which

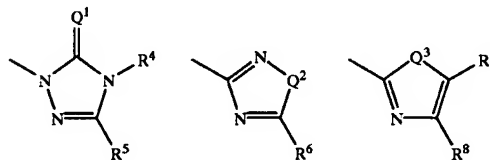
A represents a single bond, oxygen, sulphur or the group
N—R, in which R represents hydrogen, C_1 – C_6 -alkyl,
 C_2 – C_6 -alkenyl, C_2 – C_6 -alkinyl or C_3 – C_6 -cycloalkyl,

Q represents oxygen or sulphur,

R^1 represents hydrogen or formyl or represents in each
case optionally cyano-, fluoro-, chloro-, bromo-,
phenyl- or C_1 – C_4 -alkoxy-substituted alkyl, alkenyl,
alkinyl, alkylcarbonyl, alkoxy carbonyl or alkylsulpho-
nyl having in each case up to 6 carbon atoms, or
represents in each case optionally cyano-, fluoro-,
chloro-, bromo- or C_1 – C_4 -alkyl-substituted C_3 – C_6 -
cycloalkyl, C_3 – C_6 -cycloalkyl-carbonyl or C_3 – C_6 -
cycloalkyl-sulphonyl, 35

R^2 represents cyano, fluoro, chloro or bromo or represents
in each case optionally cyano-, fluoro-, chloro-, bromo-
or C_1 – C_4 -alkoxy-substituted alkyl, alkenyl, alkinyl,
alkoxy, alkenyloxy or alkinyloxy having in each case
up to 6 carbon atoms, and

R^3 represents in each case optionally substituted hetero-
cyclyl of the formulae below, 40



in which

Q^1 , Q^2 and Q^3 each represent oxygen or sulphur, and

60 R^4 represents hydrogen, hydroxyl, amino or cyano, or
represents C_2 – C_{10} -alkylideneamino, or represents
optionally fluoro-, chloro-, bromo-, cyano-, C_1 – C_4 -
alkoxy-, C_1 – C_4 -alkyl-carbonyl- or C_1 – C_4 -alkoxy-
carbonyl-substituted C_1 – C_6 -alkyl, or represents in each
case optionally fluoro-, chloro- and/or bromo-
substituted C_2 – C_6 -alkenyl or C_2 – C_6 -alkinyl, or repre-
sents in each case optionally fluoro-, chloro-, bromo-, 65

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cyano-, C₁-C₄-alkoxy- or C₁-C₄-alkoxy-carbonyl-substituted C₁-C₆-alkoxy, C₁-C₆-alkylamino or C₁-C₆-alkyl-carbonylamino, or represents C₃-C₆-alkenyl, or represents di-(C₁-C₄-alkyl)-amino, or represents in each case optionally fluoro-, chloro-, bromo-, cyano- and/or C₁-C₄-alkyl-substituted C₃-C₆-cycloalkyl, C₃-C₆-cycloalkylamino or C₃-C₆-cycloalkyl-C₁-C₄-alkyl, or represents in each case optionally fluoro-, chloro-, bromo-, cyano-, nitro-, C₁-C₄-alkyl-, trifluoromethyl- and/or C₁-C₄-alkoxy-substituted phenyl, phenylamino or phenyl-C₁-C₄-alkyl,

R⁵ represents hydrogen, hydroxyl, mercapto, amino, cyano, fluoro, chloro, bromo or iodo, or represents optionally fluoro-, chloro-, bromo-, cyano-, C₁-C₄-alkoxy-, C₁-C₄-alkyl-carbonyl- or C₁-C₄-alkoxy-carbonyl-substituted C₁-C₆-alkyl, or represents in each case optionally fluoro-, chloro- and/or bromo-substituted C₂-C₆-alkenyl or C₂-C₆-alkinyl, or represents in each case optionally fluoro-, chloro-, cyano-, C₁-C₄-alkoxy- or C₁-C₄-alkoxy-carbonyl-substituted C₁-C₆-alkoxy, C₁-C₆-alkylthio, C₁-C₆-alkylamino or C₁-C₆-alkyl-carbonylamino, or represents C₃-C₆-alkenyl, C₃-C₆-alkenylthio, C₃-C₆-alkenylthio, C₃-C₆-alkenylamino or C₃-C₆-alkinylamino, or represents di-(C₁-C₄-alkyl)-amino, or represents in each case optionally methyl- and/or ethyl-substituted aziridino, pyrrolidino, piperidino or morpholino, or represents in each case optionally fluoro-, chloro-, bromo-, cyano- and/or C₁-C₄-alkyl-substituted C₃-C₆-cycloalkyl, C₃-C₆-cycloalkenyl, C₃-C₆-cycloalkylthio, C₃-C₆-cycloalkylthio, C₃-C₆-cycloalkylamino C₃-C₆-cycloalkyl-C₁-C₄-alkyl, C₃-C₆-cycloalkyl-C₁-C₄-alkoxy, C₃-C₆-cycloalkyl-C₁-C₄-alkylthio or C₃-C₆-cycloalkyl-C₁-C₄-alkylamino, or represents in each case optionally fluoro-, chloro-, bromo-, cyano-, nitro-, C₁-C₄-alkyl-, trifluoromethyl-, C₁-C₄-alkoxy- and/or C₁-C₄-alkoxy-carbonyl-substituted phenyl, phenyl-C₁-C₄-alkyl, phenoxy, phenyl-C₁-C₄-alkoxy, phenylthio, phenyl-C₁-C₄-alkylthio, phenylamino or phenyl-C₁-C₄-alkylamino, or

R⁴ and R⁵ together represent optionally branched alkanediyl having 3 to 11 carbon atoms, and also

R⁶, R⁷ and R⁸ are identical or different and represent hydrogen, cyano, fluoro, chloro or bromo, or represent in each case optionally fluoro-, chloro-, bromo- or C₁-C₄-alkoxy-substituted alkyl, alkenyl, alkynyl, alkoxy, alkenyloxy, alkynyloxy, alkylthio, alkenylthio, alkynylthio, alkylsulphonyl and alkylsulphonyl having in each case up to 6 carbon atoms, or represents in each case optionally cyano-, fluoro-, chloro-, bromo- or C₁-C₄-alkyl-substituted cycloalkyl having 3 to 6 carbon atoms,

the previously known compounds 4,5-dimethoxy-2-(2,5-dimethoxy-phenylsulphonylaminocarbonyl)-2,4-dihydro-3H-1,2,4-triazol-3-one, 4,5-diethoxy-2-(2,5-dimethoxy-phenylsulphonylaminocarbonyl)-2,4-dihydro-3H-1,2,4-triazol-3-one and N-(2,5-dimethoxy-phenylsulphonyl)-1,5-dimethyl-1H-pyrazole-3-carboxamide being excluded by disclaimer.

The invention also relates preferably to sodium, potassium, magnesium, calcium, ammonium, C₁-C₄-alkylammonium, di-(C₁-C₄-alkyl)-ammonium, tri-(C₁-C₄-alkyl)-ammonium, tetra-(C₁-C₄-alkyl)-ammonium, tri-(C₁-C₄-alkyl)-sulphonium, C₅- or C₆-cycloalkyl-ammonium and di-(C₁-C₂-alkyl)-benzylammonium salts

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of compounds of the formula (I) in which A, Q, R¹, R² and R³ have the meanings indicated above as preferred.

The invention relates in particular to compounds of the formula (I) in which

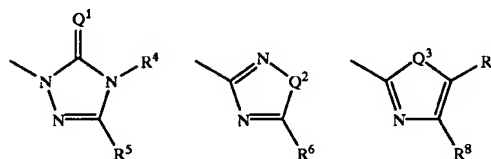
A represents a single bond, oxygen or the group N—R, in which R represents hydrogen, methyl, ethyl, n- or i-propyl, n-, i- or s-butyl, propenyl, butenyl, propinyl, butinyl, cyclopropyl, cyclobutyl, cyclopentyl or cyclohexyl,

Q represents oxygen or sulphur,

R¹ represents hydrogen or formyl, or represents in each case optionally fluoro-, chloro-, bromo-, methoxy- or ethoxy-substituted methyl, ethyl, n- or i-propyl, n-, i- or s-butyl, propenyl, butenyl, propinyl, butinyl, acetyl, propionyl, butyryl, methoxycarbonyl, ethoxycarbonyl, n- or i-propoxycarbonyl, methylsulphonyl, ethylsulphonyl, n- or i-propylsulphonyl, n-, i-, s- or t-butylsulphonyl, or represents in each case optionally fluoro-, chloro- or methyl-substituted cyclopropyl, cyclopropylcarbonyl or cyclopropylsulphonyl,

R² represents cyano, fluoro, chloro or bromo, or represents in each case optionally fluoro-, chloro-, methoxy- or ethoxy-substituted methyl, ethyl, n- or i-propyl, n-, i- or s-butyl, propenyl, butenyl, propinyl, butinyl, methoxy, ethoxy, n- or i-propoxy, n-, i- or s-butoxy, propenyloxy, butenyloxy, propinyloxy or butinyloxy and

R³ represents in each case optionally substituted heterocyclyl of the formulae below,



in which

Q¹, Q² and Q³ each represent oxygen or sulphur, and R⁴ represents hydrogen, hydroxyl or amino, or represents C₃-C₆-alkyl-deneamino, or represents in each case optionally fluoro-, chloro-, cyano-, methoxy- or ethoxy-substituted methyl, ethyl, n- or i-propyl, n-, i-, s- or t-butyl, or represents in each case optionally fluoro-, chloro- or bromo-substituted propenyl, butenyl, propinyl or butinyl, or represents in each case optionally fluoro-, chloro-, cyano-, methoxy- or ethoxy-substituted methoxy, ethoxy, n- or i-propoxy, n-, i-, s- or t-butoxy, methylamino, ethylamino, n- or i-propylamino, n-, i-, s- or t-butylamino, or represents propenyloxy or butenyloxy, or represents dimethylamino or diethylamino, or represents in each case optionally fluoro-, chloro-, methyl- and/or ethyl-substituted cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl, cyclopropylamino, cyclobutylamino, cyclopentylamino, cyclohexylamino, cyclopropylmethyl, cyclobutylmethyl, cyclopentylmethyl or cyclohexylmethyl, or represents in each case optionally fluoro-, chloro-, methyl-, trifluoromethyl- and/or methoxy-substituted phenyl or benzyl,

R⁵ represents hydrogen, hydroxyl, mercapto, amino, fluoro, chloro or bromo, or represents in each case optionally fluoro-, chloro-, cyano-, methoxy- or ethoxy-substituted methyl, ethyl, n- or i-propyl, n-, i-,

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s- or t-butyl, or represents in each case optionally fluoro-, chloro- or bromo-substituted ethenyl, propenyl, butenyl, propinyl or butinyl, or represents in each case optionally fluoro-, chloro-, cyano-, methoxy- or ethoxy-substituted methoxy, ethoxy, n- or i-propoxy, n-, i-, s- or t-butoxy, methylthio, ethylthio, n- or i-propylthio, n-, i-, s- or t-butylthio, methylamino, ethylamino, n- or i-propylamino, n-, i-, s- or t-butylamino, or represents propenyloxy, butenyloxy, propinyloxy, butinyloxy, propenylthio, propadienylthio, butenylthio, propinylthio, butinylthio, propenylamino, butenylamino, propinylamino or butinylamino, or represents dimethylamino, diethylamino or dipropylamino, or represents in each case optionally fluoro-, chloro-, methyl- and/or ethyl-substituted cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl, cyclopentenyl, cyclohexenyl, cyclopropyloxy, cyclobutyloxy, cyclopentyloxy, cyclohexyloxy, cyclopropylthio, cyclobutylthio, cyclopentylthio, cyclohexylthio, cyclopropylamino, cyclobutylamino, cyclopentylamino, cyclohexylamino, cyclopropylmethyl, cyclobutylmethyl, cyclopentylmethyl, cyclohexylmethyl, cyclopropylmethoxy, cyclobutylmethoxy, cyclopentylmethoxy, cyclohexylmethoxy, cyclopropylmethylthio, cyclobutylmethylthio, cyclopentylmethylthio, cyclohexylmethylthio, cyclopropylmethylamino, cyclobutylmethylamino, cyclopentylmethylamino or cyclohexylmethylamino, or represents in each case optionally fluoro-, chloro-, methyl-, trifluoromethyl-, methoxy- and/or methoxycarbonyl substituted phenyl, benzyl, phenoxy, benzoyloxy, phenylthio, benzylthio, phenylamino or benzylamino, or

R⁴ and R⁵ together represent optionally branched alkanediyl having 3 to 11 carbon atoms, and also

R⁶, R⁷ and R⁸ are identical or different and represent hydrogen, cyano, fluoro, chloro or bromo, or represent in each case optionally fluoro-, chloro-, methoxy- or ethoxy-substituted methyl, ethyl, n- or i-propyl, n-, i-, s- or t-butyl, propenyl, butenyl, propinyl, butinyl, methoxy, ethoxy, n- or i-propoxy, n-, i-, s- or t-butoxy, propenyloxy, butenyloxy, propinyloxy, butinyloxy, methylthio, ethylthio, n- or i-propylthio, n-, i-, s- or t-butylthio, propenylthio, butenylthio, propinylthio, butinylthio, methylsulphonyl, ethylsulphonyl, methylsulphonyl or ethylsulphonyl, or represent cyclopropyl, the previously known compounds 4,5-dimethoxy-2-(2,5-dimethoxy-phenylsulphonylamino-carbonyl)-2,4-dihydro-3H-1,2,4-triazol-3-one, 4,5-diethoxy-2-(2,5-dimethoxy-phenylsulphonylamino-carbonyl)-2,4-dihydro-3H-1,2,4-triazol-3-one and N-(2,5-dimethoxy-phenylsulphonyl)-1,5-dimethyl-1H-pyrazole-3-carboxamide being excluded by disclaimer.

A very particularly preferred group of compounds according to the invention are the compounds of the formula (I), in which

A represents a single bond,

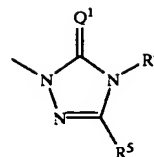
Q represents oxygen or sulphur,

R¹ represents methyl, ethyl, n- or i-propyl,

R² represents chloro or methyl- in each case in position 5 or 6- and

R³ represents optionally substituted triazolinyl of the formula below

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in which

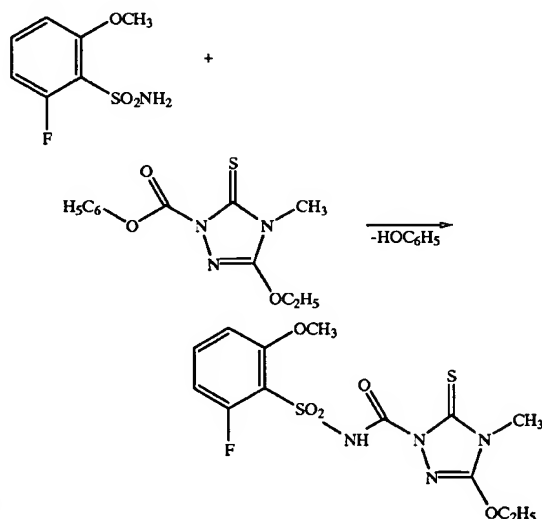
Q¹ represents oxygen or sulphur, and

R⁴ represents in each case optionally fluoro-, chloro-, cyano-, methoxy- or ethoxy-substituted methyl, ethyl, n- or i-propyl, or represents propenyl or propinyl, or represents methoxy, ethoxy, n- or i-propoxy, or represents cyclopropyl, and

R⁵ represents hydrogen, chloro or bromo, or represents in each case optionally fluoro-, chloro-, cyano-, methoxy- or ethoxy-substituted methyl, ethyl, n- or i-propyl, or represents in each case optionally fluoro and/or chloro-substituted propenyl or propinyl, or represents in each case optionally fluoro-, chloro-, cyano-, methoxy- or ethoxy-substituted methoxy, ethoxy, n- or i-propoxy, methylthio, ethylthio, n- or i-propylthio, or represents propenyloxy or cyclopropyl.

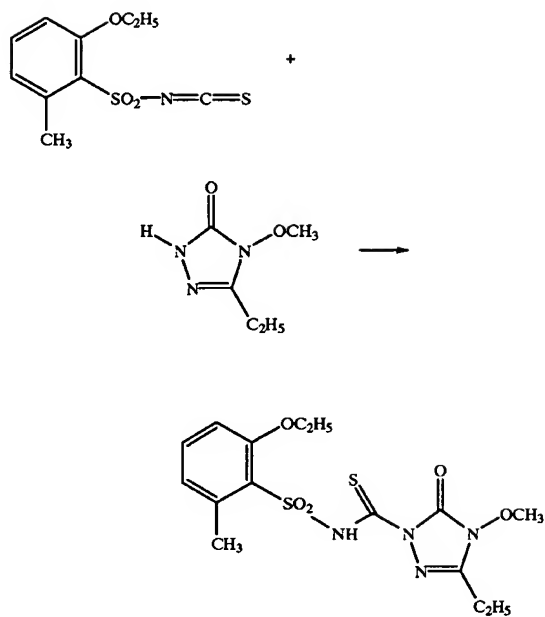
The radical definitions listed above, whether general or listed in ranges of preference, apply not only to the end products of the formula (I) but also, correspondingly, to the starting materials and/or intermediates required in each case of the preparation. These radical definitions can be combined as desired with one another, thus including combinations between the preferred ranges indicated.

Using, for example, 2-fluoro-6-methoxy-benzenesulphonamide and 5-ethoxy-4-methyl-2-phenoxycarbonyl-2,4-dihydro-3H-1,2,4-triazole-3-thione as starting materials, the course of reaction in the process (a) according to the invention can be illustrated by the following equation:

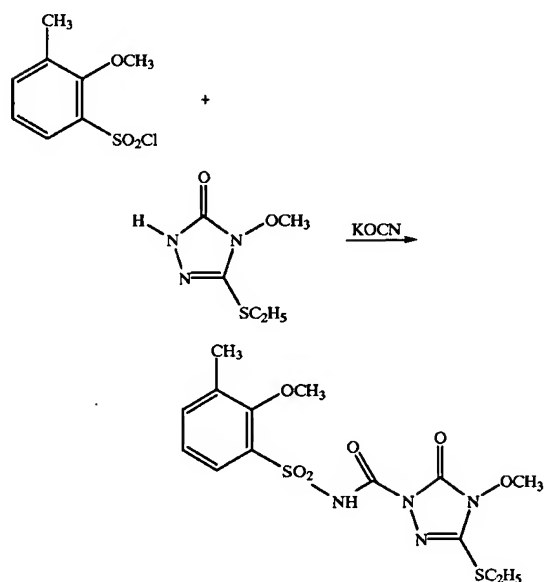


Using, for example, 2-ethoxy-6-methyl-phenylsulphonyl-isothiocyanate and 5-ethyl-4-methoxy-2,4-dihydro-3H-1,2,4-triazol-3-one as starting materials, the course of reaction in the process (b) according to the invention can be illustrated by the following equation:

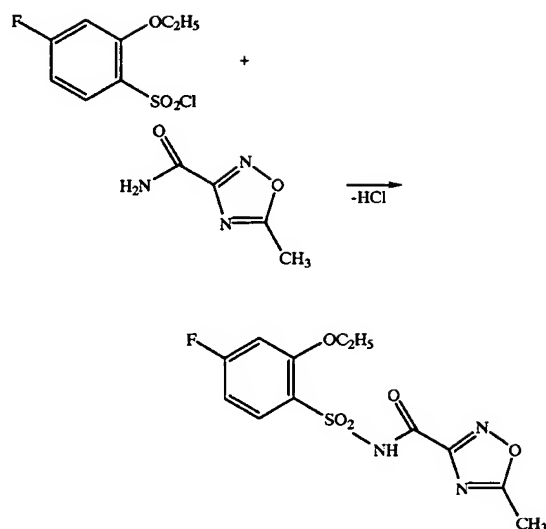
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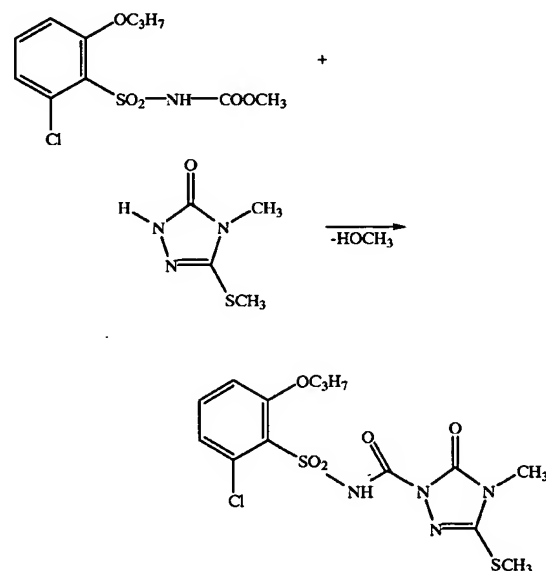
Using, for example, 2-ethoxy-6-methylbenzenesulphochloride, 5-ethylthio-4-methoxy-2,4-dihydro-3H-1,2,4-triazol-3-one and potassium cyanate as starting materials, the course of reaction in the process (c) according to the invention can be illustrated by the following equation:



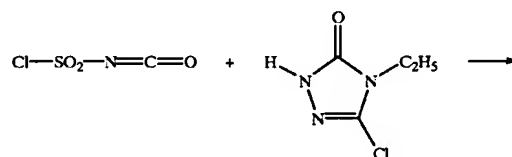
Using, for example, 2-ethoxy-4-fluorobenzenesulphochloride and 5-methyl-1,2,4-oxadiazole-3-carboxamide as starting materials, the course of reaction in the process (d) according to the invention can be illustrated by the following equation:



Using, for example, N-(2-chloro-6-propoxyphenylsulphonyl)-O-methyl-urethane and 4-methyl-5-methylthio-2,4-dihydro-3H-1,2,4-triazol-3-one as starting materials, the course of reaction in the process (e) according to the invention can be illustrated by the following equation:

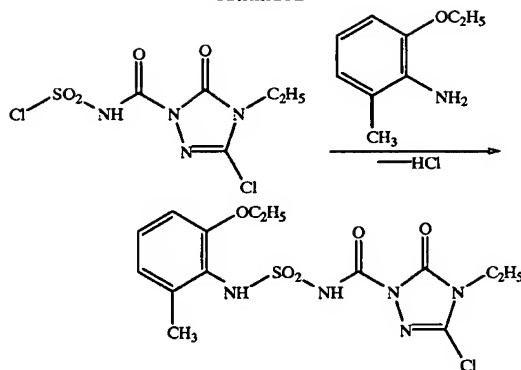


Using, for example, 5-chloro-4-ethyl-2,4-dihydro-3H-1,2,4-triazol-3-one and chlorosulphonylisocyanate and then 2-ethoxy-6-methyl-aniline as starting materials, the course of reaction in the process (f) according to the invention can be illustrated by the following equation:



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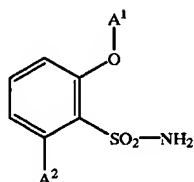
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A general definition of the aminosulphonyl compounds to be used as starting materials in the process (a) according to the invention for the preparation of compounds of the formula (I) is given by the formula (II). In the formula (II) A, R¹ and R² preferably or in particular have that meaning which has already been indicated above, in connection with the description of the compounds of the formula (I) to be prepared in accordance with the invention, as being preferable or, respectively, particularly preferable for A, R¹ and R².

The starting materials of the formula (II) are known and/or can be prepared by methods known per se (cf. EP 216504, DE 3208189, EP 44807, EP 23422).

Compounds not yet known from the literature, and which as novel substances are likewise a subject of the present application, are the sulphonamides of the general formula (IIa),

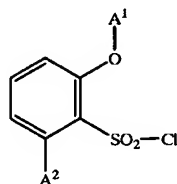


in which

A¹ represents ethyl, n- or i-propyl, n-, i-, s- or t-butyl, trifluoromethyl, fluoroethyl, chloroethyl, difluoroethyl, trifluoroethyl, chlorotrifluoroethyl, methoxyethyl, ethoxyethyl, allyl, propargyl or benzyl, and

A² represents methyl, ethyl, n- or i-propyl, n-, i-, s- or t-butyl.

The novel sulphonamides of the formula (IIa) are obtained if sulphonyl chlorides of the formula (VIa)



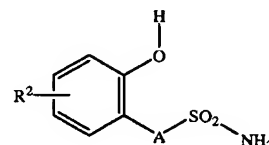
in which

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A¹ and A² have the meanings given above

are reacted with ammonia, optionally in the presence of a diluent, for example water, at temperatures between 0° C. and 50° C. (cf. the Preparation Examples).

The starting materials of the formula (II) can in general be obtained also by reacting phenol derivatives of the formula (IIb)



in which

A and R² have the meanings given above

with alkylating agents of the formula (XI)



in which

R¹ has the meaning given above, and

X represents halogen or the group R¹-O-SO₂-O-, optionally in the presence of an acid acceptor, for example potassium carbonate, and optionally in the presence of a diluent, for example toluene, at temperatures between 10° C. and 150° C. (cf. the Preparation Examples).

The phenol derivatives of the formula (IIb) required as precursors are known and/or can be prepared by methods known per se (cf. EP 44807, Metalloberfläche [Metal surface]—Angew. Elektrochemie 27 (1973), 217–227—cited in Chem. Abstracts 79:86733; Preparation Examples).

The alkylating agents of the formula (XI) which are also required as precursors are known synthesis chemicals.

A general definition of the (thio)carboxylic acid derivatives also to be used as starting materials in the process (a) according to the invention for the preparation of the compounds of the formula (I) is given by the formula (III). In the formula (III), Q and R³ preferably or in particular have that meaning which has already been indicated above, in connection with the description of the compounds of the formula (I) to be prepared in accordance with the invention, as being preferable or, respectively, particularly preferable for Q and R³; Z preferably represents fluorine, chlorine, bromine, C₁–C₄-alkoxy, phenoxy or benzyloxy, and in particular represents chlorine, methoxy, ethoxy or phenoxy.

The starting materials of the formula (III) are known and/or can be prepared by methods known per se (cf. EP 459244, EP 341489, EP 422469, EP 425948, EP 431291, EP 507171, EP 534266).

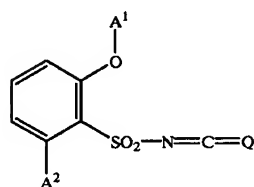
A general definition of the sulphonyl iso(thio)cyanate to be used as starting materials in the process (b) according to the invention for the preparation of the compounds of the formula (I) is given with the formula (IV). In the formula (IV), A, Q, R¹ and R² preferably or in particular have that meaning which has already been indicated above, in connection with the description of the compounds of the formula (I) to be prepared in accordance with the invention, as being preferable or particularly preferable for A, Q, R¹ and R².

The starting materials of the formula (IV) are known and/or can be prepared by methods known per se (cf. EP 23422, EP 216504).

Compounds not yet known from the literature, and which as novel substances are likewise a subject of the present

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application, are the sulphonyl iso(thio)cyanates of the general formula (IVa)



in which

Q represents oxygen or sulphur,

A¹ represents ethyl, n- or i-propyl, n-, i-, s- or t-butyl, trifluoromethyl, fluoroethyl, chloroethyl, difluoroethyl, trifluoroethyl, chlorotrifluoroethyl, methoxyethyl, ethoxyethyl, allyl, propargyl or benzyl, and

A² represents methyl, ethyl, n- or i-propyl, n-, i-, s- or t-butyl.

The novel sulphonyl iso(thio)cyanates of the formula (IVa) are obtained if sulphonamides of the formula (IIa)—above—are reacted with phosgene or, respectively, thiophosgene, optionally in the presence of an alkyl isocyanate, for example butyl isocyanate, optionally in the presence of a reaction auxiliary, for example diazabicyclo [2.2.2]octane, and in the presence of a diluent, for example toluene, xylene or chlorobenzene, at temperatures between 80° C. and 150° C., and, after the end of the reaction, the volatile components are distilled off under reduced pressure.

A general definition of the heterocycles also to be used as starting materials in the processes (b), (c), (e) and (f) according to the invention for the preparation of the compounds of the formula (I) is given by the formula (V). In the formula (V), R³ preferably or in particular has that meaning which has already been indicated above, in connection with the description of the compounds of the formula (I) to be prepared in accordance with the invention, as being preferable or particularly preferable for R³.

The starting materials of the formula (V) are known and/or can be prepared by methods known per se (cf. EP 341489, EP 422469, EP 425948, EP 431291, EP 507171, EP 534266).

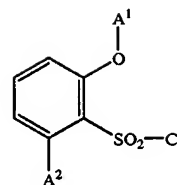
A general definition of the chlorosulphonyl compounds to be used as starting materials in the processes (c) and (d) according to the invention for the preparation of compounds of the formula (I) is given by the formula (VI). In the formula (VI), A, R¹ and R² preferably or in particular have that meaning which has already been indicated above, in connection with the description of the compounds of the formula (I) to be prepared in accordance with the invention, as being preferable or particularly preferable for A, R¹ and R².

The starting materials of the formula (VI) are known and/or can be prepared by methods known per se (cf. EP 511826, DE 3208189, EP23422).

Compounds not yet known from the literature, which as novel substances are likewise a subject of the present application, are the sulphonyl chlorides of the formula (VIa)

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(VIa)

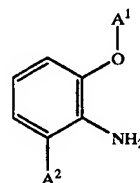


in which

A¹ represents ethyl, n- or i-propyl, n-, i-, s- or t-butyl, trifluoromethyl, fluoroethyl, chloroethyl, difluoroethyl, trifluoroethyl, chlorotrifluoroethyl, methoxyethyl, ethoxyethyl, allyl, propargyl or benzyl, and

A² represents methyl, ethyl, n- or i-propyl, n-, i-, s- or t-butyl.

The novel sulphonyl chlorides of the formula (VIa) are obtained if aniline derivatives of the formula (XII)

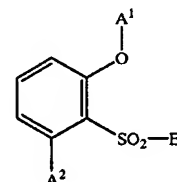


in which

A¹ and A² have the meanings given above
dare reacted with an alkali metal nitrite, for example sodium nitrite, in the presence of hydrochloric acid at temperatures between -10° C. and +10° C., and the diazonium salt solution thus obtained is reacted with sulphur dioxide in the presence of a diluent, for example dichloromethane or 1,2-dichloro-ethane, and in the presence of a catalyst, for example copper(I) chloride, optionally in the presence of a further catalyst, for example dodecyltrimethylammonium bromide, at temperatures between -10° C. and +50° C. (cf. the Preparation Examples).

The aniline derivatives of the formula (XII) required as precursors are known and/or can be prepared by methods known per se (cf. EP 511826, US 4992091, EP 185128, DE 2405479, Preparation Examples).

The abovementioned novel benzenesulphonic acid derivatives of the formulae (IIa), (IVa) and (VIa) can be defined comprehensively by the following formula (XIII):



in which

E represents -NH₂, -N=C+Q or -Cl, where

Q represents O or S, and also

A¹ represents ethyl, n- or i-propyl, n-, i-, s- or t-butyl, trifluoromethyl, fluoroethyl, chloroethyl, difluoroethyl,

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trifluoroethyl, chlorotrifluoroethyl, methoxyethyl, ethoxyethyl, allyl, propargyl or benzyl, and A² represents methyl, ethyl, n- or i-propyl, n-, i-, s- or t-butyl.

A general definition of the (thio)carboxamides to be used as starting materials in the process (d) according to the invention for the preparation of the compounds of the formula (I) is given by the formula (VIII). In the formula (VIII), Q and R³ preferably or in particular have that meaning which has already been indicated above, in connection with a description of the compounds of the formula (I) to be prepared in accordance with the invention, as being preferable or particularly preferable for Q and R³.

The starting materials of the formula (VIII) are known and/or can be prepared by methods known per se (cf. EP 459244).

A general definition of the sulphonylamino(thio)carbonyl compounds to be used as starting materials in the process (e) according to the invention for the preparation of the compounds of the formula (I) is given by the formula (IX). In the formula (IX), A, Q, R¹ and R² preferably or in particular have that meaning which has already been indicated above, in connection with the description of the compounds of the formula (I) to be prepared in accordance with the invention, as being preferable or particularly preferable for A, Q, R¹ and R²; Z preferably represents fluoro, chloro, bromo, C₁-C₄-alkoxy, phenoxy or benzyloxy, and in particular represents chlorine, methoxy, ethoxy or phenoxy.

A general definition of the benzene derivatives to be used as starting materials in the process (f) according to the invention for the preparation of the compounds of the formula (I) is given by the formula (X). In the formula (X), A, R¹ and R² preferably or in particular have that meaning which has already been indicated above, in connection with the description of the compounds of the formula (I) to be prepared in accordance with the invention, as being preferable or particularly preferable for A, R¹ and R².

Starting materials of the formula (X) are known and/or can be prepared by methods known per se (cf. EP 511826, U.S. Pat. No. 4,992,091, EP 185128, DE 2405479, Preparation Examples).

The processes (a), (b), (c), (d), (e) and (f) according to the invention for the preparation of the novel compounds of the formula (I) are preferably carried out using diluents. Suitable diluents in this context are virtually all inert organic solvents. These include, preferably, aliphatic and aromatic, optionally halogenated hydrocarbons such as pentane, hexane, heptane, cyclohexane, petroleum ether, benzene, ligroin, benzene, toluene, xylene, methylene chloride, ethylene chloride, chloroform, tetrachloromethane, chlorobenzene and o-dichlorobenzene; ethers such as diethyl ether and dibutyl ether, glycol dimethyl ether and diglycol dimethyl ether, tetrahydrofuran and dioxanes; ketones such as acetone, methyl ethyl ketone, methyl isopropyl ketone and methyl isobutyl ketone; esters such as methyl acetate and ethyl acetate; nitrites, for example acetonitrile and propionitrile; amides, for example dimethylformamide, dimethylacetamide and N-methylpyrrolidone, and also dimethyl sulphoxide, tetramethylene sulphone and hexamethylphosphoric triamide.

As reaction auxiliaries and/or as acid acceptors in the processes (a), (b), (c), (d), (e) and (f) according to the invention it is possible to employ all acid-binding agents which can customarily be used for such reactions. Preferred among suitable examples are alkali metal hydroxides, for example sodium hydroxide and potassium hydroxide, alkaline earth metal hydroxides, for example calcium hydroxide, alkali metal carbonates and alcoholates, such as sodium carbonate and potassium carbonate, sodium tert-butyrate and potassium tert-butyrate, and also basic nitrogen compounds, such as trimethylamine, triethylamine, tripropylamine,

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tributylamine, diisobutylamine, dicyclohexylamine, ethyldiisopropylamine, ethyldicyclohexylamine, N,N-dimethylbenzylamine, N,N-dimethyl-aniline, pyridine, 2-methyl-, 3-methyl-, 4-methyl-, 2,4-dimethyl-, 2,6-dimethyl-, 2-ethyl-, 4-ethyl- and 5-ethyl-2-methyl-pyridine, 1,5-diazabicyclo[4.3.0]non-5-ene (DBN), 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU) and 1,4-diazabicyclo-[2.2.2]-octane (DABCO).

The reaction temperatures in the processes (a), (b), (c), (d), (e) and (f) according to the invention can be varied within a relatively wide range. They are in general carried out at temperatures of between -20° C. and +150° C., preferably at temperatures between 0° C. and +100° C.

The processes (a), (b), (c), (d), (e) and (f) according to the invention are generally carried out under atmospheric pressure. However it is also possible to operate under increased or reduced pressure.

For carrying out processes (a), (b), (c), (d), (e) and (f) according to the invention, the starting materials required in each case are in general employed in approximately equimolar quantities. However, it is also possible to use one of the components employed in each case in a relatively large excess. The reactions are in general carried out in a suitable diluent in the presence of an acid acceptor, and the reaction mixture is stirred for a number of hours at the particular temperature required. Working up in the case of the processes (a), (b), (c), (d), (e) and (f) according to the invention is in each case by customary methods (cf. the Preparation Examples).

Salts of the compounds of the general formula (I) according to the invention can be prepared if desired. Such salts are obtained in a simple manner by customary methods of forming salts, for example by dissolving or dispersing a compound of the formula (I) in an appropriate solvent, for example methylene chloride, acetone, tert-butyl methyl ether or toluene, and adding an appropriate base. The salts can then—if desired after prolonged stirring—be isolated by concentration or filtration with suction.

The active compounds according to the invention can be used as defoliants, desiccants, agents for destroying broad-leaved plants and, especially, as weed-killers. By weeds, in the broadest sense, there are to be understood all plants which grow in locations where they are not wanted. Whether the substances according to the invention act as total or selective herbicides depends essentially on the amount used.

The active compounds according to the invention can be used, for example, in connection with the following plants:

Dicotyledon weeds of the genera: Sinapis, Lepidium, Galium, Stellaria, Matricaria, Anthemis, Galinsoga, Chenopodium, Urtica, Senecio, Amaranthus, Portulaca, Xanthium, Convolvulus, Ipomoea, Polygonum, Sesbania, Ambrosia, Cirsium, Carduus, Sonchus, Solanum, Rorippa, Rotalia, Lindernia, Lamium, Veronica, Abutilon, Emex, Datura, Viola, Galeopsis, Papaver, Centaurea, Trifolium, Ranunculus and Taraxacum.

Dicotyledon cultures of the genera: Gossypium, Glycine, Beta, Daucus, Phaseolus, Pisum, Solanum, Linum, Ipomoea, Vicia, Nicotiana, Lycopersicon, Arachis, Brassica, Lactuca, Cucumis and Cucurbita.

Monocotyledon weeds of the genera: Echinochloa, Setaria, Panicum, Digitaria, Phleum, Poa, Festuca, Eleusine, Brachiaria, Lolium, Bromus, Avena, Cyperus, Sorghum, Agropyron, Cynodon, Monochoria, Fimbristylis, Sagittaria, Eleocharis, Scirpus, Paspalum, Ischaemum, Spenoclea, Dactyloctenium, Agrostis, Alopecurus and Apera.

Monocotyledon cultures of the genera: Oryza, Zea, Triticum, Hordeum, Avena, Secale, Sorghum, Panicum, Saccharum, Ananas, Asparagus and Allium.

However, the use of the active compounds according to the invention is in no way restricted to these genera, but also extends in the same manner to other plants.

The compounds are suitable, depending on the concentration, for the total combating of weeds, for example on industrial terrain and rail tracks, and on paths and squares with or without tree plantings. Equally, the compounds can be employed for combating weeds in perennial cultures, for example afforestations, decorative tree plantings, orchards, vineyards, citrus groves, nut orchards, banana plantations, coffee plantations, tea plantations, rubber plantations, oil palm plantations, cocoa plantations, soft fruit plantings and hopfields, in lawns, turf and pasture-land, and for the selective combating of weeds in annual cultures.

The compounds of the formula (I) according to the invention are preferably suitable for combating monocotyledon and dicotyledon broad-leaved weeds, both pre-emergence and post-emergence. They exhibit strong herbicidal activity and a broad spectrum action when used on the soil and on above-ground parts of the plants.

The active compounds can be converted into the customary formulations, such as solutions, emulsions, wettable powders, suspensions, powders, dusting agents, pastes, soluble powders, granules, suspension-emulsion concentrates, natural and synthetic materials impregnated with active compound, and very fine capsules in polymeric substances.

These formulations are produced in a known manner, for example by mixing the active compounds with extenders, that is liquid solvents and/or solid carriers, optionally with the use of surface-active agents, that is emulsifying agents and/or dispersing agents and/or foam-forming agents.

In the case of the use of water as an extender, organic solvents can, for example, also be used as auxiliary solvents. As liquid solvents, there are suitable in the main: aromatics, such as xylene, toluene or alkyl-naphthalenes, chlorinated aromatics and chlorinated aliphatic hydrocarbons, such as chlorobenzenes, chloroethylenes or methylene chloride, aliphatic hydrocarbons, such as cyclohexane or paraffins, for example petroleum fractions, mineral and vegetable oils, alcohols, such as butanol or glycol as well as their ethers and esters, ketones, such as acetone, methyl ethyl ketone, methyl isobutyl ketone or cyclohexanone, strongly polar solvents, such as dimethylformamide and dimethyl sulphoxide, as well as water.

As solid carriers there are suitable:

for example ammonium salts and ground natural minerals, such as kaolins, clays, talc, chalk, quartz, attapulgite, montmorillonite or diatomaceous earth, and ground synthetic minerals, such as highly disperse silica, alumina and silicates, as solid carriers for granules there are suitable: for example crushed and fractionated natural rocks such as calcite, marble, pumice, sepiolite and dolomite, as well as synthetic granules of inorganic and organic meals, and granules of organic material such as sawdust, coconut shells, maize cobs and tobacco stalks; as emulsifying and/or foam forming agents there are suitable: for example non-ionic and anionic emulsifiers, such as polyoxyethylene fatty acid esters, polyoxyethylene fatty alcohol ethers, for example alkylaryl polyglycol ethers, alkylsulphonates, alkyl sulphates, arylsulphonates as well as albumen hydrolysis products; as dispersing agents there are suitable: for example lignin-sulphite waste liquors and methylcellulose.

Adhesives such as carboxymethylcellulose and natural and synthetic polymers in the form of powders, granules or latexes, such as gum arabic, polyvinyl alcohol and polyvinyl acetate, as well as natural phospholipids, such as cephalins and lecithins, and synthetic phospholipids, can be used in the formulations. Further additives can be mineral and vegetable oils.

It is possible to use colorants such as inorganic pigments, for example iron oxide, titanium oxide and Prussian Blue,

and organic dyes, such as alizarin dyes, azo dyes and metal phthalocyanine dyes, and trace nutrients such as salts of iron, manganese, boron, copper, cobalt, molybdenum and zinc.

The formulations in general contain between 0.1 and 95 per cent by weight of active compound, preferably between 0.5 and 90%.

For controlling weeds, the active compounds according to the invention, as such or in the form of their formulations, can also be used as mixtures with known herbicides, finished formulations or tank mixes being possible.

Possible components for the mixtures are known herbicides, for example anilides, for example diflufenican and propanil; arylcarboxylic acids, for example dichloropicolinic acid, dicamba and picloram; aryloxyalkanoic acids, for example 2,4-D, 2,4-DB, 2,4-DP, fluroxypyr, MCPA, MCPP and triclopyr; aryloxy-phenoxy-alkanoic esters, for example diclofop-methyl, fenoxaprop-ethyl, fluazifop-butyl, haloxyfop-methyl and quizalofop-ethyl; azinones, for example chloridazon and norflurazon; carbamates, for example chlorpropham, desmedipham, phenmedipham and propham; chloroacetanilides, for example alachlor, acetochlor, butachlor, metazachlor, metolachlor, pretilachlor and propachlor; dinitroanilines, for example oryzalin, pendimethalin and trifluralin; diphenyl ethers, for example acifluorfen, bifenox, fluoroglycofen, fomesafen, halosafen, lactofen and oxyfluorfen; ureas, for example chlorotoluron, diuron, fluometuron, isoproturon, linuron and methabenzthiazuron; hydroxylamines, for example alloxydim, clethodim, cycloxydim, sethoxydim and tralkoxydim; imidazolinones, for example imazethapyr, imazamethabenz, imazapyr and imazaquin; nitriles, for example bromoxynil, dichlobenil and ioxynil; oxyacetamides, for example mefenacet; sulphonyl-ureas, for example amidosulfuron, bensulfuron-methyl, chlorimuron-ethyl, chlorsulfuron, cinosulfuron, metsulfuron-methyl, nicosulfuron, primisulfuron, pyrazosulfuron-ethyl, thifensulfuron-methyl, trisulfuron and tribenuron-methyl; thiocarbamates, for example butylate, cycloate, diallate, EPTC, esprocarb, molinate, prosulfocarb, thiobencarb and triallate; triazines, for example atrazine, cyanazine, simazine, simetryne, terbutryne and terbutylazine; triazinones, for example hexazinone, metamitron and metribuzin; others, for example aminotriazole, benfuresate, bentazone, cinnemethylin, clomazone, clopyralid, difenzoquat, dithiopyr, ethofumesate, fluorchloridone, glufosinate, glyphosate, isoxaben, pyridate, quinchlorac, quinmerac, sulphosate and tridiphan.

Mixtures with other known active compounds, such as fungicides, insecticides, acaricides, nematocides, bird repellants, plant nutrients and agents which improve soil structure, are also possible.

The active compounds can be used as such, in the form of their formulations or in the use forms prepared therefrom by further dilution, such as ready-to-use solutions, suspensions, emulsions, powders, pastes and granules. They are used in the customary manner, for example by watering, spraying, atomizing or scattering.

The active compounds according to the invention can be applied either before or after emergence of the plants. They can also be incorporated into the soil before sowing.

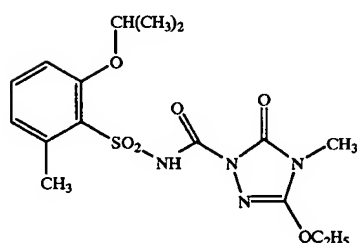
The amount of active compound used can vary within a substantial range. It depends essentially on the nature of the desired effect. In general, the amounts used are between 1 and 10 kg of active compound per hectare of soil surface, preferably between 5 g and 5 kg per ha.

The preparation and use of the active compounds according to the invention can be seen from the following examples.

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PREPARATION EXAMPLES

Example 1

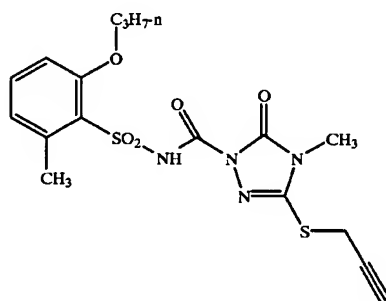


(Process (a))

A mixture of 2.5 g (10 mmol) of 5-ethoxy-4-methyl-2-phenoxy-carbonyl-2,4-dihydro-3H-1,2,4-triazol-3-one, 2.3 g (10 mmol) of 2-isopropoxy-6-methyl-benzenesulphonamide, 1.5 g (10 mmol) of diazabicyclo [5.4.0]undec-7-ene (DBU) and 50 ml of acetonitrile is stirred at 20° C. for 5 hours. It is then concentrated under a water pump vacuum and the residue is stirred with 50 ml of 1N hydrochloric acid, the mixture is filtered with suction, the filter product is stirred with diethyl ether and the mixture is again filtered with suction.

2.4 g (60% of theory) of 5-ethoxy-4-methyl-2-(2-isopropoxy-6-methyl-phenylsulphonyl-aminocarbonyl)-2,4-dihydro-3H-1,2,4-triazol-3-one are obtained of melting point 155° C.

Example 2



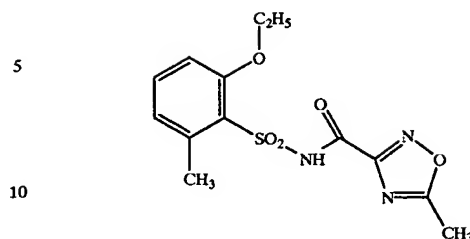
(Process (c))

A mixture of 1.7 g (10 mmol) of 4-methyl-5-propargylthio-2,4-dihydro-3H-1,2,4-triazol-3-one, 1.3 g (20 mmol) of sodium cyanate, 2.5 g (10 mmol) of 2-methyl-6-n-propoxy-benzenesulphochloride and 50 ml of acetonitrile is heated under reflux for 3 hours. It is then concentrated under a water pump vacuum, the residue is stirred with 1N hydrochloric acid and the mixture is subjected three times to extraction with 50 ml of methylene chloride each time. The combined organic extraction solutions are concentrated, the residue is digested with isopropanol and the crystalline product is isolated by filtration with suction.

2.2 g (52% of theory) of 4-methyl-5-propargylthio-2-(2-methyl-6-n-propoxy-phenylsulphonyl-aminocarbonyl)-2,4-dihydro-3H-1,2,4-triazol-3-one are obtained of melting point 151° C.

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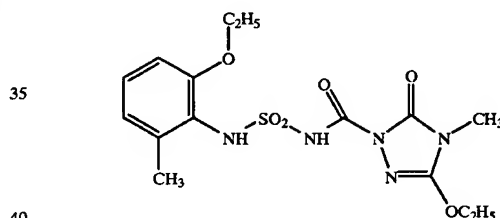
Example 3



(Process (d))

A mixture of 3.2 g (25 mmol) of 5-methyl-1,2,4-oxadiazole-3-carboxamide, 4.2 g (75 mmol) of potassium hydroxide (powder) and 200 ml of dioxane is stirred at 60° C. for 30 minutes. It is then concentrated to about half its volume under a water pump vacuum, and a solution of 7 g (30 mmol) of 2-ethoxy-6-methyl-benzenesulphochloride in 10 ml of dioxane is added dropwise at about 20° C. The reaction mixture is then stirred at 20° C. for about 15 hours more. It is then concentrated under a water pump vacuum, the residue is stirred with 50 ml of 1N hydrochloric acid and the crystalline product is isolated by filtration with suction. 4.0 g (49% of theory) of N-(2-ethoxy-6-methyl-phenylsulphonyl)-5-methyl-1,2,4-oxadiazole-3-carboxamide are obtained of melting point 168° C.

Example 4



(Process (f))

1.7 g (12 mmol) of chlorosulphonyl isocyanate are added to a solution, cooled to 5° C., of 1.4 g (10 mmol) of 5-ethoxy-4-methyl-2,4-dihydro-3H-1,2,4-triazol-3-one in 50 ml of methylene chloride, and then a solution of 1.5 g (10 mmol) of 2-ethoxy-6-methyl-aniline and 1.0 g (10 mmol) of triethylamine in 10 ml of methylene chloride is added dropwise, likewise at 5° C. The reaction mixture is then stirred at about 20° C. For 15 hours. Subsequently, 100 ml of 1N hydrochloric acid are added. After a thorough stirring, the organic phase is separated off, dried over sodium sulphate and filtered. The filtrate is concentrated under water pump vacuum, the residue is digested with isopropanol and the crystalline product is isolated by filtration with suction.

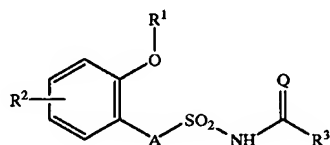
1.8 g (45% of theory) of 5-ethoxy-4-methyl-2-(2-ethoxy-6-methyl-phenyl-aminosulphonyl-aminocarbonyl)-2,4-dihydro-3H-1,2,4-triazol-3-one are obtained of melting point 147° C.

In analogy to Example 1 to 4 and in accordance with the general description of the preparation processes according to the invention, it is also possible, for example, to prepare the compounds of the formula (I) listed in Table 1 below (showing Examples Nos. 5-965), as well as the compounds of the formula (IA) listed in Table IA thereafter (showing Examples Nos. 966-1637):

TABLE 1

Examples of the compounds of the formula (I)

(I)

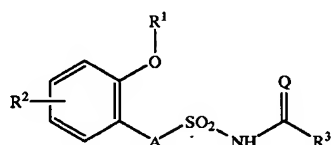


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
5	—	O	n-C ₃ H ₇	(6-)CH ₃		117
6	—	O	C ₂ H ₅	(6-)Cl		156
7	—	O	C ₂ H ₅	(6-)CH ₃		110
8	—	O	C ₂ H ₅	(6-)CH ₃		141
9	—	O	C ₂ H ₅	(6-)CH ₃		162
10	—	O	n-C ₃ H ₇	(6-)CH ₃		126

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

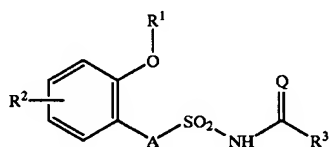


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
11	—	O	n-C ₃ H ₇	(6-)CH ₃		150
12	—	O	n-C ₃ H ₇	(6-)CH ₃		129
13	—	O	CH ₃	(6-)CH ₃		153
14	—	O	CH ₃	(6-)CH ₃		167
15	—	O	CH ₃	(6-)CH ₃		167
16	—	O	i-C ₃ H ₇	(6-)CH ₃		125

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

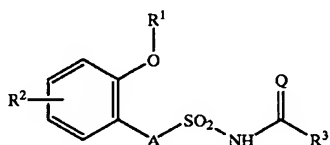


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
17	—	O	i-C ₃ H ₇	(6-)CH ₃		131
18	—	O	C ₂ H ₅	(5-)CH ₃		222
19	—	O	C ₂ H ₅	(5-)CH ₃		139
20	—	O	C ₂ H ₅	(4-)CH ₃		189
21	—	O	C ₂ H ₅	(5-)CH ₃		131
22	—	O	—C ₂ H ₄ OC ₂ H ₅	(6-)CH ₃		118

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

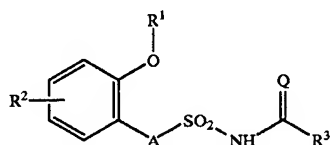


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
23	—	O	—CH ₂ CH ₂ Cl	(6-)CH ₃		137
24	—	O	—CH ₂ CH ₂ Cl	(6-)CH ₃		149
25	—	O	i-C ₃ H ₇	(5-)CH ₃		125
26	—	O	i-C ₃ H ₇	(5-)CH ₃		140
27	—	O	n-C ₃ H ₇	(5-)CH ₃		119
28	—	O	n-C ₃ H ₇	(5-)CH ₃		134

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

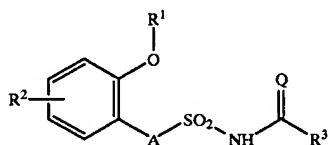


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
29	—	O	n-C ₃ H ₇	(5-)CH ₃		110
30	—	O	C ₂ H ₅	(6-)CH ₃		108
31	—	O	C ₂ H ₅	(6-)CH ₃		173
32	—	O	C ₂ H ₅	(6-)CH ₃		119
33	—	O	C ₂ H ₅	(6-)CH ₃		121
34	—	O	C ₂ H ₅	(6-)CH ₃		109

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

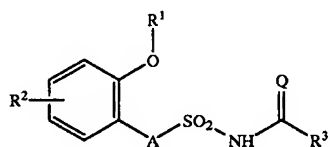


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
35	—	O	C ₂ H ₅	(6-)CH ₃		111
36	—	O	n-C ₃ H ₇	(6-)CH ₃		91
37	—	O	n-C ₃ H ₇	(6-)CH ₃		130
38	—	O	n-C ₃ H ₇	(6-)CH ₃		126
39	—	O	n-C ₃ H ₇	(6-)CH ₃		101
40	—	O	n-C ₃ H ₇	(6-)CH ₃		152

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)



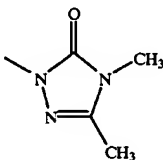
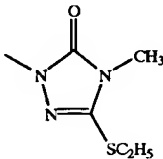
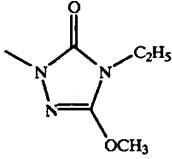
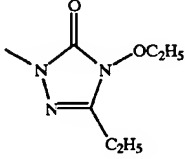
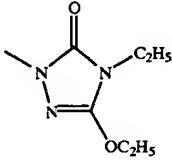
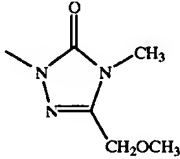
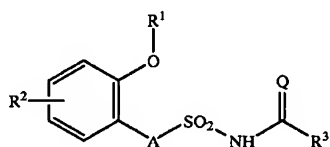
Ex. No.	A	Q	R¹	(position-) R²	R³	Melting point (° C.)
41	—	O	n-C₃H₇	(6-)CH₃		100
42	—	O	n-C₃H₇	(6-)CH₃		120
43	—	O	n-C₃H₇	(6-)CH₃		117
44	—	O	n-C₃H₇	(6-)CH₃		126
45	—	O	n-C₃H₇	(6-)CH₃		113
46	—	O	i-C₃H₇	(6-)CH₃		130

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

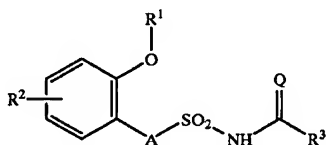


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
47	—	O	$i\text{-C}_3\text{H}_7$	(6-)CH ₃		139
48	—	O	$i\text{-C}_3\text{H}_7$	(6-)CH ₃		121
49	—	O	$i\text{-C}_3\text{H}_7$	(6-)CH ₃		119
50	—	O	$i\text{-C}_3\text{H}_7$	(6-)CH ₃		128
51	—	O	$i\text{-C}_3\text{H}_7$	(6-)CH ₃		134
52	—	O	$i\text{-C}_3\text{H}_7$	(6-)CH ₃		130

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

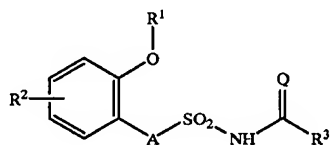


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
53	—	O	$i\text{-C}_3\text{H}_7$	(6-)CH ₃		117
54	—	O	$i\text{-C}_3\text{H}_7$	(6-)CH ₃		134
55	—	O	$i\text{-C}_3\text{H}_7$	(6-)CH ₃		141
56	—	O	$i\text{-C}_3\text{H}_7$	(6-)CH ₃		132
57	—	O	$i\text{-C}_3\text{H}_7$	(6-)CH ₃		166
58	—	O	$i\text{-C}_3\text{H}_7$	(6-)CH ₃		118

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

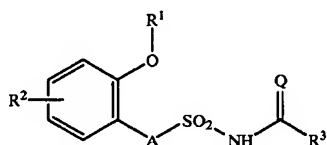


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
59	—	O	i-C ₃ H ₇	(6-)CH ₃		150
60	—	O	i-C ₃ H ₇	(6-)CH ₃		144
61	—	O	i-C ₃ H ₇	(6-)CH ₃		170
62	—	O	i-C ₃ H ₇	(6-)CH ₃		120
63	—	O	n-C ₃ H ₇	(6-)CH ₃		124
64	—	O	n-C ₃ H ₇	(6-)CH ₃		125

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

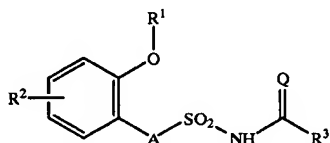


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
65	—	O	n-C ₃ H ₇	(6-)CH ₃		116
66	—	O	n-C ₃ H ₇	(6-)CH ₃		152
67	—	O	n-C ₃ H ₇	(6-)CH ₃		143
68	—	O	C ₂ H ₅	(6-)CH ₃		160
69	—	O	C ₂ H ₅	(6-)CH ₃		133
70	—	O	C ₂ H ₅	(6-)CH ₃		97

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

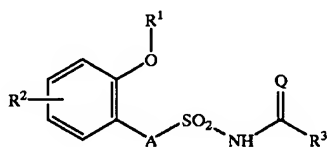


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point ($^{\circ}$ C.)
71	—	O	C_2H_5	(6-) CH_3		96
72	—	O	C_2H_5	(6-) CH_3		156
73	—	O	C_2H_5	(6-) CH_3		145
74	—	O	C_2H_5	(6-) CH_3		120
75	—	O	C_2H_5	(6-) CH_3		125
76	—	O	C_2H_5	(6-) CH_3		140

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

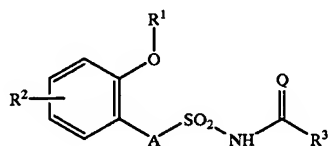


Ex. No.	A	Q	R¹	(position-) R²	R³	Melting point (° C.)
77	—	O	H	(6-)CH₃		88
78	—	O	C₂H₅	(5-)CH₃		130
79	—	O	n-C₃H₇	(6-)CH₃		141
80	—	O	n-C₃H₇	(6-)CH₃		98
81	—	O	n-C₃H₇	(6-)CH₃		141
82	—	O	n-C₃H₇	(6-)CH₃		101

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

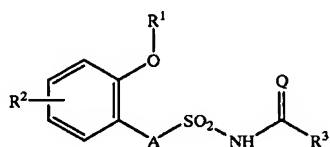


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
83	—	O	n-C ₃ H ₇	(6-)CH ₃	<p>Chemical structure of R³ for example 83: A 1,3,4-oxadiazole ring with a carbonyl group (=O) at position 2, a methyl group (-CH₃) at position 4, and a 2,2,2-trifluoroethoxy group (-OCH₂CF₂CHF₂) at position 5.</p>	136
84	—	O	n-C ₃ H ₇	(6-)CH ₃	<p>Chemical structure of R³ for example 84: A 1,3,4-oxadiazole ring with a carbonyl group (=O) at position 2, an ethyl group (-C₂H₅) at position 4, and an isopropoxy group (-OCH(CH₃)₂) at position 5.</p>	96
85	—	O	n-C ₃ H ₇	(6-)CH ₃	<p>Chemical structure of R³ for example 85: A 1,3,4-oxadiazole ring with a carbonyl group (=O) at position 2, a methyl group (-CH₃) at position 4, and a 1-ethoxypropyl group (-OCH₂CH(CH₃)C₂H₅) at position 5.</p>	90
86	—	O	n-C ₃ H ₇	(6-)CH ₃	<p>Chemical structure of R³ for example 86: A 1,3,4-oxadiazole ring with a carbonyl group (=O) at position 2, a methyl group (-CH₃) at position 4, and a 2-phenoxyethyl group (-OCH₂CH₂C₆H₅) at position 5.</p>	136
87	—	O	C ₂ H ₅	(6-)CH ₃	<p>Chemical structure of R³ for example 87: A 1,3,4-oxadiazole ring with a carbonyl group (=O) at position 2, a methyl group (-CH₃) at position 4, and a prop-1-ynyl group (-SCH₂CH₂C≡CH) at position 5.</p>	122

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

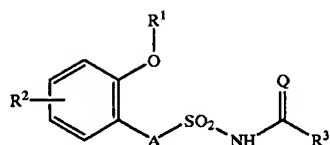


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
88	—	O	C_2H_5	(6-)CH ₃		154
89	—	O	<i>i</i> -C ₃ H ₇	(6-)CH ₃		139
90	—	O	C_2H_5	(6-)CH ₃		142
91	—	O	C_2H_5	(6-)CH ₃		153
92	—	O	C_2H_5	(6-)CH ₃		145
93	—	O	C_2H_5	(6-)CH ₃		132

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

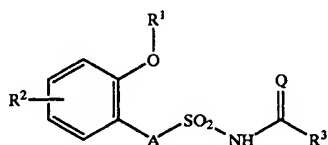


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
94	—	O	C ₂ H ₅	(6-)CH ₃		141
95	—	O	C ₂ H ₅	(6-)CH ₃		130
96	—	O	CH ₃	(5-)CH ₃		156
97	—	O	CH ₃	(5-)CH ₃		177
98	—	O	CH ₃	(4-)CH ₃		115
99	—	O	CH ₃	(4-)CH ₃		166

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

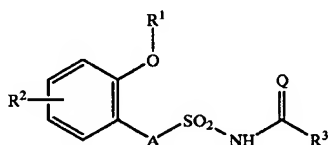


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
100	—	O	CH_3	(3-) CH_3		162
101	—	O	CH_3	(3-) CH_3		143
102	—	O	CH_3	(3-) CH_3		165
103	—	O	CHF_2	(5-) CH_3		176
104	—	O	CHF_2	(5-) CH_3		119
105	—	O	CHF_2	(5-) CH_3		126

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

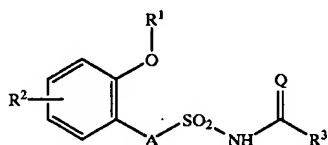


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
106	—	O	CHF_2	(5-)CH ₃		151
107	—	O	CHF_2	(5-)CH ₃		188
108	—	O	CHF_2	(5-)CH ₃		137
109	—	O	CHF_2	(5-)CH ₃		117
110	—	O	CHF_2	(5-)CH ₃		155
111	—	O	CHF_2	(4-)CH ₃		152

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

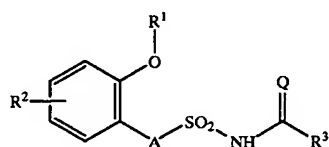


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
112	—	O	CHF ₂	(4-)CH ₃		176
113	—	O	CHF ₂	(4-)CH ₃		108
114	—	O	CHF ₂	(6-)CH ₃		163
115	—	O	CHF ₂	(6-)CH ₃		136
116	—	O	CHF ₂	(6-)CH ₃		118
117	—	O	CHF ₂	(6-)CH ₃		104

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

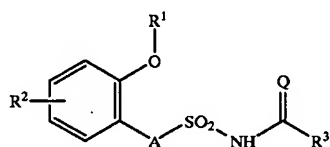


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
118	—	O	CHF ₂	(5-)CH ₃		98
119	—	O	CHF ₂	(6-)CH ₃		128
120	—	O	CHF ₂	(6-)CH ₃		165
121	—	O	CHF ₂	(6-)CH ₃		155
122	—	O	CHF ₂	(6-)CH ₃		105
123	—	O	CHF ₂	(6-)CH ₃		81

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

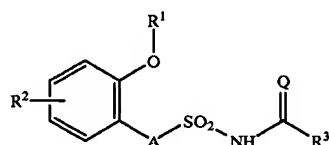


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
124	—	O	CHF ₂	(6-)CH ₃		174
125	—	O	CHF ₂	(5-)CH ₃		150
126	—	O	CHF ₂	(6-)CH ₃		124
127	—	O	CHF ₂	(6-)CH ₃		200
128	—	S	n-C ₃ H ₇	(6-)CH ₃		160
129	—	S	n-C ₃ H ₇	(6-)CH ₃		148

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

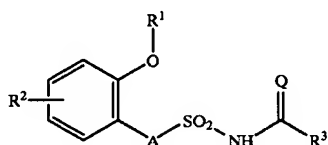


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
130	—	S	C ₂ H ₅	(6-)CH ₃		141
131	—	S	C ₂ H ₅	(6-)CH ₃		125
132	—	S	C ₂ H ₅	(6-)CH ₃		158
133	—	S	i-C ₃ H ₇	(6-)CH ₃		155
134	—	S	n-C ₃ H ₇	(6-)CH ₃		153
135	—	S	n-C ₃ H ₇	(6-)CH ₃		131

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

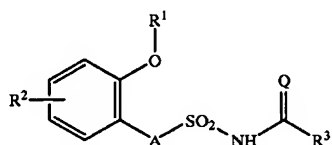


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
136	—	S	n-C ₃ H ₇	(6-)CH ₃		120
137	—	O	C ₂ H ₅	(6-)Cl		149
138	—	O	C ₂ H ₅	(6-)Cl		99
139	—	O	CH ₃	(6-)Cl		176
140	—	O	CH ₃	(6-)Cl		192
141	—	O	C ₂ H ₅	(6-)Cl		144

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

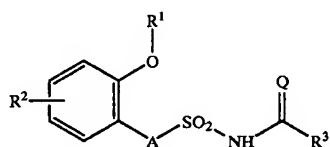


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
142	—	O	CH ₃	(6-)Cl		114
143	—	O	C ₂ H ₅	(6-)Cl		144
144	—	O	CH ₃	(6-)Cl		157
145	—	O	C ₂ H ₅	(6-)Cl		142
146	—	O	CH ₃	(6-)Cl		191
147	—	O	C ₂ H ₅	(6-)Cl		116

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

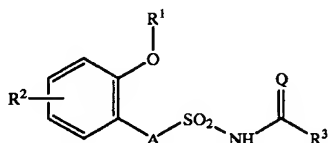


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
148	—	O	CH_3	(6-)Cl		205
149	—	O	CH_3	(6-)Cl		147
150	—	O	C_2H_5	(6-)Cl		117
151	—	O	CH_3	(6-)Cl		149
152	—	O	CH_3	(6-)Cl		176
153	—	O	CH_3	(6-)Cl		150

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

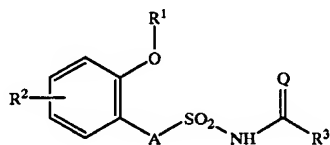


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
154	—	O	CH_3	(6-)Cl		146
155	—	O	CH_3	(6-)Cl		191
156	—	O	CH_3	(6-)Cl		127
157	—	O	CH_3	(6-)Cl		174
158	—	O	$n-C_3H_7$	(6-)Cl		117
159	—	O	$n-C_3H_7$	(6-)Cl		134

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

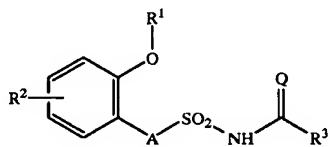


Ex. No.	A	Q	R¹	(position-) R²	R³	Melting point (° C.)
160	—	O	n-C₃H₇	(6-)Cl	 <chem>CN1C=NC(=O)N1SC</chem>	115
161	—	O	n-C₃H₇	(6-)Cl	 <chem>CCOC1=NC(=O)N1C2CC2</chem>	137
162	—	O	n-C₃H₇	(6-)Cl	 <chem>CN1C=NC(=O)N1COCC(F)(F)F</chem>	125
163	—	O	n-C₃H₇	(6-)Cl	 <chem>CN1C=NC(=O)N1SCC#CC</chem>	119
164	—	O	H	(6-)Cl	 <chem>CCOC1=NC(=O)N1C</chem>	147
165	—	O	n-C₃H₇	(6-)Cl	 <chem>CN1C=NC(=O)N1C2C(=O)OC2c3ccccc3</chem>	148

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

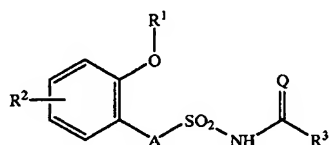


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
166	—	O	i-C ₃ H ₇	(6-)Cl		143
167	—	O	i-C ₃ H ₇	(6-)Cl		122
168	—	O	CH ₃	(6-)Cl		165
169	—	O	n-C ₃ H ₇	(5-)Cl		154
170	—	O	n-C ₃ H ₇	(5-)Cl		136

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

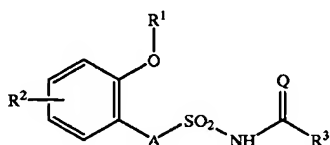


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
171	—	O	n-C ₃ H ₇	(5-)Cl		128
172	—	O	CH ₃	(6-)Cl		143
173	—	O	i-C ₃ H ₇	(6-)Cl		136
174	—	O	i-C ₃ H ₇	(6-)Cl		121
175	—	O	i-C ₃ H ₇	(6-)Cl		158
176	—	O	i-C ₃ H ₇	(6-)Cl		141

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

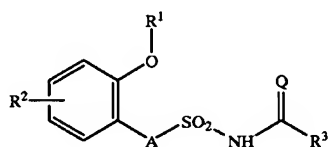


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
177	—	O	i-C ₃ H ₇	(6-)Cl		127
178	—	O	i-C ₃ H ₇	(6-)Cl		143
179	—	O	i-C ₃ H ₇	(6-)Cl		129
180	—	O	i-C ₃ H ₇	(6-)Cl		95
181	—	O	n-C ₃ H ₇	(6-)CH ₃		74
182	—	O	n-C ₃ H ₇	(6-)CH ₃		114

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

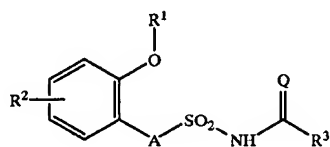


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
183	—	O	n-C ₃ H ₇	(6-)CH ₃		140
184	—	O	n-C ₃ H ₇	(6-)CH ₃		159
185	—	O	n-C ₃ H ₇	(6-)CH ₃		107
186	—	O	n-C ₃ H ₇	(6-)CH ₃		132
187	—	O	n-C ₃ H ₇	(6-)CH ₃		132
188	—	O	n-C ₃ H ₇	(6-)CH ₃		110

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

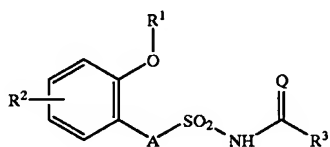


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
189	—	O	CH ₃	(6-)CH ₃		159
190	—	O	n-C ₃ H ₇	(6-)CH ₃		138
191	—	O	n-C ₃ H ₇	(6-)CH ₃		147
192	—	O	n-C ₃ H ₇	(6-)CH ₃		114
193	—	O	n-C ₃ H ₇	(6-)CH ₃		125
194	—	O	n-C ₃ H ₇	(6-)CH ₃		126

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

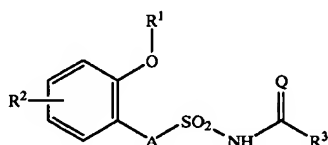


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
195	—	O	n-C ₃ H ₇	(6-)CH ₃		151
196	—	O	n-C ₃ H ₇	(6-)CH ₃		121
197	—	O	n-C ₃ H ₇	(6-)CH ₃		147
198	NH	O	C ₂ H ₅	(6-)CH ₃		135
199	—	O	i-C ₃ H ₇	(6-)CH ₃		263 (Sodium salt)
200	—	O	C ₂ H ₅	(6-)CH ₃		119

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

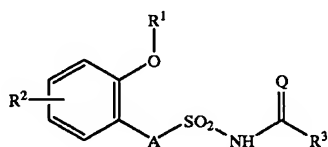


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
201	—	O	C ₂ H ₅	(6-)CH ₃		146
202	—	O	C ₂ H ₅	(6-)CH ₃		128
203	—	O	C ₂ H ₅	(6-)CH ₃		186
204	—	O	C ₂ H ₅	(6-)CH ₃		239
205	—	O	C ₂ H ₅	(6-)CH ₃		152
206	—	O	C ₂ H ₅	(6-)CH ₃		155

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

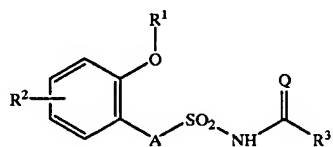


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
207	—	O	C ₂ H ₅	(6-)CH ₃		145 (Na salt)
208	—	O	i-C ₃ H ₇	(6-)CH ₃		209 (Na salt)
209	—	O	C ₂ H ₅	(6-)CH ₃		147 (Na salt)
210	—	O	C ₂ H ₅	(6-)CH ₃		140
211	—	O	C ₂ H ₅	(6-)CH ₃		118
212	—	O	C ₂ H ₅	(6-)CH ₃		156

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

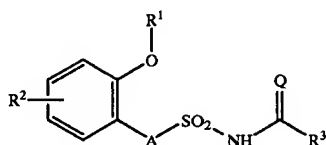


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
213	—	O	C ₂ H ₅	(6-)CH ₃		110
214	—	O	C ₂ H ₅	(6-)CH ₃		133
215	—	O	i-C ₃ H ₇	(6-)CH ₃		138
216	—	O	i-C ₃ H ₇	(6-)CH ₃		154
217	—	O	i-C ₃ H ₇	(6-)CH ₃		149
218	—	O	i-C ₃ H ₇	(6-)CH ₃		112

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

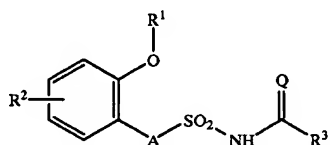


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
219	—	O	i-C ₃ H ₇	(6-)CH ₃		162
220	—	O	i-C ₃ H ₇	(6-)CH ₃		99
221	—	O	i-C ₃ H ₇	(6-)CH ₃		146
222	—	O	C ₂ H ₅	(6-)CH ₃		134
223	—	O	CH ₃	(6-)CH ₃		199
224	—	O	CH ₃	(6-)CH ₃		176

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

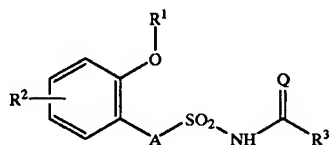


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
225	—	O	CH_3	(6-) CH_3		145
226	—	O	$i-C_3H_7$	(6-) CH_3		133
227	NH	O	$n-C_3H_7$	(5-) CH_3		127
228	—	O	$i-C_3H_7$	(6-) CH_3		144
229	—	O	$i-C_3H_7$	(6-) CH_3		141
230	—	O	$i-C_3H_7$	(6-) CH_3		152

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

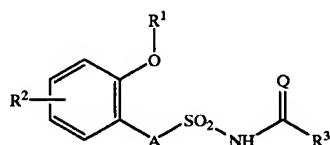


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
231	—	O	i-C ₃ H ₇	(6-)CH ₃		132
232	—	O	i-C ₃ H ₇	(6-)CH ₃		147
233	—	O	i-C ₃ H ₇	(6-)CH ₃		163
234	—	O	i-C ₃ H ₇	(6-)CH ₃		102
235	—	O	C ₂ H ₅	(6-)CH ₃		121
236	—	O	C ₂ H ₅	(6-)CH ₃		113

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

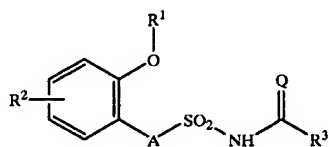


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
237	—	O	C_2H_5	(6-)CH ₃		145
238	—	O	C_2H_5	(6-)CH ₃		137
239	—	O	C_2H_5	(6-)CH ₃		172
240	—	O	C_2H_5	(6-)CH ₃		148
241	—	O	C_2H_5	(6-)CH ₃		157

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

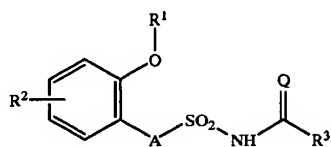


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
242	—	O	C ₂ H ₅	(6-)CH ₃		186
243	NH	O	CH ₃	(6-)OCH ₃		170
244	—	O	CHF ₂	(5-)CH ₃		
245	—	S	i-C ₃ H ₇	(6-)CH ₃		160
246	—	O	CH ₃	(6-)CF ₃		205
247	—	S	CH ₃	(6-)CF ₃		92

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

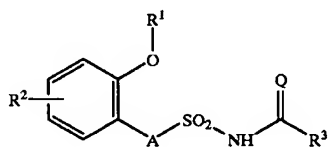


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
248	—	S	CH ₃	(6-)CF ₃		154
249	—	S	CH ₃	(6-)CF ₃		157
250	—	O	C ₂ H ₅	(6-)CF ₃		176
251	—	O	n-C ₃ H ₇ n	(6-)CF ₃		166
252	—	O	i-C ₃ H ₇	(6-)CF ₃		190
253	—	O	CH ₃	(6-)CF ₃		203
254	—	O	CH ₃	(6-)CF ₃		156

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

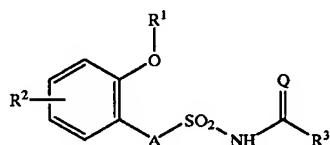


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
255	—	O	CH_3	(6-) CF_3		170
256	—	O	CH_3	(6-) CF_3		198
257	—	O	CH_3	(6-) CF_3		213
258	—	O	CH_3	(6-) CF_3		152
259	—	O	CH_3	(6-) CF_3		187
260	—	O	CH_3	(6-) CF_3		210

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

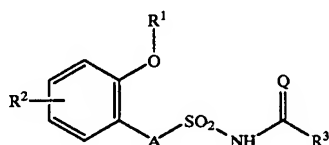


Ex. No.	A	Q	R¹	(position-) R²	R³	Melting point (° C.)
261	—	O	CH₃	(6-)CF₃		172
262	—	O	CH₃	(6-)CF₃		145
263	—	O	CH₃	(6-)CF₃		136
264	—	O	CH₃	(6-)CF₃		153
265	—	O	CH₃	(6-)CF₃		136
266	—	O	CH₃	(6-)CF₃		210

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

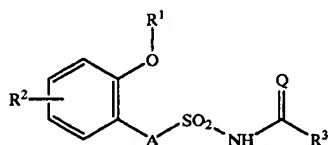


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
267	—	O	CH ₃	(6-)CF ₃		147
268	—	O	CH ₃	(6-)CF ₃		169
269	—	O	CH ₃	(6-)CF ₃		215
270	—	O	CH ₃	(6-)CF ₃		138
271	—	O	CH ₃	(6-)CF ₃		182
272	—	S	C ₂ H ₅	(6-)CF ₃		112

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

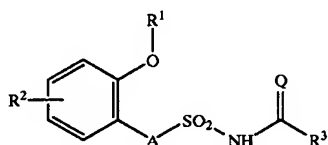


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
273	—	S	C ₂ H ₅	(6-)CF ₃		167
274	—	S	C ₂ H ₅	(6-)CF ₃		152
275	—	S	n-C ₃ H ₇	(6-)CF ₃		119
276	—	S	n-C ₃ H ₇	(6-)CF ₃		157
277	—	S	n-C ₃ H ₇	(6-)CF ₃		154
278	—	S	i-C ₃ H ₇	(6-)CF ₃		137

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

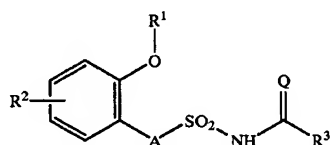


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
279	—	S	$i\text{-C}_3\text{H}_7$	(6-) CF_3		167
280	—	S	$i\text{-C}_3\text{H}_7$	(6-) CF_3		137
281	—	O	C_2H_5	(6-) CF_3		154
282	—	O	C_2H_5	(6-) CF_3		160
283	—	O	C_2H_5	(6-) CF_3		139
284	—	O	C_2H_5	(6-) CF_3		134

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

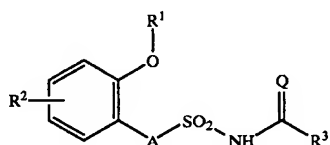


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
285	—	O	C ₂ H ₅	(6-)CF ₃		142
286	—	O	C ₂ H ₅	(6-)CF ₃		120
287	—	O	n-C ₃ H ₇	(6-)CF ₃		130
288	—	O	n-C ₃ H ₇	(6-)CF ₃		127
289	—	O	n-C ₃ H ₇	(6-)CF ₃		116
290	—	O	n-C ₃ H ₇	(6-)CF ₃		126

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

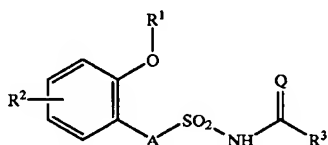


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
291	—	O	n-C ₃ H ₇	(6-)CF ₃		113
292	—	O	i-C ₃ H ₇	(6-)CF ₃		151
293	—	O	i-C ₃ H ₇	(6-)CF ₃		157
294	—	O	i-C ₃ H ₇	(6-)CF ₃		171
295	—	O	i-C ₃ H ₇	(6-)CF ₃		137
296	—	O	i-C ₃ H ₇	(6-)CF ₃		125

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

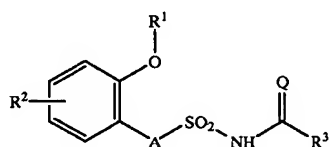


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
297	—	O	n-C ₃ H ₇	(6-)CF ₃		109
298	—	O	n-C ₃ H ₇	(6-)CF ₃		138
299	—	O	i-C ₃ H ₇	(6-)CF ₃		130
300	—	O	C ₂ H ₅	(6-)CF ₃		165
301	—	O	i-C ₃ H ₇	(6-)CF ₃		148
302	—	O	i-C ₃ H ₇	(6-)CF ₃		147

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

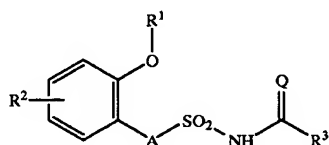


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
303	—	O	i-C ₃ H ₇	(6-)CF ₃		172
304	—	O	i-C ₃ H ₇	(6-)CF ₃		147
305	—	O	i-C ₃ H ₇	(6-)CF ₃		136
306	—	O	C ₂ H ₅	(6-)CF ₃		124
307	—	O	C ₂ H ₅	(6-)CF ₃		98
308	—	O	C ₂ H ₅	(6-)CF ₃		125

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

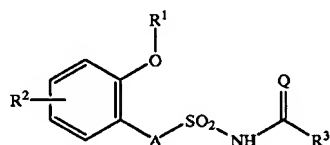


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
309	—	O	C ₂ H ₅	(6-)CF ₃		179
310	—	O	C ₂ H ₅	(6-)CF ₃		153
311	—	O	C ₂ H ₅	(6-)CF ₃		171
312	—	O	n-C ₃ H ₇	(6-)CF ₃		113
313	—	O	n-C ₃ H ₇	(6-)CF ₃		138
314	—	O	n-C ₃ H ₇	(6-)CF ₃		110

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

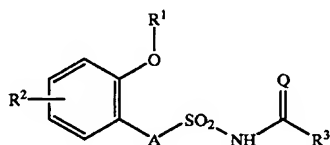


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
315	—	O	n-C ₃ H ₇	(6-)CF ₃		134
316	—	O	i-C ₃ H ₇	(6-)CF ₃		167
317	—	O	i-C ₃ H ₇	(6-)CF ₃		120
318	—	O	i-C ₃ H ₇	(6-)CF ₃		117
319	—	O	i-C ₃ H ₇	(6-)CF ₃		160
320	—	O	i-C ₃ H ₇	(6-)CF ₃		154

TABLE 1-continued

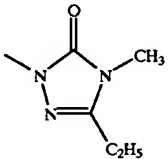
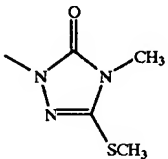
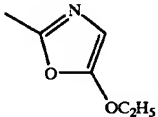
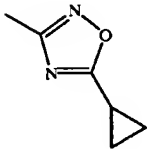
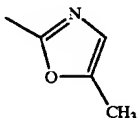
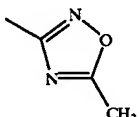
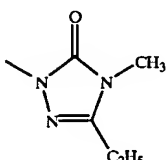
Examples of the compounds of the formula (I)

(I)



Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
321	—	O	i-C ₃ H ₇	(6-)CF ₃		159
322	—	O	n-C ₃ H ₇	(6-)CF ₃		141
323	—	O	i-C ₃ H ₇	(6-)CF ₃		146
324	—	O	i-C ₃ H ₇	(6-)CF ₃		134
325	—	O	i-C ₃ H ₇	(6-)CF ₃		168
326	—	O	CH ₃	(6-)C ₃ H _{7-n}		158

TABLE 1-continued

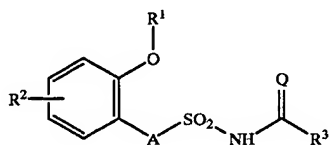
Examples of the compounds of the formula (I)						
Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
327	—	O	CH ₃	(6-)C ₃ H _{7-n}		172
328	—	O	CH ₃	(6-)C ₃ H _{7-n}		147
329	—	O	C ₂ H ₅	(6-)CF ₃		66
330	—	O	C ₂ H ₅	(6-)CH ₃		134
331	—	O	C ₂ H ₅	(6-)CH ₃		149
332	—	O	n-C ₃ H ₇	(6-)CH ₃		114
333	—	O	H	(6-)Cl		102

(I)

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

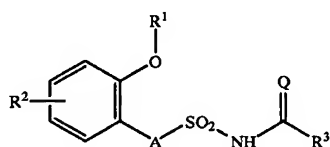


Ex. No.	A	Q	R¹	(position-) R²	R³	Melting point (° C.)
334	—	O	H	(6-)Cl		143
335	—	O		(6-)Cl		130
336	—	O	—CH₂C₆H₅	(6-)Cl		143
337	—	O	—CH₂C₆H₅	(6-)Cl		99
338	—	O	—CH₂C≡CH	(6-)Cl		161
339	—	O	n-C₃H₇	(6-)CH₃		133

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

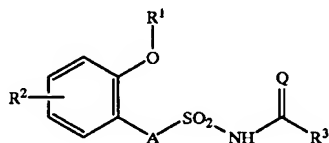


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
340	—	O	H	(6-)CH ₃		100
341	—	O	H	(6-)CH ₃		147
342	—	O	—CH ₂ C ₆ H ₅	(6-)CH ₃		157
343	—	O		(6-)CH ₃		150
344	—	O	—CH ₂ C≡CH	(6-)CH ₃		172
345	—	O	i-C ₃ H ₇	(6-)CH ₃		263 (Na salt)

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

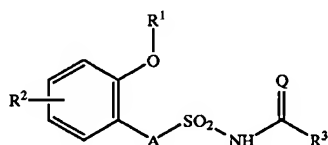


Ex. No.	A	Q	R¹	(position-) R²	R³	Melting point (° C.)
346	—	O	C₂H₅	(6-)CH₃		136
347	—	O	i-C₃H₇	(6-)CH₃		113
348	—	O	i-C₃H₇	(6-)CH₃		175
349	—	O	i-C₃H₇	(6-)CH₃		135
350	—	O	i-C₃H₇	(6-)CH₃		78
351	—	O	i-C₃H₇	(6-)CH₃		125

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

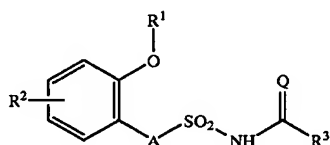


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
352	—	O	$i\text{-C}_3\text{H}_7$	(6-)CH ₃		140
353	—	O	CH ₃	(6-)CH ₃		161
354	—	O	CH ₃	(6-)CH ₃		142
355	—	O	CH ₃	(6-)CH ₃		124
356	—	O	CH ₃	(6-)CH ₃		153
357	—	O	CH ₃	(6-)CH ₃		170

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

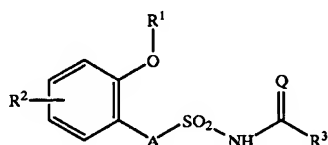


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
358	—	O	CH ₃	(6-)CH ₃		116
359	—	O	CH ₃	(6-)CH ₃		120
360	—	O	CH ₃	(5-)Cl		172
361	—	O	CH ₃	(5-)Cl		175
362	—	O	CH ₃	(5-)Cl		192
363	—	O	CH ₃	(5-)Cl		195

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

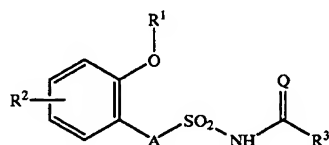


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
364	—	O	CH_3	(5-)Cl		174
365	—	O	CH_3	(5-)Cl		160
366	—	O	CH_3	(5-)Cl		214
367	—	O	CH_3	(5-)Cl		185
368	—	O	CH_3	(5-)Cl		191
369	NH	O	CH_3	(6-)CH ₃		161

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

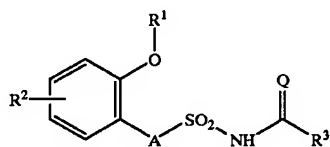


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
370	NH	O	i-C ₃ H ₇	(6-)CH ₃		132
371	NH	O	CH ₃	(6-)OCH ₃		151
372	NH	O	CH ₃	(6-)CH ₃		161
373	NH	O	i-C ₃ H ₇	(6-)CH ₃		128
374	NH	O	CH ₃	(6-)CH ₃		140
375	—	S	i-C ₃ H ₇	(6-)CH ₃		108

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

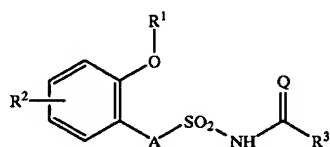


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
376	—	O	CHF ₂	(4-)CH ₃		131
377	—	O	CH ₃	(6-)CF ₃		187
378	—	O	CH ₃	(6-)CF ₃		154
379	—	O	CH ₃	(6-)C ₂ H ₅		179
380	—	O	CH ₃	(6-)C ₂ H ₅		178
381	—	O	CH ₃	(6-)C ₂ H ₅		167

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

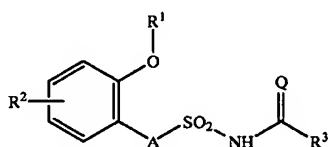


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
382	—	O	C ₂ H ₅	(6-)C ₂ H ₅		135
383	—	O	C ₂ H ₅	(6-)C ₂ H ₅		146
384	—	O	C ₂ H ₅	(6-)C ₂ H ₅		174
385	—	O	CH ₃	(6-)C ₂ H ₅		130
386	—	O	CH ₃	(6-)C ₂ H ₅		195
387	—	O	CH ₃	(6-)C ₂ H ₅		183

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

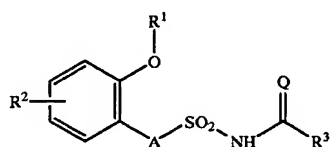


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
388	—	O	C ₂ H ₅	(6-)C ₃ H _{7-n}		135
389	—	O	C ₂ H ₅	(6-)C ₃ H _{7-n}		149
390	—	O	CH ₃	(6-)C ₃ H _{7-n}		193
391	—	O	CH ₃	(6-)C ₃ H _{7-n}		125
392	—	O	CH ₃	(6-)C ₃ H _{7-n}		182
393	—	O	C ₂ H ₅	(6-)C ₃ H _{7-n}		120

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

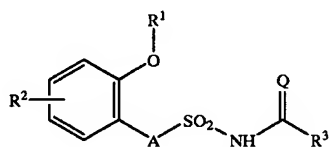


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
394	—	O	C ₂ H ₅	(6-)C ₃ H _{7-n}		158
395	—	O	i-C ₃ H ₇	(6-)CH ₃		180
396	—	O	n-C ₄ H ₉	(6-)CH ₃		132
397	—	O	n-C ₄ H ₉	(6-)CH ₃		143
398	—	O	n-C ₄ H ₉	(6-)CH ₃		106
399	—	O	n-C ₄ H ₉	(6-)CH ₃		98

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

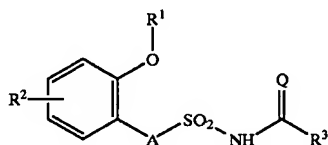


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
400	—	O	n-C ₄ H ₉	(6-)CH ₃		140
401	—	O	CH ₃	(6-)CH ₃		147
402	—	O	CH ₃	(6-)CH ₃		123
403	—	O	CH ₃	(6-)CH ₃		185
404	—	O	CH ₃	(6-)CH ₃		154
405	NH	O	i-C ₃ H ₇	(6-)CH ₃		150

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

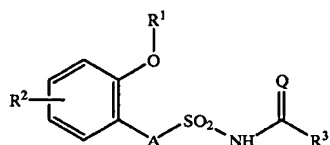


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
406	—	O	C ₂ H ₅	(6-)C ₂ H ₅		135
407	—	O	C ₂ H ₅	(6-)C ₂ H ₅		134
408	—	O	C ₂ H ₅	(6-)C ₂ H ₅		178
409	—	O	C ₃ H _{7-i}	(6-)C ₂ H ₅		109
410	—	O	C ₃ H _{7-i}	(6-)C ₂ H ₅		125
411	—	O	C ₃ H _{7-i}	(6-)C ₂ H ₅		161

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

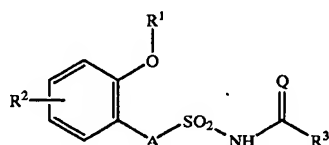


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
412	—	O	C_3H_7-i	(6-) C_2H_5		114
413	—	O	C_3H_7-i	(6-) C_2H_5		142
414	—	O	C_3H_7-i	(6-) C_2H_5		124
415	—	O	CH_3	(6-) C_2H_5		175 (Na salt)
416	—	O	C_2H_5	(6-) C_3H_7-n		126
417	—	O	C_2H_5	(6-) C_3H_7-n		121

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

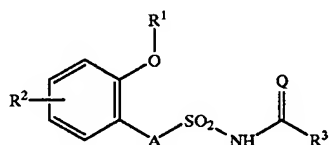


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
418	—	O	CH_3	(6-) C_3H_7-n		109
419	—	O	CH_3	(6-) C_3H_7-n		145
420	—	O	CH_3	(6-) C_3H_7-n		126
421	—	O	CH_3	(6-) C_3H_7-n		130
422	—	O	CH_3	(6-) C_3H_7-n		155
423	—	O	CH_3	(6-) C_3H_7-n		133

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

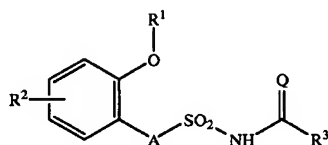


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
424	—	O	CH ₃	(6-)C ₃ H _{7-n}		145
425	—	O	—SO ₂ CH ₃	(6-)CH ₃		95
426	—	O	—SO ₂ CH ₃	(6-)CH ₃		153
427	—	O	C ₄ H _{9-n}	(6-)CH ₃		154
428	—	O	—CH ₂ C≡CH	(6-)CH ₃		167
429	—	O	—CH ₂ C≡CH	(6-)CH ₃		170

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

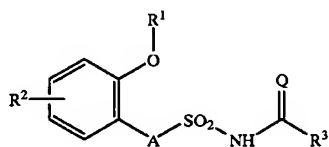


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
430	—	O	—CH ₂ C≡CH	(6-)CH ₃		153
431	—	O	C ₄ H ₉ -i	(6-)CH ₃		123
432	—	O	C ₄ H ₉ -i	(6-)CH ₃		145
433	—	O	C ₄ H ₉ -i	(6-)CH ₃		143
434	—	O	C ₃ H ₇ -i	(6-)CH ₃		138
435	—	O	C ₃ H ₇ -i	(6-)CH ₃		161

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)



Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
436	—	O	C ₃ H ₇ -i	(6-)CH ₃		128
437	—	O	C ₃ H ₇ -i	(6-)CH ₃		177
438	—	O	C ₃ H ₇ -i	(6-)CH ₃		165
439	—	O	CH ₃	(6-)CH ₃		157
440	—	O	CH ₃	(6-)CH ₃		168
441	—	O	CH ₃	(6-)CH ₃		164

TABLE 1-continued

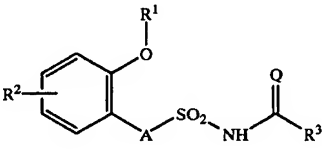
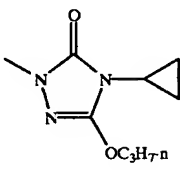
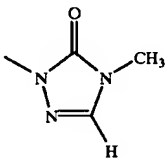
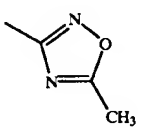
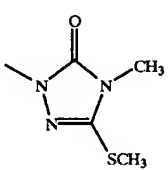
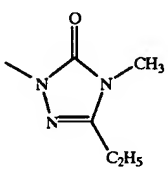
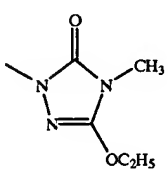
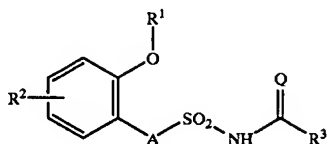
Examples of the compounds of the formula (I)						
Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
<div style="text-align: center;">  <p>(I)</p> </div>						
442	—	O	CH ₃	(6-)CH ₃		125
443	—	O	CH ₃	(6-)CH ₃		162
444	—	O	CH ₃	(6-)OCH ₃		154
445	—	S	CH ₃	(4-)C ₃ H ₇ -i		116
446	—	S	CH ₃	(4-)C ₃ H ₇ -i		110
447	—	S	CH ₃	(4-)C ₃ H ₇ -i		134

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

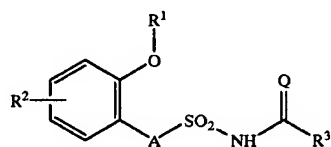


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
448	—	O	CH ₃	(4-)C ₃ H ₇ -i	<p>The structure of R³ for example 448 is a 1,3,4-thiazole ring with a carbonyl group (=O) at position 2, a methyl group (-CH₃) at position 4, and a methylsulfenyl group (-SCH₃) at position 5.</p>	152
449	—	O	CH ₃	(4-)C ₃ H ₇ -i	<p>The structure of R³ for example 449 is a 1,3,4-thiazole ring with a carbonyl group (=O) at position 2, a methyl group (-CH₃) at position 4, and an ethoxymethyl group (-OC₂H₅) at position 5.</p>	159
450	—	O	CH ₃	(4-)C ₃ H ₇ -i	<p>The structure of R³ for example 450 is a 1,3,4-thiazole ring with a carbonyl group (=O) at position 2, a methyl group (-CH₃) at position 4, and an ethyl group (-C₂H₅) at position 5.</p>	150
451	—	O	C ₂ H ₅	(6-)OCH ₃	<p>The structure of R³ for example 451 is a 1,3,4-thiazole ring with a carbonyl group (=O) at position 2, a methyl group (-CH₃) at position 4, and an ethyl group (-C₂H₅) at position 5.</p>	107
452	—	O	C ₂ H ₅	(6-)OCH ₃	<p>The structure of R³ for example 452 is a 1,3,4-thiazole ring with a carbonyl group (=O) at position 2, a methyl group (-CH₃) at position 4, and an ethoxymethyl group (-OC₂H₅) at position 5.</p>	104

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

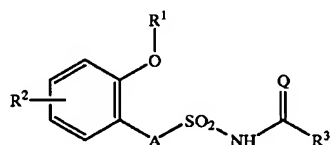


Ex. No.	A	Q	R¹	(position-) R²	R³	Melting point (° C.)
453	—	O	C₂H₅	(6-)OCH₃		100
454	—	S	C₂H₅	(6-)OCH₃		103
455	—	S	C₂H₅	(6-)OCH₃		95
456	—	S	C₂H₅	(6-)OCH₃		105
457	—	O	C₂H₅	(6-)OC₂H₅		130

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

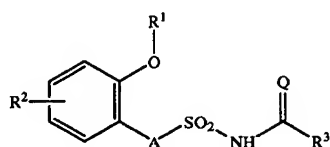


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
458	—	O	C ₂ H ₅	(6-)OC ₂ H ₅		100
459	—	O	C ₃ H ₇ i	(6-)OC ₃ H ₇ i		114
460	—	O	C ₃ H ₇ i	(6-)OC ₃ H ₇ i		125
461	—	S	C ₃ H ₇ i	(6-)OC ₃ H ₇ i		143
462	—	O	C ₂ H ₅	(6-)OC ₂ H ₅		120
463	—	O	—CF ₂ CHFCI	(6-)CH ₃		124

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

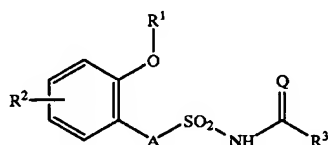


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
464	—	O	—CF ₂ CHFCl	(6-)CH ₃		115
465	—	O	—CF ₂ CHFCl	(6-)CH ₃		150
466	—	O	CH ₃	(3-)CH ₃		178
467	—	O	CH ₃	(3-)Cl		188
468	—	O	C ₂ H ₅	(3-)Cl		159
469	—	O	CH ₃	(3-)F		176

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

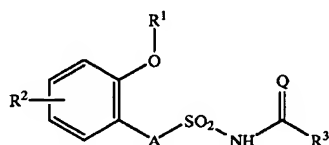


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
470	—	O	C ₂ H ₅	(6-)CF ₃		124
471	—	O	C ₂ H ₅	(6-)CF ₃		34
472	—	O	C ₂ H ₅	(6-)CF ₃		68
473	—	O	C ₃ H ₇ i	(6-)CF ₃		41
474	—	O	C ₂ H ₅	(6-)CF ₃		127
475	—	O	H	(6-)CF ₃		125

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

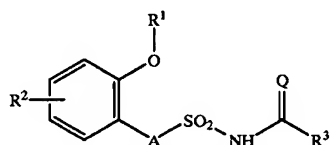


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
476	—	S	C_2H_5	(6-)CH ₃		214 (Na salt)
477	NH	O	C_2H_5	(6-)CH ₃		128
478	NH	O	C_2H_5	(6-)CH ₃		148
479	NH	O	C_3H_7	(6-)CH ₃		127
480	NH	O	C_3H_7	(6-)CH ₃		57
481	NH	O	C_2H_5	(6-)CH ₃		125

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

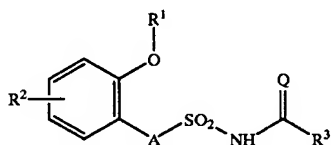


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
482	NH	O	C_3H_7n	(6-)CH ₃		115
483	NH	O	C_3H_7n	(6-)CH ₃		151
484	NH	O	C_2H_5	(5-)CH ₃		132
485	NH	O	C_2H_5	(5-)CH ₃		106
486	NH	O	C_2H_5	(5-)CH ₃		163
487	NH	O	C_2H_5	(5-)CH ₃		137

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

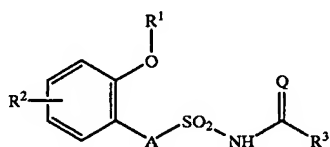


Ex. No.	A	Q	R¹	(position-) R²	R³	Melting point (° C.)
488	—	O	C₂H₅	(6-)C₃H₇-i		166
489	—	O	C₂H₅	(6-)C₃H₇-i		169
490	—	O	C₂H₅	(6-)C₃H₇-i		130
491	NH	O		(6-)CH₃		148
492	NH	O		(6-)CH₃		138
493	NH	O		(6-)CH₃		147

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

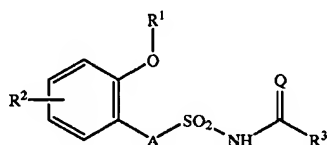


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
494	—	O		(6-)CH ₃		124
495	—	O		(6-)CH ₃		152
496	—	O		(6-)CH ₃		141
497	—	O		(6-)CH ₃		127
498	—	O		(6-)CH ₃		144
499	—	O		(6-)CH ₃		107

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

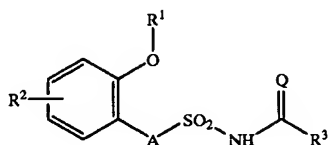


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
500	—	O	C ₃ H ₇ n	(6-)CH ₃		265 (Na salt)
501	—	O	C ₂ H ₅	(6-)CH ₃		269 (Na salt)
502	—	O	C ₃ H ₇ i	(6-)CH ₃		237 (Na salt)
503	—	O		(6-)CH ₃		73
504	—	O	C ₃ H ₇ i	(6-)CH ₃		220 (Na salt)
505	—	O	C ₃ H ₇ i	(6-)CH ₃		140

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

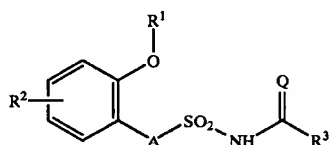


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
506	—	O	C ₄ H ₉ -s	(6-)CH ₃		120
507	—	O	C ₄ H ₉ -s	(6-)CH ₃		117
508	—	O	C ₄ H ₉ -s	(6-)CH ₃		128
509	—	O	C ₄ H ₉ -s	(6-)CH ₃		232
510	—	O	C ₄ H ₉ -s	(6-)CH ₃		268
511	—	O	C ₄ H ₉ -s	(6-)CH ₃		130

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)



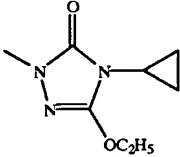
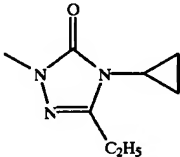
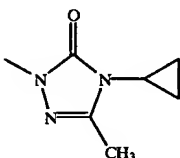
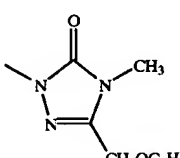
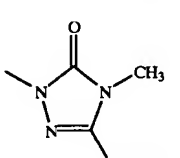
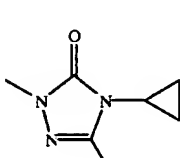
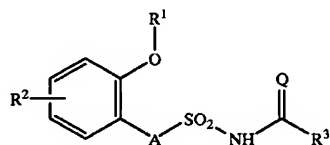
Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
512	—	O	C ₆ H ₅ -s	(6-)CH ₃		137
513	—	O	C ₆ H ₅ -s	(6-)CH ₃		145
514	—	O	C ₆ H ₅ -s	(6-)CH ₃		164
515	—	O	C ₆ H ₅ -s	(6-)CH ₃		89
516	—	O	C ₆ H ₅ -s	(6-)CH ₃		86
517	—	O	C ₆ H ₅ -s	(6-)CH ₃		98

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

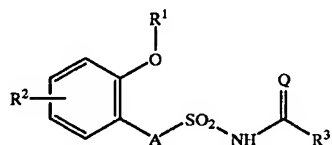


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
518	—	O	C_6H_5	(6-)CH ₃		122
519	—	O	C_6H_5	(6-)CH ₃		135
520	—	O	C_2H_5	(6-)CH ₃		142
521	—	O	CH ₃	(6-)CH ₃		157
522	—	O	C_7H_{17}	(6-)CH ₃		126

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

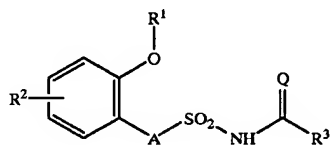


Ex. No.	A	O	R ¹	(position-) R ²	R ³	Melting point (° C.)
523	—	O	C ₄ H ₉ -s	(6-)CH ₃		140
524	—	O	C ₂ H ₅	(6-)CH ₃		164
525	—	O	CH ₃	(6-)CH ₃		166
526	—	O	C ₃ H ₇ -n	(6-)CH ₃		145
527	—	O	CH ₃	(6-)CH ₃		239 (Na salt)

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

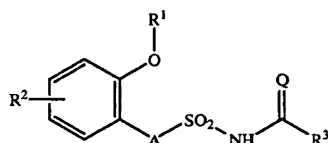


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point ($^{\circ}$ C.)
528	—	O	C_3H_7-n	(6-) CH_3		206 (Na salt)
529	—	O	C_2H_5	(6-) CH_3		211 (Na salt)
530	—	O	CH_3	(6-) C_3H_7-i		166
531	—	O	CH_3	(6-) C_3H_7-i		178
532	—	O	C_3H_7-i	(6-) C_3H_7-i		131
533	—	O	C_3H_7-i	(6-) C_3H_7-i		130

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

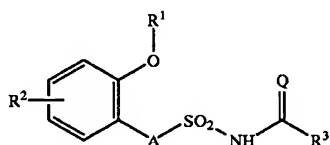


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
534	—	O	C ₃ H ₇ -i	(6-)C ₃ H ₇ -i		133
535	—	O	C ₃ H ₇ -i	(6-)C ₃ H ₇ -i		116
536	—	O	CH ₃	(6-)C ₃ H ₇ -i		192
537	—	O	CH ₃	(6-)C ₃ H ₇ -i		200
538	—	O	CH ₃	(6-)C ₃ H ₇ -i		200
539	—	O	CH ₃	(6-)C ₃ H ₇ -i		105

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

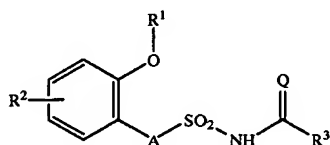


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
540	—	O	CH ₃	(6-)C ₃ H ₇ -i		154
541	—	O	CH ₃	(6-)C ₃ H ₇ -i		152
542	—	O	C ₃ H ₇ -n	(6-)C ₃ H ₇ -i		132
543	—	O	C ₃ H ₇ -n	(6-)C ₃ H ₇ -i		129
544	—	O	C ₃ H ₇ -n	(6-)C ₃ H ₇ -i		179
545	—	O	CH ₃	(6-)C ₃ H ₇ -i		174 (Na salt)

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

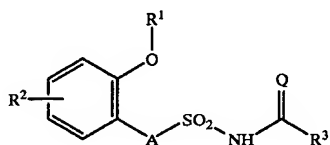


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
546	—	O	C ₃ H _{7-n}	(6-)C ₃ H _{7-i}		158 (Na salt)
547	—	O	C ₂ H ₅	(6-)C ₃ H _{7-i}		200
548	—	O	C ₂ H ₅	(6-)C ₃ H _{7-i}		147
549	—	O	C ₂ H ₅	(6-)C ₃ H _{7-i}		149
550	—	O	C ₂ H ₅	(6-)C ₃ H _{7-i}		136
551	—	O	C ₂ H ₅	(6-)C ₃ H _{7-i}		134

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

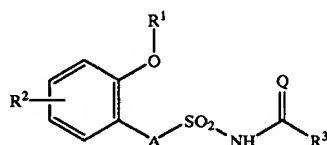


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
552	—	O	C_3H_7-i	(6-) C_3H_7-i		175
553	—	O	C_3H_7-i	(6-) C_3H_7-i		147
554	—	O	C_3H_7-i	(6-) C_3H_7-i		167
555	—	O	C_3H_7-i	(6-) C_3H_7-i		173
556	—	O	C_3H_7-i	(6-) CH_3		188
557	NH	O	CH_3	(6-) CH_3		181

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

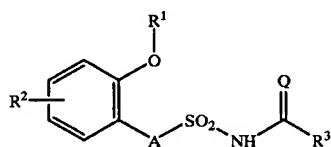


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point ($^{\circ}$ C.)
558	NH	O	C_3H_7i	(6-)CH ₃		136
559	—	O	CH ₃	(6-)CH ₃		241 (Na salt)
560	NH	O	C_3H_7n	(6-)CH ₃		129
561	—	O	C_3H_7i	(6-)OCH ₃		94
562	—	O	C_3H_7i	(6-)OCH ₃		80
563	—	O	C_3H_7i	(6-)OC ₂ H ₅		68

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

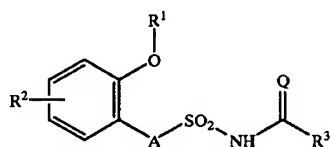


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
564	—	O	C ₃ H ₇ -i	(6-)OC ₂ H ₅		91
565	—	O	C ₃ H ₇ -n	(6-)OCH ₃		123
566	—	O	C ₃ H ₇ -n	(6-)OCH ₃		104
567	—	O	C ₃ H ₇ -n	(6-)OCH ₃		90
568	—	S	C ₃ H ₇ -n	(6-)OCH ₃		107
569	—	O	C ₃ H ₇ -n	(6-)OCH ₃		70

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

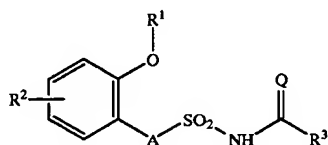


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
570	—	O	C ₂ H ₅	(6-)OC ₂ H ₅		132
571	—	S	CH ₃	(6-)OCH ₃		150
572	—	O	C ₂ H ₅	(6-)OCH ₃		127
573	—	O	C ₃ H ₇ i	(6-)OC ₃ H ₇ i		110
574	—	S	C ₃ H ₇ n	(6-)OC ₃ H ₇ n		130
575	—	S	C ₃ H ₇ n	(6-)OC ₃ H ₇ n		135

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

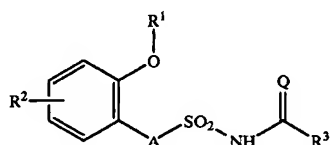


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
576	—	S	C ₃ H _{7-n}	(6-)OC ₃ H _{7-n}		199 (Na salt)
577	—	O	C ₂ H ₅	(6-)OC ₂ H ₅		173
578	—	O	C ₂ H ₅	(6-)OCH ₃		168
579	—	O	C ₃ H _{7-n}	(6-)OC ₃ H _{7-n}		125
580	—	O	C ₃ H _{7-n}	(6-)OC ₂ H ₅		140
581	—	O	C ₃ H _{7-n}	(6-)OCH ₃		115

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

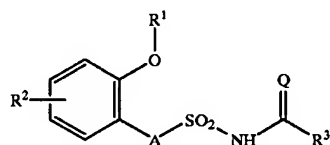


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
582	—	S	C ₃ H _{7-n}	(6-)OCH ₃		111
583	—	S	C ₃ H _{7-n}	(6-)OCH ₃		138
584	—	O	C ₃ H _{7-i}	(6-)OCH ₃		127
585	—	O	C ₃ H _{7-i}	(6-)OC ₂ H ₅		142
586	—	O	C ₃ H _{7-i}	(6-)OC ₂ H ₅		143
587	—	O	C ₂ H ₅	(6-)OCH ₃		104

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

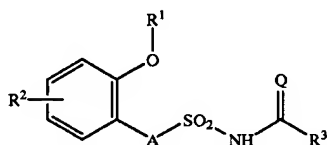


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
588	—	O	C_2H_5	(6-)OCH ₃		118
589	—	O	C_2H_5	(6-)OCH ₃		70
590	—	O	C_2H_5	(6-)OCH ₃		110
591	—	O	C_2H_5	(6-)OCH ₃		156
592	—	O	C_3H_7-n	(6-)OCH ₃		140
593	—	O	C_3H_7-n	(6-)OCH ₃		148

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

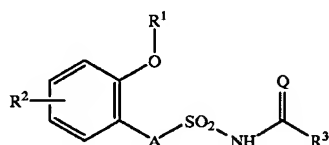


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
594	—	O	C ₃ H _{7-n}	(6-)OCH ₃		145
595	—	O	C ₂ H ₅	(6-)OCH ₃		120
596	—	O	C ₂ H ₅	(6-)OCH ₃		100
597	—	O	C ₂ H ₅	(6-)OCH ₃		130
598	—	O	C ₂ H ₅	(6-)OCH ₃		103
599	—	O	C ₂ H ₅	(6-)OCH ₃		104

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

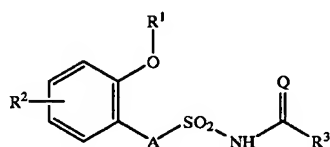


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
600	—	O	C_3H_7n	(6-)OCH ₃		185
601	—	O	C_3H_7n	(6-)OCH ₃		100
602	—	O	C_3H_7n	(6-)OCH ₃		138
603	—	O	C_3H_7n	(6-)OCH ₃		106
604	—	O	C_3H_7n	(6-)OCH ₃		112
605	—	O	C_3H_7n	(6-)OCH ₃		140

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

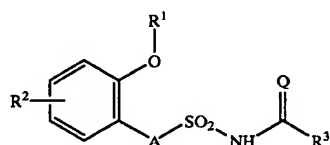


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
606	—	O	C_3H_7n	(6-)OCH ₃		160
607	—	O	C_3H_7n	(6-)OCH ₃		180
608	—	O	C_3H_7n	(6-)OCH ₃		142
609	—	O	C_3H_7n	(6-)OCH ₃		158
610	—	O	C_3H_7n	(6-)OCH ₃		134
611	—	O	C_3H_7i	(6-)OCH ₃		140

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

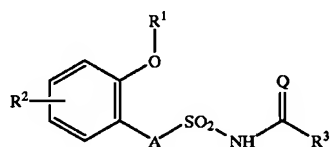


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
612	—	O	C_3H_7-i	(6-)OCH ₃		142
613	—	O	C_3H_7-i	(6-)OCH ₃		148
614	—	O	C_3H_7-i	(6-)OCH ₃		146
615	—	O	C_3H_7-i	(6-)OCH ₃		104
616	—	O	C_2H_5	(6-)OC ₂ H ₅		123
617	—	O	C_3H_7-n	(6-)OC ₃ H _{7-n}		123

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

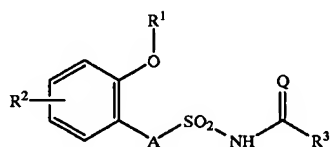


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
618	—	O	C ₃ H _{7-n}	(6-)OC ₃ H _{7-n}		82
619	—	O	C ₃ H _{7-n}	(6-)OC ₃ H _{7-n}		81
620	—	S	C ₃ H _{7-n}	(6-)OC ₃ H _{7-n}		88
621	—	S	C ₂ H ₅	(6-)OCH ₃		145
622	—	S	C ₂ H ₅	(6-)OC ₂ H ₅		147
623	—	S	C ₃ H _{7-n}	(6-)OCH ₃		205

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

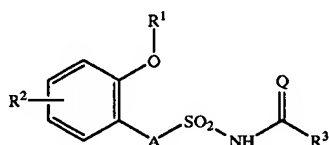


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
624	—	S	C ₃ H _{7-n}	(6-)OC ₂ H ₅		202
625	—	S	C ₃ H _{7-i}	(6-)OC ₃ H _{7-i}		152
626	—	S	C ₂ H ₅	(6-)OC ₂ H ₅		168
627	—	S	C ₃ H _{7-n}	(6-)C ₃ H _{7-n}		145
628	—	S	C ₃ H _{7-n}	(6-)OCH ₃		158
629	—	S	C ₃ H _{7-i}	(6-)OC ₃ H _{7-i}		155

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

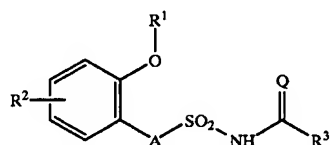


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
630	—	O	C_3H_7i	(6-)OCH ₃		145
631	—	O	C_3H_7i	(6-)OCH ₃		111
632	—	O	C_3H_7i	(6-)OCH ₃		122
633	—	O	C_3H_7i	(6-)OCH ₃		171
634	—	O	C_3H_7i	(6-)OCH ₃		160
635	—	O	C_3H_7i	(6-)OCH ₃		142

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

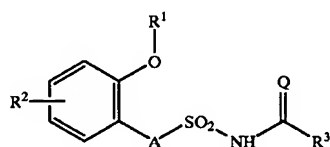


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
636	—	O	C ₂ H ₅	(6-)OC ₂ H ₅		106
637	—	O	C ₂ H ₅	(6-)OC ₂ H ₅		106
638	—	O	C ₂ H ₅	(6-)OC ₂ H ₅		159
639	—	O	C ₂ H ₅	(6-)OC ₂ H ₅		148
640	—	O	C ₂ H ₅	(6-)OC ₂ H ₅		126
641	—	O	C ₂ H ₅	(6-)OC ₂ H ₅		111

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

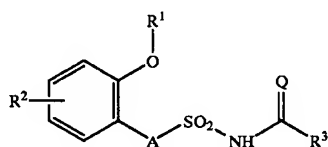


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
642	—	O	C ₂ H ₅	(6-)OC ₂ H ₅		171
643	—	O	C ₂ H ₅	(6-)OC ₂ H ₅		127
644	—	O	C ₂ H ₅	(6-)OC ₂ H ₅		148
645	—	O	C ₂ H ₅	(6-)OC ₂ H ₅		123
646	—	O	C ₃ H _{7-n}	(6-)OC ₃ H _{7-n}		138
647	—	O	C ₃ H _{7-n}	(6-)OC ₃ H _{7-n}		95

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

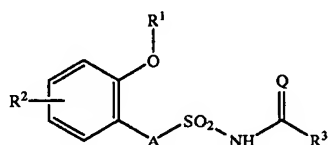


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
654	—	O	C ₃ H ₇ n	(6-)OC ₃ H ₇ n		102
655	—	O	C ₃ H ₇ n	(6-)OC ₃ H ₇ n		98
656	—	O	C ₃ H ₇ i	(6-)OC ₃ H ₇ i		138
657	—	O	C ₃ H ₇ i	(6-)OC ₃ H ₇ i		127
658	—	O	C ₃ H ₇ i	(6-)OC ₃ H ₇ i		160
659	—	O	C ₃ H ₇ i	(6-)OC ₃ H ₇ i		115

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

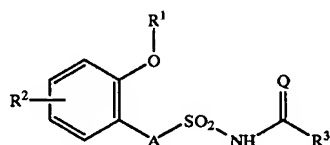


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
660	—	O	C ₃ H ₇ i	(6-)OC ₃ H ₇ i		108
661	—	O	C ₃ H ₇ i	(6-)OC ₃ H ₇ i		154
662	—	O	C ₃ H ₇ i	(6-)OC ₃ H ₇ i		144
663	—	O	C ₂ H ₄ OC ₂ H ₅	(6-)OCH ₃		124
664	—	O	C ₂ H ₄ OC ₂ H ₅	(6-)OCH ₃		138
665	—	O	C ₄ H ₉ n	(6-)OCH ₃		130

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

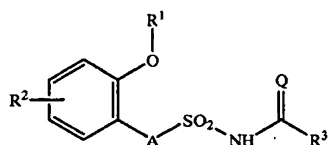


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
666	—	O	C ₄ H ₉ -n	(6-)OCH ₃	 <chem>CN1C=NC(C)=N1S</chem>	132
667	—	O	C ₃ H ₇ -i	(6-)OC ₃ H ₇ -i	 <chem>CN1C=NC(C)=N1OCC</chem>	100
668	—	O	C ₃ H ₇ -i	(6-)OC ₃ H ₇ -i	 <chem>CN1C=NC(C)=N1OCC</chem>	108
669	—	O	C ₃ H ₇ -i	(6-)OC ₃ H ₇ -i	 <chem>CN1C=NC(C)=N1OCC</chem>	130
670	—	O	C ₃ H ₇ -i	(6-)OC ₃ H ₇ -i	 <chem>CN1C=NC(C)=N1OCC</chem>	133
671	—	O	C ₃ H ₇ -i	(6-)OC ₃ H ₇ -i	 <chem>CN1C=NC(C)=N1OCC</chem>	125

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

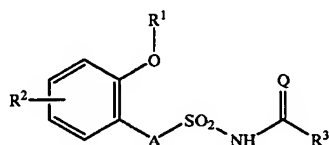


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
672	—	O	C_3H_7 -i	(6-) OC_3H_7 -i		108
673	—	O	C_3H_7 -i	(6-) OC_3H_7 -i		110
674	—	O	C_4H_9 -n	(6-) OCH_3		144
675	—	O	C_4H_9 -n	(6-) OCH_3		116
676	—	O	C_4H_9 -n	(6-) OCH_3		139
677	—	O	C_4H_9 -n	(6-) OCH_3		174

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

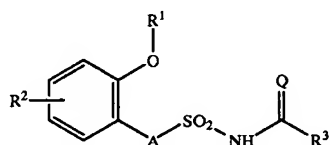


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
678	—	O	C ₄ H ₉ -n	(6-)OCH ₃		149
679	—	O	C ₄ H ₉ -n	(6-)OCH ₃		104
680	—	O	C ₄ H ₉ -n	(6-)OCH ₃		98
681	—	O	C ₄ H ₉ -n	(6-)OCH ₃		112
682	—	O	C ₄ H ₉ -n	(6-)OCH ₃		100
683	—	O	C ₄ H ₉ -n	(6-)OCH ₃		92

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

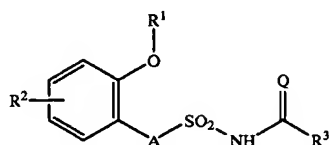


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
684	—	O	C ₄ H ₉ -n	(6-)OCH ₃	<p>Chemical structure of R³ for example 684: A 1,3,4-oxadiazole ring with a methyl group at position 1, a methyl group at position 3, and a (3-iodopropoxy) group at position 5.</p>	115
685	—	O	C ₄ H ₉ -n	(6-)OCH ₃	<p>Chemical structure of R³ for example 685: A 1,3,4-oxadiazole ring with a methyl group at position 1, a methyl group at position 3, and a (2-methoxyethyl) group at position 5.</p>	99
686	—	O	C ₄ H ₉ -n	(6-)OCH ₃	<p>Chemical structure of R³ for example 686: A 1,3,4-oxadiazole ring with a methyl group at position 1, an ethyl group at position 3, and a methoxy group at position 5.</p>	102
687	—	O	C ₄ H ₉ -n	(6-)OCH ₃	<p>Chemical structure of R³ for example 687: A 1,3,4-oxadiazole ring with a methyl group at position 1, a cyclopropyl group at position 3, and a (2-methoxyethyl) group at position 5.</p>	106
688	—	O	C ₂ H ₄ OC ₂ H ₅	(6-)OCH ₃	<p>Chemical structure of R³ for example 688: A 1,3,4-oxadiazole ring with a methyl group at position 1, a methyl group at position 3, and a bromine atom at position 5.</p>	120
689	—	O	C ₂ H ₄ OC ₂ H ₅	(6-)OCH ₃	<p>Chemical structure of R³ for example 689: A 1,3,4-oxadiazole ring with a methyl group at position 1, a methyl group at position 3, and a methoxy group at position 5.</p>	98

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

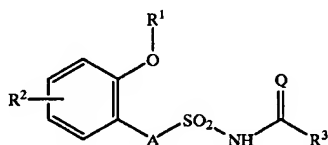


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
690	—	O	C ₂ H ₄ OC ₂ H ₅	(6-)OCH ₃		138
691	—	O	C ₄ H ₉ -i	(6-)OC ₂ H ₅		118
692	—	O	C ₂ H ₄ OC ₂ H ₅	(6-)OCH ₃		138
693	—	O	C ₂ H ₄ OC ₂ H ₅	(6-)OCH ₃		94
694	—	O	C ₂ H ₄ OC ₂ H ₅	(6-)OCH ₃		94
695	—	O	C ₂ H ₄ OC ₂ H ₅	(6-)OCH ₃		65

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

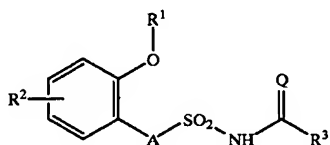


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
696	—	O	$C_2H_4OC_2H_5$	(6-)OCH ₃		60
697	—	O	$C_2H_4OC_2H_5$	(6-)OCH ₃		104
698	—	O	C_4H_9-n	(6-)OC ₄ H _{9-n}		104
699	—	O	C_4H_9-n	(6-)OC ₄ H _{9-n}		137
700	—	O	C_4H_9-n	(6-)OC ₄ H _{9-n}		70
701	—	O	C_4H_9-n	(6-)OC ₄ H _{9-n}		110

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

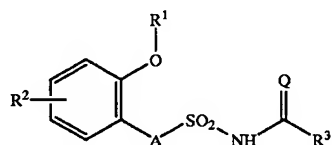


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
702	—	O	C ₆ H ₅ -n	(6-)OC ₄ H ₉ -n		102
703	—	O	C ₃ H ₇ -i	(6-)OC ₃ H ₇ -i		137
704	—	O	C ₃ H ₇ -i	(6-)OCH ₃		160
705	—	O	C ₃ H ₇ -n	(6-)OC ₃ H ₇ -n		94
706	—	O	C ₃ H ₇ -n	(6-)OC ₂ H ₅		155
707	—	O	C ₃ H ₇ -n	(6-)OC ₂ H ₅		130

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

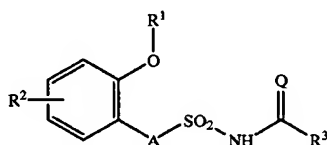


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
708	—	O	C ₃ H _{7-n}	(6-)OC ₂ H ₅		142
709	—	O	C ₃ H _{7-n}	(6-)OC ₂ H ₅		85
710	—	O	C ₃ H _{7-n}	(6-)OC ₂ H ₅		106
711	—	O	C ₃ H _{7-n}	(6-)OC ₂ H ₅		87
712	—	S	C ₃ H _{7-i}	(6-)OCH ₃		120
713	—	S	C ₄ H _{9-n}	(6-)OCH ₃		143

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

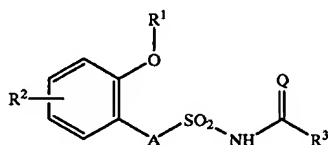


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
714	—	S	C ₃ H ₇ -i	(6-)OC ₃ H ₇ -i		152
715	—	S	C ₂ H ₄ OC ₂ H ₅	(6-)OCH ₃		112
716	—	S	C ₃ H ₇ -i	(6-)OCH ₃		130
717	—	S	C ₃ H ₇ -i	(6-)OCH ₃		165
718	—	S	C ₃ H ₇ -i	(6-)OC ₃ H ₇ -i		161
719	—	S	C ₈ H ₅ -n	(6-)OCH ₃		111

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

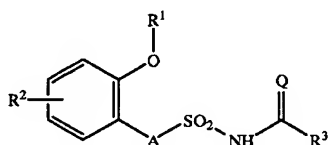


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
720	—	S	C_2H_5	(6-)OCH ₃		156
721	—	S	$C_2H_4OC_2H_5$	(6-)OCH ₃		137
722	—	S	CH ₃	(6-)OCH ₃		163
723	—	S	C_3H_7-n	(6-)OCH ₃		113
724	—	S	C_3H_7-n	(6-)OCH ₃		130
725	—	S	C_4H_9-n	(6-)OCH ₃		154

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

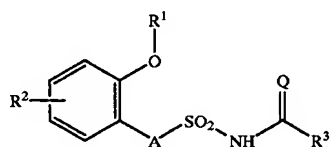


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
726	—	S	C ₃ H ₇ -i	(6-)OC ₃ H ₇ -i		157
727	—	S	C ₃ H ₇ -n	(6-)OC ₃ H ₇ -n		142
728	—	S	C ₂ H ₅	(6-)OC ₂ H ₅		162
729	—	S	CH ₃	(6-)OCH ₃		157
730	—	S	C ₂ H ₅	(6-)OCH ₃		108
731	—	S	CH ₃	(6-)OCH ₃		172

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

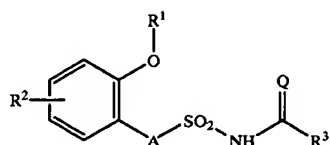


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
732	—	S	CH_3	(6-) OCH_3		147
733	—	S	C_3H_7n	(6-) OC_3H_7n		160
734	—	S	C_3H_7n	(6-) OC_3H_7n		103
735	—	O	C_3H_7n	(6-) OC_3H_7n		172
736	—	S	C_2H_5	(6-) OC_2H_5		137
737	—	S	C_2H_5	(6-) OC_2H_5		156

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

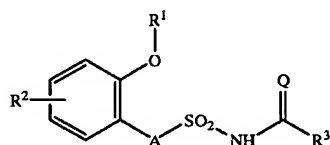


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
738	—	S	$C_2H_4OC_2H_5$	(6-)OCH ₃		103
739	—	S	C_3H_7i	(6-)OC ₃ H _{7i}		134
740	—	S	C_4H_9n	(6-)OC ₄ H _{9n}		87
741	—	S	C_4H_9n	(6-)OC ₄ H _{9n}		110
742	—	S	C_4H_9n	(6-)OC ₄ H _{9n}		88
743	—	S	C_4H_9n	(6-)OC ₄ H _{9n}		98

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

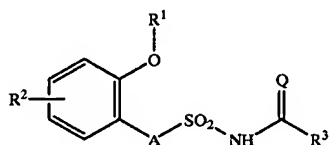


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
744	—	S	C_4H_9-n	(6-)OC ₄ H _{9-n}		98
745	—	S	C_4H_9-n	(6-)OC ₄ H _{9-n}		88
746	—	S	C_4H_9-n	(6-)OC ₄ H _{9-n}		104
747	—	S	C_4H_9-n	(6-)OC ₄ H _{9-n}		75
748	—	O	C_2H_5	(6-)OC ₂ H ₅		145
749	—	S	C_3H_7-i	(6-)OC ₂ H ₅		131

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

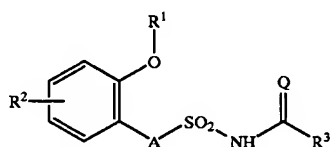


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
750	—	S	C ₃ H ₇ i	(6-)OC ₂ H ₅		158
751	—	S	C ₃ H ₇ i	(6-)OC ₂ H ₅		132
752	—	S	C ₃ H ₇ i	(6-)OC ₂ H ₅		142
753	—	S	C ₃ H ₇ i	(6-)OC ₂ H ₅		130
754	—	S	C ₃ H ₇ i	(6-)OC ₂ H ₅		170
755	—	S	C ₃ H ₇ n	(6-)OC ₂ H ₅		152

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

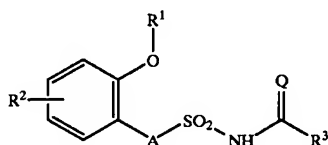


Ex. No.	A	Q	R¹	(position-) R²	R³	Melting point (° C.)
756	—	S	C ₃ H ₇ n	(6-)OC ₂ H ₅	 <chem>CN1C(=O)N(C)C(S)N1</chem>	138
757	—	S	C ₃ H ₇ n	(6-)OC ₂ H ₅	 <chem>CN1C(=O)N(C)CCN1</chem>	130
758	—	S	C ₃ H ₇ n	(6-)OC ₂ H ₅	 <chem>CN1C(=O)N(C)COC(F)(F)F1</chem>	150
759	—	S	C ₃ H ₇ n	(6-)OC ₂ H ₅	 <chem>CN1C(=O)N(C)COC1C2CC2</chem>	156
760	—	S	C ₃ H ₇ n	(6-)OC ₂ H ₅	 <chem>CN1C(=O)N(C)CCS1</chem>	110
761	—	S	C ₃ H ₇ n	(6-)OC ₂ H ₅	 <chem>CN1C(=O)N(C)C(C)N1</chem>	120

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

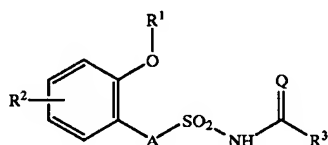


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
762	—	S	C_4H_9-n	(6-)OC ₄ H _{9-n}		104
763	—	S	C_3H_7-i	(6-)OCH ₃		105
764	—	S	C_3H_7-n	(6-)OC ₃ H _{7-n}		120
765	—	S	C_3H_7-n	(6-)OCH ₃		135
766	—	S	C_3H_7-n	(6-)OC ₃ H _{7-n}		116
767	—	S	C_3H_7-i	(6-)OC ₃ H _{7-i}		110

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

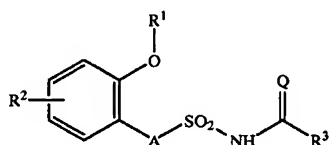


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
768	—	S	C ₂ H ₄ OC ₂ H ₅	(6-)OCH ₃		95
769	—	S	C ₃ H ₇ i	(6-)OCH ₃		112
770	—	O	C ₃ H ₇ i	(6-)OC ₂ H ₅		70
771	—	O	C ₃ H ₇ i	(6-)OC ₂ H ₅		132
772	—	O	C ₃ H ₇ i	(6-)OC ₂ H ₅		75
773	—	O	C ₃ H ₇ i	(6-)OC ₂ H ₅		118

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

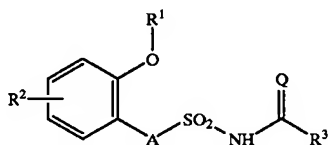


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
774	—	O	C_3H_7-i	(6-)OC ₂ H ₅		85
775	—	O	C_3H_7-i	(6-)OC ₂ H ₅		130
776	—	O	C_3H_7-i	(6-)OC ₂ H ₅		120
777	—	O	C_3H_7-i	(6-)OC ₂ H ₅		124
778	—	O	C_3H_7-i	(6-)OC ₂ H ₅		130
779	—	O	C_3H_7-i	(6-)OC ₂ H ₅		100

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

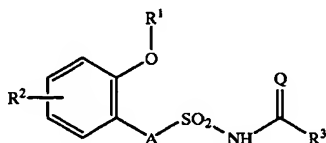


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
780	—	O	C ₂ H ₅	(6-)OCH ₃		172
781	—	O	C ₂ H ₅	(6-)OCH ₃		164
782	—	O	C ₃ H ₇ i	(6-)OC ₃ H ₇ i		118
783	—	O	C ₃ H ₇ i	(6-)OC ₃ H ₇ i		88
784	—	O	C ₄ H ₉ s	(6-)OCH ₃		124
785	—	O	C ₄ H ₉ s	(6-)OCH ₃		100

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

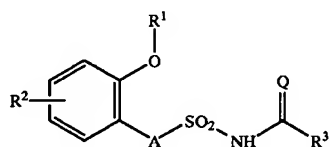


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
786	—	O	C ₄ H ₉ -n	(6-)OC ₂ H ₅		106
787	—	O	C ₄ H ₉ -n	(6-)OC ₂ H ₅		108
788	—	O	C ₄ H ₉ -n	(6-)OC ₂ H ₅		105
789	—	O	C ₄ H ₉ -s	(6-)OCH ₃		112
790	—	O	C ₄ H ₉ -n	(6-)OC ₂ H ₅		80
791	—	O	C ₄ H ₉ -s	(6-)OCH ₃		130

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

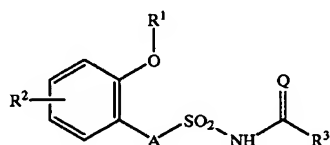


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
792	—	O	C_4H_9-n	(6-)OC ₂ H ₅		120
793	—	O	C_4H_9-n	(6-)OC ₂ H ₅		95
794	—	O	C_4H_9-n	(6-)OC ₂ H ₅		96
795	—	O	C_4H_9-s	(6-)OCH ₃		130
796	—	G	C_4H_9-s	(6-)OCH ₃		118
797	—	O	C_4H_9-s	(6-)OCH ₃		134

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

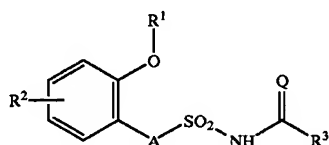


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
798	—	O	C_6H_5	(6-)OCH ₃		118
799	—	O	C_4H_9	(6-)OC ₂ H ₅		90
800	—	O	C_6H_5	(6-)OCH ₃		78
801	—	O	C_6H_5	(6-)OCH ₃		112
802	—	O	C_6H_5	(6-)OCH ₃		78
803	—	O	C_6H_5	(6-)OCH ₃		80

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

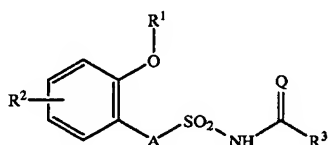


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
804	—	O	C ₄ H ₉ -n	(6-)OC ₄ H ₉ -n		55
805	—	O	C ₄ H ₉ -n	(6-)OC ₄ H ₉ -n		100
806	—	O	C ₄ H ₉ -n	(6-)OC ₄ H ₉ -n		92
807	—	O	C ₃ H ₇ -n	(6-)OC ₂ H ₅		74
808	—	O	C ₃ H ₇ -n	(6-)OC ₂ H ₅		143
809	—	O	C ₃ H ₇ -n	(6-)OC ₂ H ₅		102

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

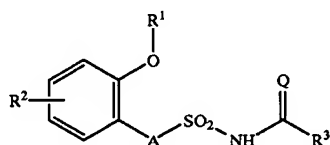


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
810	—	O	C ₃ H _{7-n}	(6-)OC ₂ H ₅		95
811	—	O	C ₄ H _{9-n}	(6-)OC ₄ H _{9-n}		82
812	—	O	C ₄ H _{9-n}	(6-)OC ₂ H ₅		92
813	—	O	C ₄ H _{9-n}	(6-)OCH ₃		90
814	—	O	C ₄ H _{9-n}	(6-)OCH ₃		124
815	—	O	C ₄ H _{9-n}	(6-)OCH ₃		85

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

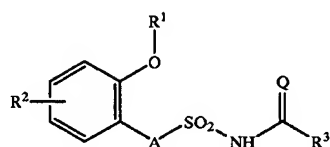


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
816	—	O	C_6H_5	(6-)OCH ₃		90
817	—	O	$C_2H_4OC_2H_5$	(6-)OCH ₃		90
818	—	O		(6-)OCH ₃		165
819	—	O		(6-)OCH ₃		130
820	—	O		(6-)OCH ₃		149
821	—	O	C_6H_5	(6-)OC ₄ H ₉		125

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

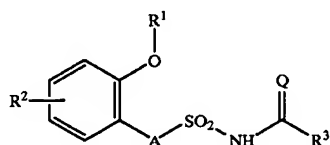


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
822	—	O		(6-)OCH ₃		112
823	—	O		(6-)OCH ₃		156
824	—	O		(6-)OCH ₃		148
825	—	O		(6-)OCH ₃		145
826	—	O		(6-)OCH ₃		156
827	—	O		(6-)OCH ₃		126

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

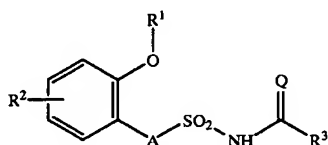


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
828	—	O		(6-)OCH ₃		134
829	—	O		(6-)OCH ₃		114
830	—	O		(6-)OCH ₃		141
831	—	O	C ₄ H ₉ -i	(6-)OC ₂ H ₅		125
832	—	O	C ₄ H ₉ -i	(6-)OC ₂ H ₅		128
833	—	O	C ₄ H ₉ -i	(6-)OC ₂ H ₅		106

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

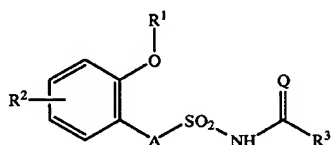


Ex. No.	A	Q	R¹	(position-) R²	R³	Melting point (° C.)
834	—	O	C ₄ H ₉ -i	(6-)OC ₂ H ₅		88
835	—	O	C ₄ H ₉ -i	(6-)OC ₂ H ₅		112
836	—	O	C ₄ H ₉ -i	(6-)OC ₂ H ₅		125
837	—	O	C ₄ H ₉ -i	(6-)OC ₂ H ₅		106
838	—	O	H	(6-)OH		172
839	—	O	C ₄ H ₉ -i	(6-)OCH ₃		102

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

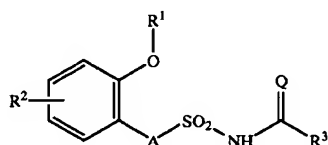


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
840	—	O	C ₆ H ₅ -i	(6-)OCH ₃		114
841	—	O	C ₆ H ₅ -i	(6-)OCH ₃		124
842	—	O	C ₆ H ₅ -i	(6-)OCH ₃		98
843	—	O	C ₆ H ₅ -i	(6-)OCH ₃		146
844	—	O	C ₆ H ₅ -i	(6-)OCH ₃		97
845	—	O	C ₆ H ₅ -i	(6-)OCH ₃		117

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

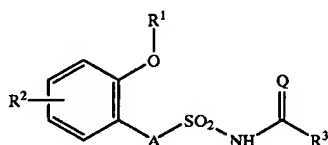


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
846	—	O	C ₄ H ₉ i	(6-)OCH ₃		142
847	—	O	C ₄ H ₉ i	(6-)OCH ₃		109
848	—	S	C ₂ H ₅	(6-)OCH ₃		138
849	—	S	C ₂ H ₅	(6-)OCH ₃		135
850	—	S	C ₂ H ₅	(6-)OCH ₃		155
851	—	S	C ₂ H ₅	(6-)OCH ₃		160

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

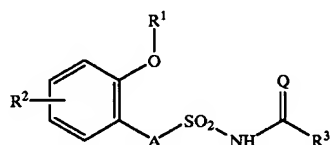


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
852	—	S	C ₄ H ₉ -s	(6-)OCH ₃		108
853	—	S	C ₄ H ₉ -s	(6-)OCH ₃		143
854	—	S	C ₄ H ₉ -s	(6-)OCH ₃		105
855	—	S	C ₄ H ₉ -s	(6-)OCH ₃		65
856	—	S	C ₄ H ₉ -s	(6-)OCH ₃		114
857	—	O	C ₂ H ₅	(3-)CH ₃		179

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

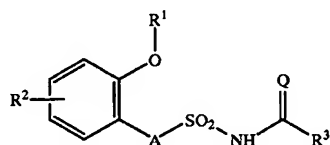


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
858	—	S	C ₄ H ₉ -s	(6-)OCH ₃		146
859	—	S	C ₃ H ₇ -i	(6-)OCH ₃		146
860	—	S	C ₄ H ₉ -n	(6-)OCH ₃		132
861	—	S	C ₄ H ₉ -n	(6-)OCH ₃		112
862	—	S	C ₄ H ₉ -n	(6-)OCH ₃		100
863	—	S	C ₄ H ₉ -n	(6-)OCH ₃		138

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

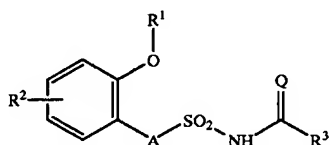


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
864	—	S	C ₃ H ₇ -i	(6-)OC ₂ H ₅		155
865	—	S	C ₃ H ₇ -i	(6-)OC ₂ H ₅		140
866	—	S	C ₄ H ₉ -n	(6-)OCH ₃		140
867	—	O	C ₄ H ₉ -i	(6-)OCH ₃		131
868	—	O	C ₄ H ₉ -i	(6-)OCH ₃		135
869	—	O	C ₄ H ₉ -i	(6-)OCH ₃		137

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

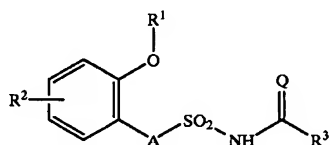


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
870	—	O	C ₄ H _{9-n}	(6-)OC ₄ H _{9-n}		123
871	—	O	C ₄ H _{9-n}	(6-)OC ₂ H ₅		118
872	—	O		(6-)OCH ₃		164
873	—	O	C ₃ H _{7-i}	(6-)OCH ₃		150
874	—	O	C ₃ H _{7-i}	(6-)OC ₃ H _{7-i}		148
875	—	O	C ₃ H _{7-n}	(6-)OCH ₃		147

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

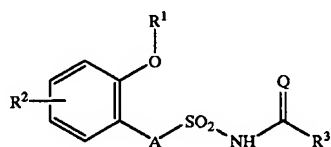


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
876	—	O	C ₂ H ₅	(6-)OC ₂ H ₅		108
877	—	O	C ₃ H ₇ i	(6-)OC ₃ H ₇ i		108
878	—	O	C ₃ H ₇ i	(6-)OC ₃ H ₇ i		148
879	—	O	C ₂ H ₅	(6-)OC ₂ H ₅		176
880	—	O	C ₃ H ₇ n	(6-)OCH ₃		144
881	—	O	C ₃ H ₇ i	(6-)OC ₃ H ₇ i		167

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

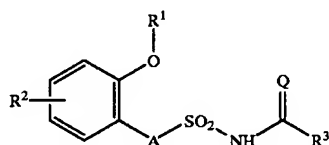


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
882	—	O	C ₂ H ₅	(6-)OC ₂ H ₅		135
883	—	O	C ₃ H ₇ n	(6-)OCH ₃		100
884	—	O	C ₂ H ₅	(6-)OC ₂ H ₅		158
885	—	O	C ₃ H ₇ n	(6-)OCH ₃		108
886	—	O	C ₃ H ₇ i	(6-)OC ₃ H ₇ i		164

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)



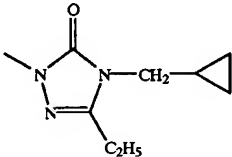
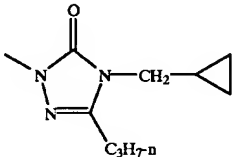
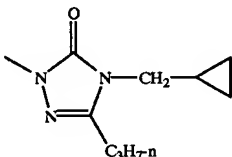
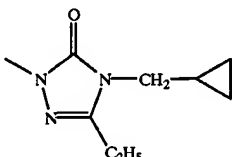
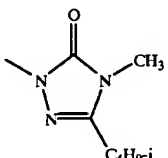
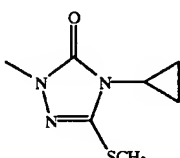
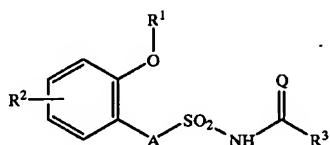
Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
887	—	O	C_3H_7i	(6-)OC ₃ H _{7i}		157
888	—	O	C_3H_7n	(6-)OCH ₃		113
889	—	O	C_3H_7i	(6-)OC ₃ H _{7i}		132
890	—	O	C_3H_7n	(6-)OCH ₃		92
891	—	O	C_3H_7i	(6-)OC ₃ H _{7i}		141
892	—	O	C_3H_7i	(6-)OC ₃ H _{7i}		159

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

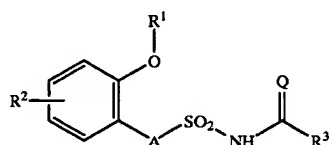


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
893	—	O	C_2H_5	(6-)OC $_2H_5$		139
894	—	O	C_2H_5	(6-)OC $_2H_5$		150
895	—	O	C_3H_7-n	(6-)OCH $_3$		126
896	—	O	C_3H_7-i	(6-)CH $_3$		148
897	—	O	C_3H_7-i	(6-)CH $_3$		157
898	—	O	C_3H_7-i	(6-)CH $_3$		125

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

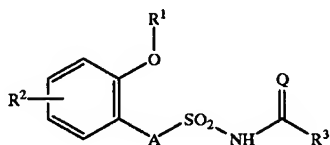


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
899	NH	O	CH ₃	(6-)OCH ₃		182
900	NH	O	CH ₃	(6-)OCH ₃		175
901	—	O	CF ₃	(6-)CH ₃		198 (Na salt)
902	—	O	CF ₃	(6-)CH ₃		129
903	—	O	CF ₃	(6-)CH ₃		149
904	—	O	CF ₃	(6-)CH ₃		163

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

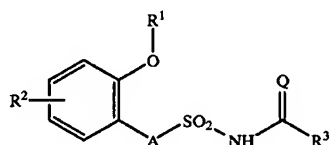


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
905	—	O	CF ₃	(6-)C ₂ H ₅		121
906	—	O	C ₃ H ₇ i	(6-)CH ₃		170
907	—	O	C ₃ H ₇ i	(6-)CH ₃		125
908	—	O	C ₃ H ₇ i	(6-)CH ₃		129
909	—	O	C ₃ H ₇ i	(6-)CH ₃		156
910	—	O	CH ₃	(6-)OCH ₃		157

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

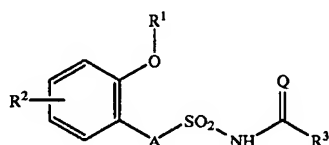


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
911	—	O	CH ₃	(6-)OCH ₃		177
912	—	O	CH ₃	(6-)OCH ₃		172
913	—	O	C ₃ H ₇ i	(6-)CH ₃		132
914	—	O	C ₃ H ₇ i	(6-)CH ₃		153
915	—	O	C ₃ H ₇ i	(6-)CH ₃		150
916	—	O	C ₃ H ₇ i	(6-)CH ₃		110

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

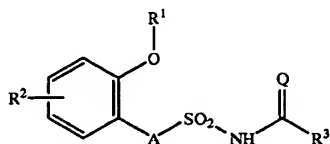


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
917	—	O	C ₃ H ₇ i	(6-)CH ₃		131
918	—	O	C ₃ H ₇ i	(6-)CH ₃		133
919	—	O	CH ₃	(6-)OCH ₃		153
920	—	O	C ₃ H ₇ i	(6-)CH ₃		122
921	—	O	C ₃ H ₇ i	(6-)CH ₃		147
922	—	O	CH ₃	(6-)OCH ₃		160

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

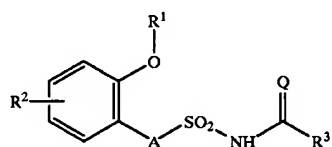


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
923	—	O	CH ₃	(6-)OCH ₃		182
924	—	O	CH ₃	(6-)OCH ₃		142
925	—	O	CH ₃	(6-)OCH ₃		178
926	—	O	C ₃ H ₇ i	(6-)CH ₃		151
927	—	O	CH ₃	(6-)OCH ₃		178
928	—	O	CH ₃	(6-)OCH ₃		130

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

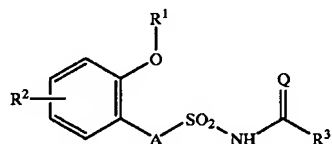


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
929	—	O	CH_3	(6-)OCH ₃		124
930	—	O	CH_3	(6-)OCH ₃		153
931	—	O	CH_3	(6-)OCH ₃		157
932	—	O	CH_3	(6-)OCH ₃		114
933	—	O	CH_3	(6-)OCH ₃		130
934	—	O	C_3H_7-i	(6-)CH ₃		151

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

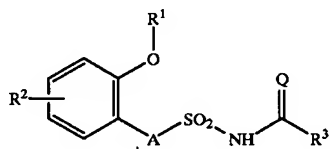


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
935	—	O	C ₃ H ₇ i	(6-)CH ₃		153
936	—	O	CH ₃	(6-)OCH ₃		167
937	—	O	CF ₂ Cl	(6-)CH ₃		180 (Na salt)
938	—	O	CH ₃	(6-)OCH ₃		157 (Na salt)
939	NH	O	CH ₂ CH ₂ F	(6-)CH ₃		163
940	NH	O	CH ₂ CHF ₂	(6-)CH ₃		160

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

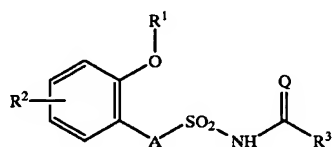


Ex. No.	A	Q	R^1	(position-) R^2	R^3	Melting point (° C.)
941	NH	O	CF_2CHFCI	(6-)CH ₃	 <chem>CN1C(=O)N(C)C(=O)N1COC</chem>	88
942	—	O	CH_2CH_2F	(6-)CH ₃	 <chem>CN1C(=O)N(C)C(=O)N1COC</chem>	178
943	—	O	CH_2CH_2F	(6-)CH ₃	 <chem>CN1C(=O)N(C)C(=O)N1COC(C)C</chem>	135
944	—	O	CH_2CH_2F	(6-)CH ₃	 <chem>CN1C(=O)N(C)C(=O)N1COC(C)C</chem>	127
945	—	O	CH_2CH_2F	(6-)CH ₃	 <chem>CN1C(=O)N(C)C(=O)N1COC(C)C</chem>	139
946	—	O	CH_2CH_2F	(6-)CH ₃	 <chem>CN1C(=O)N(C)C(=O)N1C</chem>	280 (Na salt)

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

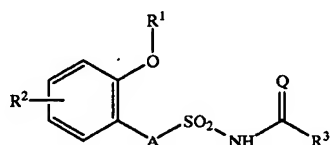


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
947	—	O	CH ₂ CH ₂ F	(6-)CH ₃		171
948	—	O	CH ₂ CH ₂ F	(6-)CH ₃		144
949	—	O	CH ₂ CHF ₂	(6-)CH ₃		273 (Na salt)
950	—	O	CH ₂ CHF ₂	(6-)CH ₃		181
951	—	O	CH ₂ CHF ₂	(6-)CH ₃		142
952	—	O	CH ₂ CHF ₂	(6-)CH ₃		114

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

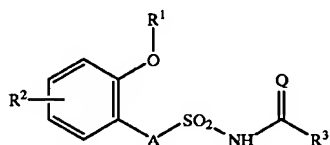


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
953	—	O	CH ₂ CHF ₂	(6-)CH ₃	 <chem>CC1=CN(C(=O)OC(C)C)C=N1</chem>	108
954	—	O	CH ₂ CHF ₂	(6-)CH ₃	 <chem>CC1=CN(C(=O)OC)C=N1C2CC2</chem>	185
955	—	O	CH ₂ CHF ₂	(6-)CH ₃	 <chem>CC1=CN(C(=O)OC(C)C)C=N1C2CC2</chem>	150
956	—	O	CF ₃	(6-)C ₂ H ₅	 <chem>CC1=CN(C(=O)OCC)C=N1C</chem>	143 (Na salt)
957	—	O	CF ₃	(6-)CH ₃	 <chem>CC1=CN(C(=O)OCC(F)(F)F)C=N1C</chem>	155
958	—	O	CF ₃	(6-)CH ₃	 <chem>CC1=CN(C(=O)OCC2CC2)C=N1C</chem>	112

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

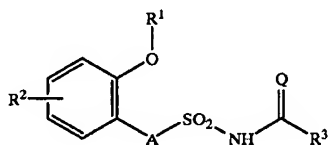


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
959	—	O	CF ₃	(6-)CH ₃		166
960	—	O	CF ₃	(6-)CH ₃		137
961	—	O	CF ₃	(6-)CH ₃		132
962	—	O	CF ₃	(6-)CH ₃		172
963	—	O	CF ₃	(6-)CH ₃		139
964	—	O	CF ₃	(6-)CH ₃		130

TABLE 1-continued

Examples of the compounds of the formula (I)

(I)

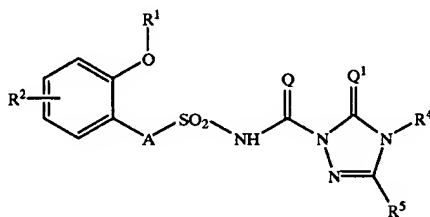


Ex. No.	A	Q	R ¹	(position-) R ²	R ³	Melting point (° C.)
965	—	O	C ₂ H ₅	(3-)CH ₃		184

TABLE 1A

Examples of the compounds of the formula (IA)

(IA)

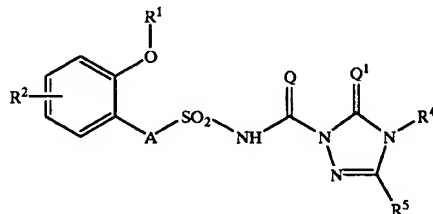


Ex. No.	A	Q	R ¹	(Position-) R ²	Q ¹	R ⁴	R ⁵	Melting point ° C.
966	NH	O	C ₄ H ₉	(6-)CH ₃	O		OC ₂ H ₅	121
967	NH	O	CH ₂ CH(CH ₃) ₂	(6-)CH ₃	O		OC ₂ H ₅	127
968	NH	O	CH ₂ CF ₃	(6-)CH ₃	O	CH ₃	OCH ₃	164
969	NH	O	CH ₂ CF ₃	(6-)CH ₃	O	CH ₃	OC ₂ H ₅	138
970	NH	O	CH ₂ CF ₃	(6-)CH ₃	O	CH ₃	OC ₃ H ₇	123
971	NH	O	CH ₂ CF ₃	(6-)CH ₃	O	CH ₃	OCH(CH ₃) ₂	141
972	—	O	CH ₂ CH ₂ F	(3-)Cl	O	CH ₃	OC ₂ H ₅	139
973	—	O	CH ₂ CH ₂ F	(3-)CH ₃	O	CH ₃	OC ₂ H ₅	154
974	—	O	C ₃ H ₇ -i	(6-)CF ₃	O		C ₄ H ₉ -s	
975	—	O	C ₃ H ₇ -i	(6-)CF ₃	O	CH ₃	C ₄ H ₉ -s	
976	—	O	C ₃ H ₇ -i	(6-)CF ₃	O	CH ₃	C ₄ H ₉	
977	—	O	C ₃ H ₇ -i	(6-)CF ₃	O	CH ₃	C ₄ H ₉ -i	
978	—	S	C ₃ H ₇ -i	(6-)CF ₃	O	CH ₃	C ₂ H ₅	
979	—	S	C ₃ H ₇	(6-)CH ₃	O		OC ₂ H ₅	

TABLE 1A-continued

Examples of the compounds of the formula (IA)

(IA)

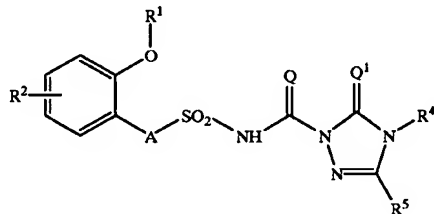


Ex. No.	A	Q	R ¹	(Position-) R ²	Q ¹	R ⁴	R ⁵	Melting point ° C.
980	—	S	C ₃ H ₇	(6-)CH ₃	O	C ₂ H ₅	OC ₂ H ₅	
981	—	3	C ₃ H _{7-i}	(6-)CH ₃	O		OC ₂ H ₅	
982	—	S	C ₃ H _{7-i}	(6-)CH ₃	O	C ₂ H ₅	OC ₂ H ₅	
983	—	S	C ₃ H ₇	(6-)CH ₃	O	CH ₃	OC ₃ H _{7-i}	
984	—	S	C ₃ H _{7-i}	(6-)CH ₃	O	CH ₃	OC ₃ H _{7-i}	
985	—	S	C ₃ H ₇	(6-)CH ₃	O	CH ₃	OC ₃ H ₇	
986	—	S	C ₃ H _{7-i}	(6-)CH ₃	O	CH ₃	OC ₃ H ₇	
987	—	S	C ₃ H ₇	(6-)CH ₃	O		OCH ₃	
988	—	O	CH ₃	(6-)CF ₃	O		C ₄ H _{9-s}	
989	—	O	C ₂ H ₅	(6-)CF ₃	O		C ₄ H _{9-s}	
990	—	O	C ₃ H ₇	(6-)CF ₃	O		C ₄ H _{9-s}	
991	—	O	CH ₃	(6-)CF ₃	O	CH ₃	C ₄ H _{9-s}	
992	—	O	C ₂ H ₅	(6-)CF ₃	O	CH ₃	C ₄ H _{9-s}	
993	—	O	CH ₃	(6-)CF ₃	O	CH ₃	C ₄ H ₉	
994	—	O	CH ₃	(6-)CF ₃	O		C ₄ H _{9-t}	
995	—	O	CH ₃	(6-)CF ₃	O		OC ₃ H ₇	
996	—	O	C ₂ H ₅	(6-)CF ₃	O		OC ₃ H ₇	
997	—	O	C ₃ H ₇	(6-)CF ₃	O		OC ₃ H ₇	
998	—	O	C ₂ H ₅	(6-)CF ₃	O	CH ₃	C ₄ H ₉	
999	—	O	C ₃ H ₇	(6-)CF ₃	O	CH ₃	C ₄ H ₉	
1000	—	O	CH ₃	(6-)CF ₃	O	CH ₃	C ₄ H _{9-i}	
1001	—	S	CH ₃	(6-)CF ₃	O	CH ₃	SCH ₃	
1002	—	S	CH ₃	(6-)CF ₃	O		OC ₂ H ₅	
1003	—	S	CH ₃	(6-)CF ₃	O	CH ₃	OC ₃ H _{7-i}	
1004	—	S	CH ₃	(6-)CF ₃	O	C ₂ H ₅	OC ₂ H ₅	

TABLE 1A-continued

Examples of the compounds of the formula (IA)

(IA)

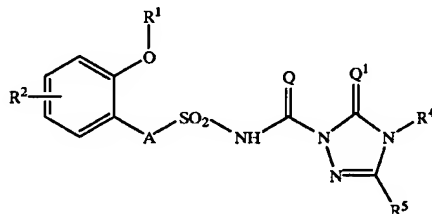


Ex. No.	A	Q	R ¹	(Position-) R ²	Q ¹	R ⁴	R ⁵	Melting point ° C.
1005	—	S	CH ₃	(6-)CF ₃	O		OC ₃ H ₇	
1006	—	S	CH ₃	(6-)CF ₃	O		OC ₃ H ₇ -i	
1007	—	S	CH ₃	(6-)CF ₃	O	CH ₃	OCH ₂ CF ₃	
1008	—	S	CH ₃	(6-)CF ₃	O	CH ₃	OC ₃ H ₇	
1009	—	S	CH ₃	(6-)CF ₃	O			
1010	—	S	CH ₃	(6-)CF ₃	O		OCH ₃	
1011	—	S	CH ₃	(6-)CF ₃	O		OCH ₂ CF ₃	
1012	—	S	CH ₃	(6-)CH ₃	O	CH ₃	CH ₃	
1013	—	S	CH ₃	(6-)CH ₃	O	CH ₃	OCH ₃	
1014	—	S	CH ₃	(6-)CH ₃	O	CH ₃	OC ₂ H ₅	
1015	—	S	CH ₃	(6-)CH ₃	O	CH ₃	C ₂ H ₅	
1016	—	S	C ₂ H ₅	(6-)CH ₃	O	CH ₃	C ₂ H ₅	
1017	—	S	CH ₃	(6-)CH ₃	O	CH ₃	SCH ₃	
1018	—	S	C ₂ H ₅	(6-)CH ₃	O	CH ₃	SCH ₃	
1019	—	S	CH ₃	(6-)CH ₃	O		OC ₂ H ₅	
1020	—	S	C ₂ H ₅	(6-)CH ₃	O		OC ₂ H ₅	
1021	—	S	CH ₃	(6-)CH ₃	O	CH ₃	OC ₃ H ₇ -i	
1022	—	S	C ₂ H ₅	(6-)CH ₃	O		OC ₃ H ₇	
1023	—	O	CH ₃	(6-)CF ₃	O		OCH ₂ CF ₃	
1024	—	O	CH ₃	(6-)CF ₃	O		C ₃ H ₇ -i	
1025	—	O	CH ₃	(6-)CF ₃	O	CH ₃	C ₄ H ₉ -t	
1026	—	O	CH ₃	(6-)CF ₃	O		C ₃ H ₇	

TABLE 1A-continued

Examples of the compounds of the formula (IA)

(IA)

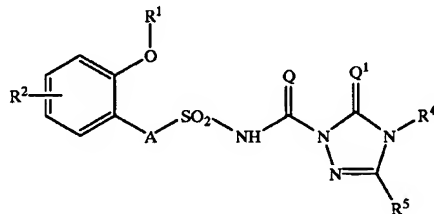


Ex. No.	A	Q	R ¹	(Position-) R ²	Q ¹	R ⁴	R ⁵	Melting point ° C.
1027	—	O	CH ₃	(6-)CF ₃	O		C ₃ H ₇ -i	
1028	—	O	CH ₃	(6-)CF ₃	O	C ₂ H ₅	C ₂ H ₅	
1029	—	O	CH ₃	(6-)CF ₃	O	C ₃ H ₇	C ₂ H ₅	
1030	—	O	CH ₃	(6-)CF ₃	O	C ₃ H ₇ -i	C ₂ H ₅	
1031	—	O	CH ₃	(6-)CF ₃	O	C ₂ H ₅	C ₃ H ₇	
1032	—	O	CH ₃	(6-)CF ₃	O	C ₃ H ₇	C ₃ H ₇	
1033	—	O	CH ₃	(6-)CF ₃	O	N(CH ₃) ₂	CH ₃	
1034	—	O	CH ₃	(6-)CF ₃	O	CH ₃	H	
1035	—	O	CH ₃	(6-)CF ₃	O	CH ₃	CH ₂ CH ₂ O	
1036	—	O	CH ₃	(6-)CF ₃	O	CH ₃		
1037	—	O	CH ₃	(6-)CF ₃	O	C ₃ H ₇ -i	C ₃ H ₇	
1038	—	O	CH ₃	(6-)CF ₃	O	C ₂ H ₅	C ₃ H ₇ -i	
1039	—	O	CH ₃	(6-)CF ₃	O	C ₃ H ₇	C ₃ H ₇ -i	
1040	—	O	CH ₃	(6-)CF ₃	O	C ₃ H ₇ -i	C ₃ H ₇ -i	
1041	—	O	CH ₃	(6-)CF ₃	O	C ₂ H ₅		
1042	—	O	CH ₃	(6-)CF ₃	O	C ₃ H ₇		
1043	—	O	CH ₃	(6-)CF ₃	O	C ₃ H ₇ -i		
1044	—	O	C ₂ H ₅	(6-)CF ₃	O	C ₃ H ₇ -i		
1045	—	O	C ₃ H ₇	(6-)CF ₃	O	C ₃ H ₇ -i		
1046	—	O	CH ₃	(6-)CF ₃	O	CH ₃	OCH ₂ CF ₂ CHF ₂	
1047	—	O	C ₃ H ₇	(6-)CF ₃	O	CH ₃	OCH ₂ CF ₂ CHF ₂	
1048	—	O	CH ₃	(6-)CF ₃	O	CH ₃	OC ₆ H ₁₁	
1049	—	O	C ₂ H ₅	(6-)CF ₃	O	CH ₃	OC ₆ H ₁₁	
1050	—	O	C ₃ H ₇	(6-)CF ₃	O	CH ₃	OC ₆ H ₁₁	
1051	—	O	CH ₃	(6-)CF ₃	O	CH ₃	O—CH—CH ₂ —CH ₃ CH ₃	
1052	—	O	CH ₃	(6-)CF ₃	O		C ₄ H ₉ -i	

TABLE 1A-continued

Examples of the compounds of the formula (IA)

(IA)

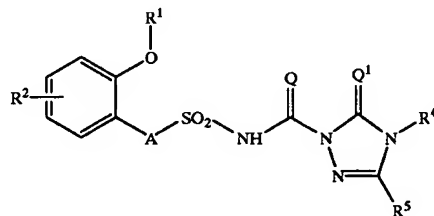


Ex. No.	A	Q	R ¹	(Position-) R ²	Q ¹	R ⁴	R ⁵	Melting point ° C.
1053	—	O	CH ₃	(6-)CF ₃	O		C ₄ H ₉	
1054	—	O	CH ₃	(6-)CF ₃	O		CH ₃	
1055	—	O	CH ₃	(6-)CF ₃	O		C ₂ H ₅	
1056	—	O	CH ₃	(6-)CF ₃	O			
1057	—	O	CH ₃	(6-)CF ₃	O	C ₂ H ₅	CH ₃	
1058	—	O	C ₃ H ₇ -i	(6-)CF ₃	O	C ₃ H ₇ -i		
1059	—	O	C ₃ H ₇ -i	(6-)CF ₃	O	CH ₃	OCH ₂ CF ₂ CHF ₂	
1060	—	O	C ₃ H ₇	(6-)CF ₃	O	CH ₃	CH ₃	
1061	—	O	C ₃ H ₇	(6-)CF ₃	O	CH ₃	C ₃ H ₇ -i	
1062	—	O	C ₃ H ₇	(6-)CF ₃	O	CH ₃	CH ₂ OCH ₃	
1063	—	O	C ₃ H ₇	(6-)CF ₃	O	CH ₃	Br	
1064	—	O	C ₃ H ₇	(6-)CF ₃	O	CH ₃	OCH ₂ CF ₃	
1065	—	O	C ₃ H ₇	(6-)CF ₃	O	OCH ₃	C ₃ H ₇	
1066	—	O	C ₃ H ₇	(6-)CF ₃	O		Br	
1067	—	O	C ₂ H ₅	(6-)CF ₃	O		CH ₂ OCH ₃	
1068	—	O	C ₃ H ₇	(6-)CF ₃	O		CH ₂ OCH ₃	
1069	—	S	C ₂ H ₅	(6-)CF ₃	O	CH ₃	SCH ₃	
1070	—	S	C ₃ H ₇	(6-)CF ₃	O	CH ₃	SCH ₃	
1071	—	S	C ₃ H ₇ -i	(6-)CF ₃	O	CH ₃	SCH ₃	
1072	—	S	C ₂ H ₅	(6-)CF ₃	O		OC ₂ H ₅	
1073	—	S	C ₃ H ₇	(6-)CF ₃	O		OC ₂ H ₅	
1074	—	S	C ₃ H ₇ -i	(6-)CF ₃	O		OC ₂ H ₅	
1075	—	S	C ₂ H ₅	(6-)CF ₃	O	CH ₃	OC ₃ H ₇ -i	
1076	—	S	C ₃ H ₇	(6-)CF ₃	O	CH ₃	OC ₃ H ₇ -i	

TABLE 1A-continued

Examples of the compounds of the formula (IA)

(IA)







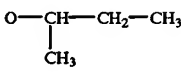



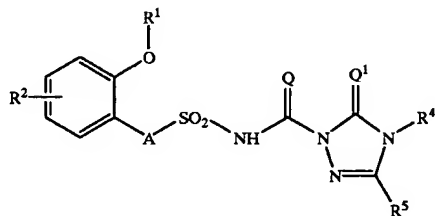
Ex. No.	A	Q	R ¹	(Position-) R ²	Q ¹	R ⁴	R ⁵	Melting point ° C.
1077	—	S	C ₃ H ₇ -i	(6-)CF ₃	O	CH ₃	OC ₃ H ₇ -i	
1078	—	S	C ₂ H ₅	(6-)CF ₃	O	C ₂ H ₅	OC ₂ H ₅	
1079	—	S	C ₃ H ₇	(6-)CF ₃	O	C ₂ H ₅	OC ₂ H ₅	
1080	—	S	C ₃ H ₇ -i	(6-)CF ₃	O	C ₂ H ₅	OC ₂ H ₅	
1081	—	S	C ₂ H ₅	(6-)CF ₃	O		OC ₃ H ₇	
1082	—	S	C ₃ H ₇	(6-)CF ₃	O		OC ₃ H ₇	
1083	—	S	C ₃ H ₇ -i	(6-)CF ₃	O		OC ₃ H ₇	
1084	—	O	C ₃ H ₇ -i	(6-)CF ₃	O		C ₄ H ₉ -t	
1085	—	O	C ₂ H ₅	(6-)CF ₃	O	CH ₃	OC ₄ H ₉ -s	
1086	—	O	C ₃ H ₇	(6-)CF ₃	O	CH ₃		
1087	—	O	C ₃ H ₇ -i	(6-)CF ₃	O	CH ₃	OC ₄ H ₉ -s	
1088	—	O	C ₂ H ₅	(6-)CF ₃	O	CH ₃	OC ₆ H ₅	
1089	—	O	C ₃ H ₇	(6-)CF ₃	O	CH ₃	OC ₆ H ₅	
1090	—	O	C ₃ H ₇ -i	(6-)CF ₃	O	CH ₃	OC ₆ H ₅	
1091	—	O	C ₂ H ₅	(6-)CF ₃	O	CH ₂ CH=CH ₂	OCH ₂ CF ₃	
1092	—	O	C ₃ H ₇	(6-)CF ₃	O	CH ₂ CH=CH ₂	OCH ₂ CF ₃	
1093	—	O	C ₃ H ₇ -i	(6-)CF ₃	O	CH ₂ CH=CH ₂	OCH ₂ CF ₃	
1094	—	O	C ₂ H ₅	(6-)CF ₃	O	CH ₃	OCH ₂ C ₆ H ₅	
1095	—	O	C ₃ H ₇	(6-)CF ₃	O	CH ₃	OCH ₂ C ₆ H ₅	
1096	—	O	C ₃ H ₇ -i	(6-)CF ₃	O	CH ₃	OCH ₂ C ₆ H ₅	
1097	—	O	C ₂ H ₅	(6-)CF ₃	O	N(CH ₃) ₂	CH ₃	
1098	—	O	C ₃ H ₇	(6-)CF ₃	O	N(CH ₃) ₂	CH ₃	
1099	—	O	C ₃ H ₇ -i	(6-)CF ₃	O	N(CH ₃) ₂	CH ₃	
1100	—	O	C ₃ H ₇	(6-)CF ₃	O	CH ₃	H	
1101	—	O	C ₂ H ₅	(6-)CF ₃	O	CH ₃	CH ₂ CH ₂ OC ₃ H ₇ -i	
1102	—	O	C ₃ H ₇ -i	(6-)CF ₃	O	CH ₃	CH ₂ CH ₂ OC ₃ H ₇ -i	
1103	—	O	C ₂ H ₅	(6-)CF ₃	O	CH ₃		
1104	—	O	C ₃ H ₇	(6-)CF ₃	O	CH ₃		
1105	—	O	C ₃ H ₇ -i	(6-)CF ₃	O	CH ₃		
1106	—	O	C ₂ H ₅	(6-)CF ₃	O	CH ₃	OCH ₂ CF ₂ CHF ₂	

TABLE 1A-continued

Examples of the compounds of the formula (IA)

(IA)

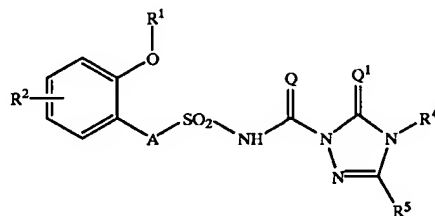


Ex. No.	A	Q	R ¹	(Position-) R ²	Q ¹	R ⁴	R ⁵	Melting point ° C.
1107	—	S	C ₃ H ₇ -i	(6-)CH ₃	O		OCH ₃	
1108	—	S	C ₃ H ₇	(6-)CH ₃	O		OC ₃ H ₇ -i	
1109	—	S	C ₃ H ₇ -i	(6-)CH ₃	O		OC ₃ H ₇ -i	
1110	—	S	C ₃ H ₇	(6-)CH ₃	O		OC ₃ H ₇	
1111	—	S	C ₃ H ₇ -i	(6-)CH ₃	O		OC ₃ H ₇	
1112	—	O	C ₂ H ₅	(6-)CF ₃	O		C ₄ H ₉ -i	
1113	—	O	C ₃ H ₇	(6-)CF ₃	O		C ₄ H ₉ -i	
1114	—	O	C ₃ H ₇ -i	(6-)CF ₃	O		C ₄ H ₉ -i	
1115	—	O	C ₂ H ₅	(6-)CF ₃	O		C ₄ H ₉	
1116	—	O	C ₃ H ₇	(6-)CF ₃	O		C ₄ H ₉	
1117	—	O	C ₃ H ₇ -i	(6-)CF ₃	O		C ₄ H ₉	
1118	—	O	C ₂ H ₅	(6-)CF ₃	O		CH ₃	
1119	—	O	C ₃ H ₇	(6-)CF ₃	O		CH ₃	
1120	—	O	C ₃ H ₇ -i	(6-)CF ₃	O		CH ₃	
1121	—	O	C ₂ H ₅	(6-)CF ₃	O		C ₂ H ₅	

TABLE 1A-continued

Examples of the compounds of the formula (IA)

(IA)

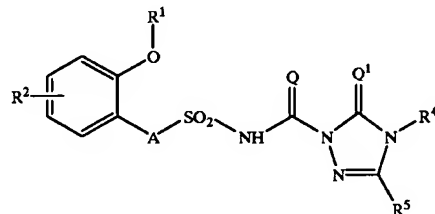


Ex. No.	A	Q	R ¹	(Position-) R ²	Q ¹	R ⁴	R ⁵	Melting point ° C.
1122	—	O	C ₃ H ₇	(6-)CF ₃	O		C ₂ H ₅	
1123	—	O	C ₃ H ₇ -i	(6-)CF ₃	O		C ₂ H ₅	
1124	—	O	C ₂ H ₅	(6-)CF ₃	O			
1125	—	O	C ₃ H ₇	(6-)CF ₃	O			
1126	—	O	C ₃ H ₇ -i	(6-)CF ₃	O			
1127	—	O	C ₂ H ₅	(6-)CF ₃	O	C ₃ H ₇		
1128	—	O	C ₃ H ₇	(6-)CF ₃	O	C ₃ H ₇		
1129	—	O	C ₃ H ₇ -i	(6-)CF ₃	O	C ₃ H ₇		
1130	—	O	C ₂ H ₅	(6-)CF ₃	O	C ₂ H ₅	CH ₃	
1131	—	O	C ₃ H ₇	(6-)CF ₃	O	C ₂ H ₅	CH ₃	
1132	—	O	C ₃ H ₇ -i	(6-)CF ₃	O	C ₂ H ₅	CH ₃	
1133	—	O	C ₂ H ₅	(6-)CF ₃	O		C ₃ H ₇	
1134	—	O	C ₃ H ₇	(6-)CF ₃	O		C ₃ H ₇	
1135	—	O	C ₃ H ₇ -i	(6-)CF ₃	O		C ₃ H ₇	
1136	—	O	C ₂ H ₅	(6-)CF ₃	O		C ₃ H ₇ -i	
1137	—	O	C ₃ H ₇	(6-)CF ₃	O		C ₃ H ₇ -i	

TABLE 1A-continued

Examples of the compounds of the formula (IA)

(IA)

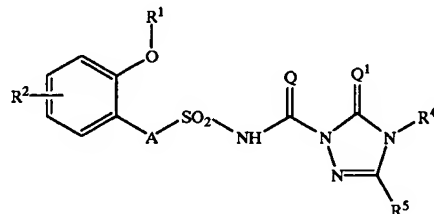


Ex. No.	A	Q	R ¹	(Position-) R ²	Q ¹	R ⁴	R ⁵	Melting point ° C.
1138	—	O	C ₃ H ₇ -i	(6-)CF ₃	O		C ₃ H ₇ -i	
1139	—	O	C ₂ H ₅	(6-)CF ₃	O	C ₂ H ₅	C ₂ H ₅	
1140	—	O	C ₃ H ₇	(6-)CF ₃	O	C ₂ H ₅	C ₂ H ₅	
1141	—	O	C ₃ H ₇ -i	(6-)CF ₃	O	C ₂ H ₅	C ₂ H ₅	
1142	—	O	C ₂ H ₅	(6-)CF ₃	O	C ₃ H ₇	C ₂ H ₅	
1143	—	O	C ₃ H ₇	(6-)CF ₃	O	C ₃ H ₇	C ₂ H ₅	
1144	—	O	C ₃ H ₇ -i	(6-)CF ₃	O	C ₃ H ₇	C ₂ H ₅	
1145	—	S	C ₂ H ₅	(6-)CF ₃	O		OC ₃ H ₇ -i	
1146	—	S	C ₃ H ₇	(6-)CF ₃	O		OC ₃ H ₇ -i	
1147	—	S	C ₃ H ₇ -i	(6-)CF ₃	O		OC ₃ H ₇ -i	
1148	—	S	C ₂ H ₅	(6-)CF ₃	O	CH ₃	OCH ₂ CF ₃	
1149	—	S	C ₃ H ₇	(6-)CF ₃	O	CH ₃	OCH ₂ CF ₃	
1150	—	S	C ₃ H ₇ -i	(6-)CF ₃	O	CH ₃	OCH ₂ CF ₃	
1151	—	S	C ₂ H ₅	(6-)CF ₃	O	CH ₃	OC ₃ H ₇	
1152	—	S	C ₃ H ₇	(6-)CF ₃	O	CH ₃	OC ₃ H ₇	
1153	—	S	C ₃ H ₇ -i	(6-)CF ₃	O	CH ₃	OC ₃ H ₇	
1154	—	S	C ₂ H ₅	(6-)CF ₃	O			
1155	—	S	C ₃ H ₇	(6-)CF ₃	O			
1156	—	S	C ₃ H ₇ -i	(6-)CF ₃	O			
1157	—	S	C ₂ H ₅	(6-)CF ₃	O		OCH ₃	
1158	—	S	C ₃ H ₇	(6-)CF ₃	O		OCH ₃	
1159	—	S	C ₃ H ₇ -i	(6-)CF ₃	O		OCH ₃	
1160	—	S	C ₂ H ₅	(6-)CF ₃	O		OCH ₂ CF ₃	

TABLE 1A-continued

Examples of the compounds of the formula (IA)

(IA)

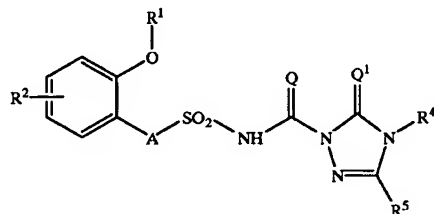


Ex. No.	A	Q	R ¹	(Position-) R ²	Q ¹	R ⁴	R ⁵	Melting point ° C.
1161	—	S	C ₃ H ₇	(6-)CF ₃	O		OCH ₂ CF ₃	
1162	—	S	C ₃ H ₇ -i	(6-)CF ₃	O		OCH ₂ CF ₃	
1163	—	O	C ₂ H ₅	(6-)CF ₃	O	C ₃ H ₇ -i	C ₃ H ₇	
1164	—	O	C ₃ H ₇	(6-)CF ₃	O	C ₃ H ₇ -i	C ₃ H ₇	
1165	—	O	C ₃ H ₇ -i	(6-)CF ₃	O	C ₃ H ₇ -i	C ₃ H ₇	
1166	—	O	C ₂ H ₅	(6-)CF ₃	O	C ₂ H ₅	C ₃ H ₇ -i	
1167	—	O	C ₃ H ₇	(6-)CF ₃	O	C ₂ H ₅	C ₃ H ₇ -i	
1168	—	O	C ₃ H ₇ -i	(6-)CF ₃	O	C ₂ H ₅	C ₃ H ₇ -i	
1169	—	O	C ₂ H ₅	(6-)CF ₃	O	C ₃ H ₇	C ₃ H ₇ -i	
1170	—	O	C ₃ H ₇	(6-)CF ₃	O	C ₃ H ₇	C ₃ H ₇ -i	
1171	—	O	C ₃ H ₇ -i	(6-)CF ₃	O	C ₃ H ₇	C ₃ H ₇ -i	
1172	—	O	C ₂ H ₅	(6-)CF ₃	O	C ₃ H ₇ -i	C ₃ H ₇ -i	
1173	—	O	C ₃ H ₇	(6-)CF ₃	O	C ₃ H ₇ -i	C ₃ H ₇ -i	
1174	—	O	C ₃ H ₇ -i	(6-)CF ₃	O	C ₃ H ₇ -i	C ₃ H ₇ -i	
1175	—	O	C ₂ H ₅	(6-)CF ₃	O	C ₃ H ₇ -i	C ₂ H ₅	
1176	—	O	C ₃ H ₇	(6-)CF ₃	O	C ₃ H ₇ -i	C ₂ H ₅	
1177	—	O	C ₃ H ₇ -i	(6-)CF ₃	O	C ₃ H ₇ -i	C ₂ H ₅	
1178	—	O	C ₂ H ₅	(6-)CF ₃	O	C ₂ H ₅	C ₃ H ₇	
1179	—	O	C ₃ H ₇	(6-)CF ₃	O	C ₂ H ₅	C ₃ H ₇	
1180	—	O	C ₃ H ₇ -i	(6-)CF ₃	O	C ₂ H ₅	C ₃ H ₇	
1181	—	O	C ₂ H ₅	(6-)CF ₃	O	C ₃ H ₇	C ₃ H ₇	
1182	—	O	C ₃ H ₇	(6-)CF ₃	O	C ₃ H ₇	C ₃ H ₇	
1183	—	O	C ₃ H ₇ -i	(6-)CF ₃	O	C ₃ H ₇	C ₃ H ₇	
1184	—	O	C ₂ H ₅	(6-)CF ₃	O	CH ₃	C ₄ H ₉ -t	
1185	—	O	C ₃ H ₇	(6-)CF ₃	O	CH ₃	C ₄ H ₉ -t	
1186	—	O	C ₃ H ₇ -i	(6-)CF ₃	O	CH ₃	C ₄ H ₉ -t	
1187	—	O	C ₂ H ₅	(6-)CF ₃	O		C ₄ H ₉ -t	
1188	—	O	C ₃ H ₇	(6-)CF ₃	O		C ₄ H ₉ -t	
1189	—	O	C ₂ H ₅	(6-)CF ₃	O	C ₂ H ₅		
1190	—	O	C ₃ H ₇	(6-)CF ₃	O	C ₂ H ₅		
1191	—	O	C ₃ H ₇ -i	(6-)CF ₃	O	C ₂ H ₅		
1192	—	O	C ₃ H ₇ -i	(6-)CF ₃	O	CH ₃	OC ₆ H ₁₁	
1193	—	S	C ₃ H ₇	(6-)CH ₃	O	CH ₃	OCH ₂ CF ₃	
1194	—	S	C ₃ H ₇ -i	(6-)CH ₃	O	CH ₃	OCH ₂ CF ₃	

TABLE 1A-continued

Examples of the compounds of the formula (IA)

(IA)

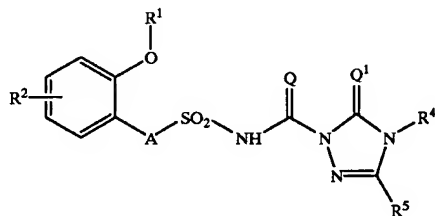


Ex. No.	A	Q	R ¹	(Position-) R ²	Q ¹	R ⁴	R ⁵	Melting point ° C.
1195	—	S	C ₃ H ₇	(6-)CH ₃	O		OCH ₂ CF ₃	
1196	—	S	C ₃ H ₇ -i	(6-)CH ₃	O		OCH ₂ CF ₃	
1197	—	S	C ₃ H ₇ -i	(6-)CH ₃	O	CH ₃	C ₂ H ₅	
1198	—	S	CH ₃	(6-)CF ₃	O	CH ₃	C ₂ H ₅	
1199	—	S	C ₃ H ₅	(6-)CF ₃	O	CH ₃	C ₂ H ₅	
1200	—	S	C ₃ H ₇	(6-)CF ₃	O	CH ₃	C ₂ H ₅	
1201	—	O	C ₂ H ₅	(6-)CF ₃	O		H	
1202	—	O	CH ₃	(6-)CF ₃	O		H	
1203	—	O	C ₃ H ₇	(6-)CF ₃	O		H	
1204	—	O	C ₃ H ₇ -i	(6-)CF ₃	O		H	
1205	—	O	C ₂ H ₅	(6-)CF ₃	O		OCH ₃	
1206	—	O	C ₃ H ₇	(6-)CF ₃	O		OCH ₃	
1207	—	O	C ₃ H ₇ -i	(6-)CF ₃	O		OCH ₃	
1208	—	O	CH ₃	(6-)CF ₃	O	CH ₂ CH=CH ₂	OC ₃ H ₇	
1209	—	O	C ₂ H ₅	(6-)CF ₃	O	CH ₂ CH=CH ₂	OC ₃ H ₇	
1210	—	O	CH ₃	(6-)CF ₃	O	CH ₃	OCH ₂ C ₄ H ₉ -t	
1211	—	O	C ₃ H ₅	(6-)CF ₃	O	CH ₃	OCH ₂ C ₄ H ₉ -t	
1212	—	O	C ₃ H ₇	(6-)CF ₃	O	CH ₃	OCH ₂ C ₄ H ₉ -t	
1213	—	O	C ₃ H ₇ -i	(6-)CF ₃	O	CH ₃	OCH ₂ C ₄ H ₉ -t	
1214	—	O	CH ₃	(6-)CF ₃	O	CH ₃	OCH ₂ CCl ₃	
1215	—	O	C ₂ H ₅	(6-)CF ₃	O	CH ₃	OCH ₂ CCl ₃	
1216	—	O	C ₃ H ₇	(6-)CF ₃	O	CH ₃	OCH ₂ CCl ₃	
1217	—	O	C ₃ H ₇ -i	(6-)CF ₃	O	CH ₃	OCH ₂ CCl ₃	
1218	—	O	CH ₃	(6-)CF ₃	O		CH=CHCH ₃	
1219	—	O	C ₂ H ₅	(6-)CF ₃	O		CH=CHCH ₃	

TABLE 1A-continued

Examples of the compounds of the formula (IA)

(IA)

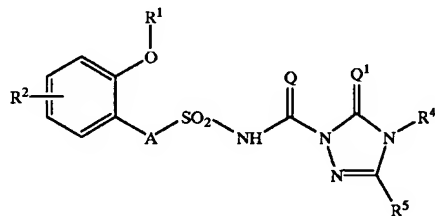


Ex. No.	A	Q	R ¹	(Position-) R ²	Q ¹	R ⁴	R ⁵	Melting point ° C.
1220	—	O	CH ₃	(4-)CH ₃	O	CH ₃	OCH ₃	
1221	—	O	CH ₃	(4-)CH ₃	O	CH ₃	OC ₂ H ₅	
1222	—	O	CH ₃	(4-)CH ₃	O		OC ₂ H ₅	
1223	—	O	CH ₃	(4-)CH ₃	O		OC ₃ H ₇	
1224	—	O	CH ₃	(4-)CH ₃	O	CH ₃	OC ₃ H ₇	
1225	—	O	CH ₃	(4-)CH ₃	O		OCH ₃	
1226	—	O	CH ₃	(6-)CF ₃	O	CH ₃	OCH ₂ C ₆ H ₅	
1227	—	O	CH ₃	(6-)CF ₃	O		CH=CHCH ₃	
1228	—	O	CH ₃	(6-)CF ₃	O	CH ₃	CH ₂ CH ₂ OCH ₃	
1229	—	O	C ₂ H ₅	(6-)CF ₃	O	CH ₃	CH ₂ CH ₂ OCH ₃	
1230	—	O	C ₂ H ₅	(6-)CF ₃	O		CH=CHCH ₃	
1231	—	O	C ₃ H ₇	(6-)CF ₃	O		CH=CHCH ₃	
1232	—	O	C ₃ H ₇ -i	(6-)CF ₃	O	CH ₃	CH ₂ CH ₂ OCH ₃	
1233	—	O	C ₃ H ₇ -i	(6-)CF ₃	O		CH=CHCH ₃	
1234	—	S	CH ₃	(6-)CH ₃	O		OC ₃ H ₇ -i	
1235	—	S	CH ₃	(6-)CH ₃	O	CH ₃	OCH ₂ CF ₃	
1236	—	S	CH ₃	(6-)CH ₃	O	CH ₃	OC ₃ H ₇	
1237	—	S	CH ₃	(6-)CH ₃	O		OCH ₃	
1238	—	S	C ₂ H ₅	(6-)CH ₃	O		OC ₃ H ₇ -i	
1239	—	S	C ₂ H ₅	(6-)CH ₃	O	CH ₃	OCH ₂ CF ₃	
1240	—	S	C ₂ H ₅	(6-)CH ₃	O	CH ₃	OC ₃ H ₇	

TABLE 1A-continued

Examples of the compounds of the formula (IA)

(IA)

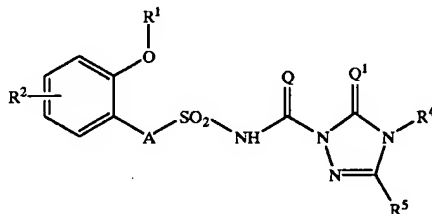


Ex. No.	A	Q	R ¹	(Position-) R ²	Q ¹	R ⁴	R ⁵	Melting point ° C.
1241	—	S	C ₂ H ₅	(6-)CH ₃	O		OCH ₃	
1242	—	S	C ₂ H ₅	(6-)CH ₃	O		OCH ₂ CF ₃	
1243	—	S	CH ₃	(6-)CH ₃	O			
1244	—	S	C ₂ H ₅	(6-)CH ₃	O			
1245	—	S	C ₃ H ₇	(6-)CH ₃	O			
1246	—	S	C ₃ H ₇ -i	(6-)CH ₃	O			
1247	—	O	CH ₃	(6-)CF ₃	O		OCH ₃	
1248	—	O	CH ₂ CH ₂ F	(6-)CF ₃	O	CH ₃	OCH ₃	
1249	—	O	CH ₂ CH ₂ F	(6-)CF ₃	O	CH ₃	OC ₂ H ₅	
1250	—	O	CH ₂ CH ₂ F	(6-)CF ₃	O		OC ₂ H ₅	
1251	—	O	CH ₂ CH ₂ F	(6-)CF ₃	O	CH ₃	OC ₃ H ₇	
1252	—	O	CH ₂ CH ₂ F	(6-)CF ₃	O	CH ₃	OC ₃ H ₇ -i	
1253	—	O	CH ₂ CH ₂ F	(6-)CF ₃	O		OC ₃ H ₇	
1254	—	O	CH ₂ CH ₂ F	(6-)CF ₃	O		OC ₃ H ₇ -i	
1255	—	O	CH ₂ CH ₂ F	(6-)CF ₃	O			
1256	—	O	CH ₃	(6-)C ₂ H ₅	O	CH ₃	CH ₂ OCH ₃	124
1257	—	O	CH ₃	(6-)C ₂ H ₅	O	CH ₃	OCH ₂ CF ₃	156
1258	—	O	CH ₃	(6-)C ₂ H ₅	O	CH ₃	SC≡CH	147
1259	—	O	CH ₃	(6-)C ₂ H ₅	O	CH ₃	CH(CH ₃) ₂	113
1260	—	O	CH ₃	(6-)C ₂ H ₅	O	CH ₃	CH(CH ₃) ₂	125
1261	—	O	CH ₃	(6-)C ₂ H ₅	O		Br	154

TABLE 1A-continued

Examples of the compounds of the formula (IA)

(IA)

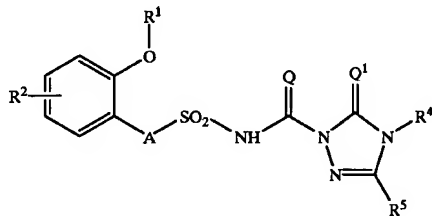


Ex. No.	A	Q	R ¹	(Position-) R ²	Q ¹	R ⁴	R ⁵	Melting point ° C.
1262	—	O	CH ₃	(6-)C ₂ H ₅	O	C ₂ H ₅	OC ₂ H ₅	>250
1263	—	O	CH ₃	(6-)C ₂ H ₅	O	CH ₃	OC ₃ H ₇	200
1264	—	O	CH ₃	(6-)C ₂ H ₅	O		OC ₂ H ₅	178
1265	—	O	CH ₃	(6-)Br	O	CH ₃	OC ₂ H ₅	228
1266	—	O	C ₂ H ₅	(6-)C ₂ H ₅	O	C ₂ H ₅	OC ₂ H ₅	
1267	—	O	C ₂ H ₅	(6-)C ₂ H ₅	O	CH ₃	OC ₃ H ₇	
1268	—	O	CH(CH ₃) ₂	(6-)C ₂ H ₅	O	C ₂ H ₅	OC ₂ H ₅	
1269	—	O	CH(CH ₃) ₂	(6-)C ₂ H ₅	O	CH ₃	OC ₃ H ₇	
1270	—	O	C ₂ H ₅	(6-)C ₂ H ₅	O	CH ₃	CH ₂ OCH ₃	
1271	—	O	C ₂ H ₅	(6-)C ₂ H ₅	O	CH ₃	OCH ₂ CF ₃	
1272	—	O	CH(CH ₃) ₂	(6-)C ₂ H ₅	O	CH ₃	CH ₂ OCH ₃	
1273	—	O	CH(CH ₃) ₂	(6-)C ₂ H ₅	O	CH ₃	OCH ₂ CF ₃	
1274	—	O	CF ₃	(6-)Br	O	CH ₃	OC ₂ H ₅	
1275	—	O	C ₂ H ₅	(6-)C ₂ H ₅	O	CH ₃	OCH(CH ₃) ₂	
1276	—	O	C ₂ H ₅	(6-)C ₂ H ₅	O		OC ₂ H ₅	
1277	—	O	C ₂ H ₅	(6-)C ₂ H ₅	O		CH ₂ OCH ₃	
1278	—	O	CH(CH ₃) ₂	(6-)C ₂ H ₅	O	CH ₃	OCH(CH ₃) ₂	
1279	—	O	CH(CH ₃) ₂	(6-)C ₂ H ₅	O	OC ₂ H ₅		
1280	—	O	CH(CH ₃) ₂	(6-)C ₂ H ₅	O		CH ₂ OCH ₃	
1281	—	O	C ₂ H ₇	(6-)Br	O	CH ₃	OC ₂ H ₅	
1282	—	O	C ₂ H ₅	(6-)C ₂ H ₅	O	CH ₃	OCH ₃	
1283	—	O	C ₂ H ₅	(6-)F	O	CH ₃	OC ₂ H ₅	165
1284	—	O	CH ₃	(6-)F	O	CH ₃	OC ₂ H ₅	211
1285	—	O	CH ₃	(6-)CN	O	CH ₃	OC ₂ H ₅	255
1286	—	O	CH ₃	(6-)CN	O	CH ₃	OC ₂ H ₅	
1287	—	O	CH ₃	(6-)CN	O	CH ₃	C ₂ H ₅	162
1288	—	O	CH ₃	(6-)CN	O	CH ₃	SCH ₃	268
1289	—	O	CH ₃	(6-)CN	O	CH ₃	OCH ₃	242
1290	—	O	CH ₃	(6-)CH(CH ₃) ₂	O	CH ₃	OCH ₃	183
1291	—	O	C ₂ H ₅	(6-)CH(CH ₃) ₂	O	CH ₃	CH ₃	283
1292	—	O	C ₂ H ₅	(6-)CH(CH ₃) ₂	O	CH ₃	OCH ₃	227
1293	—	O	CH ₃	(6-)F	O	CH ₃	OC ₂ H ₅	
1294	—	O	CH ₃	(6-)F	O	CH ₃	C ₂ H ₅	156
1295	—	O	CH ₃	(6-)F	O	CH ₃	SCH ₃	179
1296	—	O	CH ₃	(6-)F	O	CH ₃	CH ₃	217
1297	—	O	CH ₃	(6-)F	O		OCH(CH ₃) ₂	147
1298	—	O	CH(CH ₃) ₂	(6-)F	O	CH ₃	OC ₂ H ₅	165
1299	—	O	CH(CH ₃) ₂	(6-)F	O	CH ₃	C ₂ H ₅	113
1300	—	O	CH(CH ₃) ₂	(6-)F	O	CH ₃	OCH ₃	149

TABLE 1A-continued

Examples of the compounds of the formula (IA)

(IA)

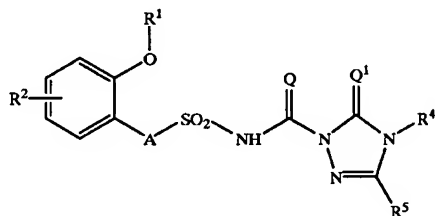


Ex. No.	A	Q	R ¹	(Position-) R ²	Q ¹	R ⁴	R ⁵	Melting point ° C.
1301	—	O	CH(CH ₃) ₂	(6-)F	O		CH ₃	147
1302	—	O	CH(CH ₃) ₂	(6-)F	O		OCH(CH ₃) ₂	127
1303	—	O	C ₂ H ₅	(6-)F	O	CH ₃	OC ₂ H ₅	139
1304	—	O	C ₂ H ₅	(6-)F	O	CH ₃	C ₂ H ₅	116
1305	—	O	C ₂ H ₅	(6-)F	O	CH ₃	OCH ₃	129
1306	—	O	C ₂ H ₅	(6-)F	O		CH ₃	120
1307	—	O	C ₂ H ₅	(6-)F	O	CH ₃	OC ₂ H ₅	>160
1308	—	O	C ₂ H ₅	(6-)CH ₃	O	CH ₃	OC ₄ H ₉	114
1309	—	O	C ₂ H ₅	(6-)CH ₃	O		OC ₄ H ₉	126
1310	—	O	C ₂ H ₅	(6-)CH ₃	O	CH ₃	OCH ₂ CH(CH ₃) ₂	144
1311	—	O	C ₂ H ₅	(6-)CH ₃	O		OCH ₂ CH(CH ₃) ₂	145
1312	—	O	C ₂ H ₅	(6-)CH ₃	O		OCH ₂ CF ₃	133
1313	—	O	C ₂ H ₅	(6-)CH ₃	O		OCH ₃	181
1314	—	O	C ₂ H ₅	(6-)CH(CH ₃) ₂	O	CH ₃	OC ₄ H ₉	170
1315	—	O	C ₂ H ₅	(6-)CH(CH ₃) ₂	O		OC ₄ H ₉	129
1316	—	O	CH ₃	(6-)CH ₃	O	CH ₃	OCH ₂ CH(CH ₃) ₂	137
1317	—	O	CH ₃	(6-)CH ₃	O		OCH ₂ CH(CH ₃) ₂	165
1318	—	O	CH ₃	(6-)CH ₃	O	CH ₃		150
1319	—	O	CH ₃	(6-)CH ₃	O			186
1320	—	O	C ₂ H ₅	(6-)CH(CH ₃) ₂	O	CH ₃	OCH ₂ CH(CH ₃) ₂	163

TABLE 1A-continued

Examples of the compounds of the formula (IA)

(IA)

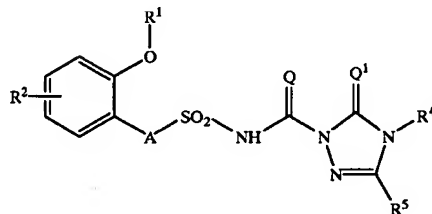


Ex. No.	A	Q	R^1	(Position-) R^2	Q^1	R^4	R^5	Melting point ° C.
1321	—	O	C_2H_5	$(6-)CH(CH_3)_2$	O	CH_3		134
1322	—	O	C_2H_5	$(6-)CH(CH_3)_2$	O			187
1323	—	O	C_2H_5	$(6-)CH(CH_3)_2$	O		CH_3	158
1324	—	O	C_2H_5	$(6-)CH(CH_3)_2$	O		C_2H_5	172
1325	—	O	C_2H_5	$(6-)CH(CH_3)_2$	O		C_3H_7	142
1326	—	O	C_2H_5	$(6-)CH(CH_3)_2$	O		OCH_2CF_3	150
1327	—	O	C_2H_5	$(6-)CH(CH_3)_2$	O		OCH_3	137
1328	—	O	C_2H_5	$(6-)CH(CH_3)_2$	O		$CH(CH_3)_2$	168
1329	—	O	C_2H_5	$(6-)CH_3$	O		SCH_3	167
1330	—	O	$CH(CH_3)_2$	$(6-)CH_3$	O		SCH_3	167
1331	—	O	C_2H_5	$(6-)CH_3$	O		SC_2H_5	150
1332	—	O	$CH(CH_3)_2$	$(6-)CH_3$	O		SC_2H_5	150
1333	—	O	C_3H_7	$(6-)CH_3$	O		SC_2H_5	140
1334	—	O	C_2H_5	$(6-)F$	O	CH_3	SCH_3	165
1335	—	O	C_2H_5	$(6-)F$	O	CH_3	C_2H_5	135
1336	—	O	C_2H_5	$(6-)F$	O	CH_3	OCH_3	168
1337	—	O	C_2H_5	$(6-)F$	O	CH_3	CH_3	140

TABLE 1A-continued

Examples of the compounds of the formula (IA)

(IA)

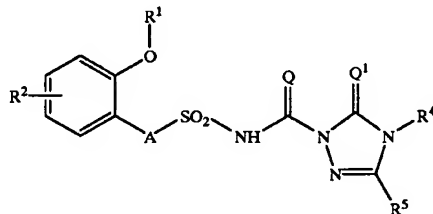


Ex. No.	A	Q	R ¹	(Position-) R ²	Q ¹	R ⁴	R ⁵	Melting point ° C.
1338	—	O	C ₂ H ₅	(6-)F	O		OCH ₃	138
1339	—	O	C ₃ H ₇	(6-)OC ₃ H ₇	O	CH ₃	OC ₂ H ₅	
1340	—	O	C ₃ H ₇	(6-)OC ₃ H ₇	O	CH ₃	OC ₂ H ₅	
1341	—	O	C ₃ H ₇	(6-)OCH ₃	O	CH ₃	OC ₂ H ₅	
1342	—	O	CH(CH ₃) ₂	(6-)OCH(CH ₃) ₂	O	CH ₃	OC ₂ H ₅	
1343	—	O	C ₂ H ₅	(6-)OC ₂ H ₅	O	CH ₃	OC ₂ H ₅	
1344	—	O	C ₄ H ₉	(6-)OC ₄ H ₉	O	CH ₃	OC ₂ H ₅	
1345	—	O	CH(CH ₃)C ₂ H ₅	(6-)OCH ₃	O	CH ₃	CH ₃	157
1346	—	O	CH(CH ₃)C ₂ H ₅	(6-)OCH ₃	O	CH ₃	C ₃ H ₇	86
1347	—	O	CH ₂ CH(CH ₃) ₂	(6-)OC ₂ H ₅	O	CH ₃	Br	108
1348	—	O	CH ₂ CH(CH ₃) ₂	(6-)OC ₂ H ₅	O	CH ₃	CH ₃	82
1349	—	O	CH ₂ CH(CH ₃) ₂	(6-)OC ₂ H ₅	O		Br	120
1350	—	O	C ₃ H ₇	(6-)OCH ₃	O		OCH(CH ₃) ₂	141
1351	—	O	C ₃ H ₇	(6-)OCH ₃	O		OC ₃ H ₇	102
1352	—	O	C ₂ H ₅	(6-)OC ₂ H ₅	O		OCH(CH ₃) ₂	110
1353	—	O	C ₂ H ₅	(6-)OC ₂ H ₅	O		OC ₃ H ₇	132
1354	—	O	C ₂ H ₅	(6-)OC ₂ H ₅	O		OCH ₂ CF ₃	114
1355	—	O	CH(CH ₃) ₂	(6-)OCH ₃	O		OCH(CH ₃) ₂	172
1356	—	O	CH(CH ₃) ₂	(6-)OCH ₃	O		OC ₃ H ₇	156
1357	—	O	C ₃ H ₇	(6-)OC ₃ H ₇	O		OCH(CH ₃) ₂	141
1358	—	O	C ₂ H ₅	(6-)OCH ₃	O		OC ₃ H ₇	134
1359	—	O	C ₂ H ₅	(6-)OC ₃ H ₇	O		OCH(CH ₃) ₂	132

TABLE 1A-continued

Examples of the compounds of the formula (IA)

(IA)

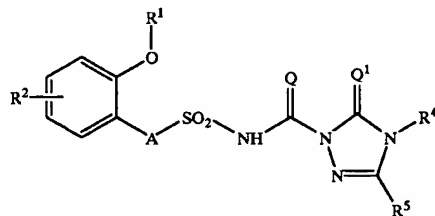


Ex. No.	A	Q	R ¹	(Position-) R ²	Q ¹	R ⁴	R ⁵	Melting point ° C.
1360	—	O	CH(CH ₃) ₂	(6-)OC ₂ H ₅	O		OCH(CH ₃) ₂	147
1361	—	O	(CH ₂) ₂ OC ₃ H ₇	(6-)OCH ₃	O		OCH ₂ CF ₃	134
1362	—	O	CH ₂ CH(CH ₃) ₂	(6-)OC ₂ H ₅	O		OCH(CH ₃) ₂	96
1363	—	O	C ₃ H ₇	(6-)OC ₂ H ₅	O		OCH ₃	120
1364	—	O	CH(CH ₃) ₂	(6-)OC ₂ H ₅	O		OCH ₃	108
1365	—	O	CH(CH ₃) ₂	(6-)OC ₂ H ₅	O		OC ₃ H ₇	127
1366	—	O	C ₄ H ₉	(6-)OC ₄ H ₉	O		OCH ₃	118
1367	—	O	CH(CH ₃) ₂	(6-)OCH(CH ₃) ₂	O		154	
1368	—	O	C ₃ H ₇	(6-)OCH ₃	O		CH ₃	122
1369	—	O	CH(CH ₃) ₂	(6-)OCH(CH ₃) ₂	O		CH ₃	134
1370	—	O	CH(CH ₃) ₂	(6-)OCH(CH ₃) ₂	O	CH ₃	CH(CH ₃)C ₂ H ₅	97
1371	—	O	C ₃ H ₇	(6-)OCH ₃	O	CH ₃	CH(CH ₃) ₂	76
1372	—	O	C ₃ H ₇	(6-)OCH ₃	O	CH ₃	SC ₂ H ₅	91
1373	—	O	C ₂ H ₅	(6-)OC ₂ H ₅	O			102
1374	—	O	C ₂ H ₅	(6-)OC ₂ H ₅	O		CH ₃	125
1375	—	O	C ₂ H ₅	(6-)OC ₂ H ₅	O		C ₂ H ₅	112
1376	—	O	C ₂ H ₅	(6-)OC ₂ H ₅	O		C ₃ H ₇	113

TABLE 1A-continued

Examples of the compounds of the formula (IA)

(IA)

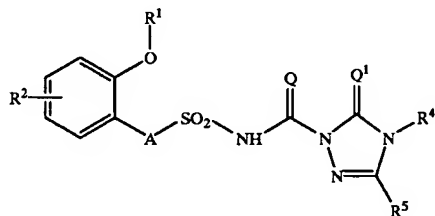


Ex. No.	A	Q	R ¹	(Position-) R ²	Q ¹	R ⁴	R ⁵	Melting point ° C.
1377	—	O	C ₂ H ₅	(6-)OC ₂ H ₅	O	CH ₃	CH(CH ₃)C ₂ H ₅	100
1378	—	O	C ₂ H ₅	(6-)OC ₂ H ₅	O	CH ₃	CH(CH ₃) ₂	118
1379	—	O	C ₂ H ₅	(6-)OC ₂ H ₅	O	CH ₃	C ₄ H ₉	81
1380	—	O	C ₂ H ₅	(6-)OC ₂ H ₅	O	CH ₃	CH ₂ CH(CH ₃) ₂	93
1381	—	O	C ₃ H ₇	(6-)OCH ₃	O	CH ₃	CH(CH ₃)C ₂ H ₅	79
1382	—	O	C ₃ H ₇	(6-)OCH ₃	O	CH ₃	CH ₂ CH(CH ₃) ₂	83
1383	—	O	C ₂ H ₅	(6-)OC ₂ H ₅	O		OC ₂ H ₅	67
1384	—	O	C ₂ H ₅	(6-)OC ₂ H ₅	O		C ₃ H ₇	105
1385	—	O	C ₂ H ₅	(6-)OC ₂ H ₅	O		OCH(CH ₃) ₂	97
1386	—	O	C ₃ H ₇	(6-)OCH ₃	O		OCH(CH ₃) ₂	82
1387	—	O	C ₂ H ₅	(6-)OC ₂ H ₅	O		CH ₂ CH(CH ₃) ₂	103
1388	—	O	C ₃ H ₇	(6-)OCH ₃	O		CH ₂ CH(CH ₃) ₂	116
1389	—	O	C ₂ H ₅	(6-)OC ₂ H ₅	O		OCH ₃	104
1390	—	O	CH(CH ₃) ₂	(6-)OCH(CH ₃) ₂	O		OCH ₃	127
1391	—	O	C ₃ H ₇	(6-)OCH ₃	O		OCH ₃	93
1392	—	O	C ₂ H ₅	(6-)OC ₂ H ₅	O		OC ₃ H ₇	107
1393	—	O	CH(CH ₃) ₂	(6-)OCH(CH ₃) ₂	O		OC ₃ H ₇	130
1394	—	O	C ₃ H ₇	(6-)OCH ₃	O		OC ₃ H ₇	80
1395	—	O	C ₂ H ₅	(6-)OC ₂ H ₅	O		C ₄ H ₉	77

TABLE 1A-continued

Examples of the compounds of the formula (IA)

(IA)

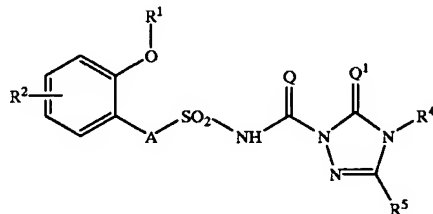


Ex. No.	A	Q	R ¹	(Position-) R ²	Q ¹	R ⁴	R ⁵	Melting point ° C.
1396	—	O	C ₃ H ₇	(6-)CH ₃	O		C ₄ H ₉	107
1397	—	O	C ₂ H ₅	(6-)OC ₂ H ₅	O			107
1398	—	O	C ₃ H ₇	(6-)OCH ₃	O			85
1399	—	O	CH(CH ₃) ₂	(6-)OCH(CH ₃) ₂	O		C ₄ H ₉	89
1400	—	O	CH(CH ₃) ₂	(6-)OCH(CH ₃) ₂	O			100
1401	—	O	CH(CH ₃) ₂	(6-)OCH(CH ₃) ₂	O	CH ₃	CH(CH ₃) ₂	118
1402	—	O	CH(CH ₃) ₂	(6-)OCH(CH ₃) ₂	O	CH ₃	C ₄ H ₉	82
1403	—	O	C ₃ H ₇	(6-)OCH ₃	O	CH ₃	C ₄ H ₉	75
1404	—	O	CH(CH ₃) ₂	(6-)OCH(CH ₃) ₂	O		CH ₂ CH(CH ₃) ₂	95
1405	—	O	C ₂ H ₅	(6-)OC ₂ H ₅	O	CH ₃	C(CH ₃)=CH ₂	118
1406	—	O	CH(CH ₃) ₂	(6-)OCH(CH ₃) ₂	O	CH ₃	C(CH ₃)=CH ₂	103
1407	—	O	C ₃ H ₇	(6-)OCH ₃	O	CH ₃	C(CH ₃)=CH ₂	92
1408	—	O	C ₃ H ₇	(6-)OH	O	CH ₃	OC ₂ H ₅	152
1409	—	O	C ₃ H ₇	(6-)OH	O	CH ₃	OCH ₃	137
1410	—	O	C ₃ H ₇	(6-)OH	O	CH ₃	OC ₃ H ₇	88
1411	—	O	C ₃ H ₇	(6-)OH	O	CH ₃	OCH(CH ₃) ₂	125
1412	—	O	CH ₃	(6-)OH	O	CH ₃	OCH ₃	148
1413	—	O	CH ₃	(6-)OH	O	CH ₃	OC ₂ H ₅	144
1414	—	O	CH ₃	(6-)OH	O	CH ₃	OC ₃ H ₇	128
1415	—	O	CH ₃	(6-)OH	O	CH ₃	OCH(CH ₃) ₂	158
1416	—	O	CH ₃	(6-)OH	O		OCH(CH ₃) ₂	137
1417	—	O	CH ₃	(6-)OH	O		OC ₂ H ₅	125
1418	—	O	CH ₃	(6-)OH	O	C ₂ H ₅	OCH ₃	137
1419	—	O	CH ₃	(6-)OH	O	CH ₃	SCH ₃	130
1420	—	O	CH ₃	(6-)OH	O	CH ₃	C ₂ H ₅	40
1421	—	O	C ₃ H ₇	(6-)OCHF ₂	O	CH ₃	OC ₂ H ₅	230
1422	—	O	CH ₃	(6-)OCHF ₂	O	CH ₃	OC ₂ H ₅	202
1423	—	O	CH ₃	(6-)OCHF ₂	O	CH ₃	SCH ₃	202
1424	—	O	CH ₃	(6-)OCHF ₂	O	CH ₃	C ₂ H ₅	98
1425	—	S	C ₃ H ₇	(6-)OCHF ₂	O	CH ₃	OC ₃ H ₇	198

TABLE 1A-continued

Examples of the compounds of the formula (IA)

(IA)

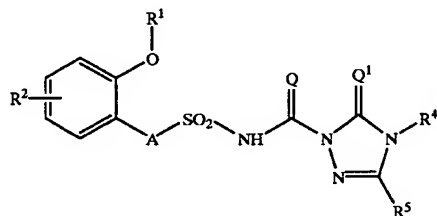


Ex. No.	A	Q	R ¹	(Position-) R ²	Q ¹	R ⁴	R ⁵	Melting point ° C.
1426	—	O	C ₃ H ₇	(6-)OCHF ₂	O		OC ₂ H ₅	124
1427	—	O	CHF ₂	(6-)OCRF ₂	O	CH ₃	OCH ₃	145
1428	—	O	CHF ₂	(6-)OCHF ₂	O	CH ₃	OC ₂ H ₅	177
1429	—	O	CHF ₂	(6-)OCHF ₂	O	CH ₃	OC ₂ H ₅	>220
1430	—	O	CHF ₂	(6-)OCHF ₂	O		OC ₂ H ₅	148
1431	—	O	CHF ₂	(6-)OCHF ₂	O	CH ₃	C ₂ H ₅	123
1432	—	O	CHF ₂	(6-)OCHF ₂	O	CH ₃	SCH ₃	133
1433	—	O	CHF ₂	(6-)OCHF ₂	O		CH(CH ₃) ₂	130
1434	—	O	C ₃ H ₇	(6-)OCHF ₂	O	CH ₃	OCH(CH ₃) ₂	
1435	—	O	CH ₂ CH=CH ₂	(6-)OC ₂ H ₅	O		OC ₂ H ₅	84
1436	—	O	CH ₂ CH=CH ₂	(6-)OC ₂ H ₅	O		OCH ₃	120
1437	—	O	CH ₂ CH=CH ₂	(6-)OC ₂ H ₅	O	CH ₃	OCH(CH ₃) ₂	133
1438	—	O	CH ₂ CH=CH ₂	(6-)OC ₂ H ₅	O	CH ₃	C ₂ H ₅	137
1439	—	O	CH ₂ CH=CH ₂	(6-)OC ₂ H ₅	O	CH ₃	OC ₂ H ₅	100
1440	—	O	CH ₂ CH=CH ₂	(6-)OC ₂ H ₅	O	CH ₃	OCH ₃	94
1441	—	O	CH ₃	(6-)OCH ₃	O	CH ₃	N(CH ₃) ₂	168
1442	—	O	CH ₃	(6-)C ₂ H ₅	O	CH ₃	N(CH ₃) ₂	172
1443	—	O	CF ₃	(6-)C ₂ H ₅	O	CH ₃	N(CH ₃) ₂	128
1444	—	O	CF ₃	(6-)CH ₃	O	CH ₃	OC ₂ H ₅	224
1445	—	O	CH ₃	(6-)OCH ₃	O	OCH ₃	SCH ₃	157
1446	—	O	CH ₃	(6-)OCH ₃	O	OCH ₃	SC ₂ H ₅	149
1447	—	O	CH ₃	(6-)OCH ₃	O	OC ₂ H ₅	SCH ₃	147
1448	—	O	CH ₃	(6-)OCH ₃	O	OC ₂ H ₅	SC ₂ H ₅	124
1449	—	O	CH ₃	(6-)C ₂ H ₅	O	NH ₂	CH ₃	204
1450	—	O	C ₃ H ₅	(6-)OC ₂ H ₅	O	OCH ₃	SCH ₃	130
1451	—	O	C ₃ H ₅	(6-)OC ₂ H ₅	O	OCH ₃	SC ₂ H ₅	112
1452	—	O	C ₃ H ₅	(6-)OC ₂ H ₅	O	OC ₂ H ₅	SCH ₃	113
1453	—	O	OC ₂ H ₅	(6-)OC ₂ H ₅	O	OC ₂ H ₅	SC ₂ H ₅	157
1454	—	O	CF ₂ CF ₂ H	(6-)CH ₃	O	CH ₃	OCH ₃	127
1455	—	O	CF ₂ CF ₂ H	(6-)CH ₃	O	CH ₃	OC ₂ H ₅	156
1456	—	O	CF ₂ CF ₂ H	(6-)CH ₃	O	CH ₃	OC ₃ H ₇	132
1457	—	O	CF ₂ CF ₂ H	(6-)CH ₃	O	CH ₃	OCH(CH ₃) ₂	172
1458	—	O	CF ₂ CF ₂ H	(6-)CH ₃	O	CH ₃	C ₃ H ₇	90
1459	—	O	CF ₃	(6-)C ₂ H ₅	O	CH ₃	OC ₃ H ₇	185
1460	—	O	CF ₃	(6-)C ₂ H ₅	O	CH ₃	OCH(CH ₃) ₂	192
1461	—	O	CF ₃	(6-)C ₂ H ₅	O	CH ₃	OCH ₂ CF ₃	159
1462	—	O	CF ₃	(6-)C ₂ H ₅	O	CH ₃	O—CH ₂ —	192

TABLE 1A-continued

Examples of the compounds of the formula (IA)

(IA)

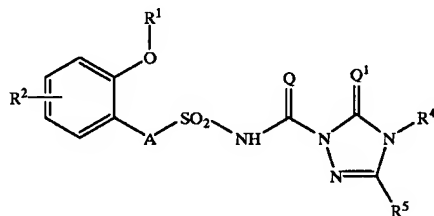


Ex. No.	A	Q	R^1	(Position-) R^2	Q^1	R^4	R^5	Melting point ° C.
1463	—	O	CF_3	$(6-)C_2H_5$	O	CH_3	CH_3	>270
1464	—	O	CF_3	$(6-)C_2H_5$	O	CH_3	C_2H_5	259
1465	—	O	CF_3	$(6-)C_2H_5$	O	CH_3	C_3H_7	241
1466	—	O	CF_3	$(6-)C_2H_5$	O	CH_3	SCH_3	174
1467	—	O	CF_3	$(6-)C_2H_5$	O		OCH_3	165
1468	—	O	CF_3	$(6-)C_2H_5$	O		OC_2H_5	145
1469	—	O	CF_3	$(6-)C_2H_5$	O		OC_3H_7	196
1470	—	O	CF_3	$(6-)C_2H_5$	O		$OCH(CH_3)_2$	170
1471	—	O	CF_3	$(6-)C_2H_5$	O		OCH_2CF_3	183
1472	—	O	CF_3	$(6-)C_2H_5$	O		$O-CH_2-$	154
1473	—	O	CF_3	$(6-)C_2H_5$	O	CH_3		188
1474	—	O	CF_3	$(6-)C_2H_5$	O		C_2H_5	163
1475	—	O	CF_3	$(6-)C_2H_5$	O		C_3H_7	190
1476	—	O	CF_3	$(6-)C_2H_5$	O		SCH_3	159
1477	—	O	CF_3	$(6-)C_2H_5$	O		CH_3	151
1478	—	O	CF_3	$(6-)C_2H_5$	O		C_3H_7	163
1479	—	O	CF_3	$(6-)C_2H_5$	O		$CH=CH(CH_3)$	176

TABLE 1A-continued

Examples of the compounds of the formula (IA)

(IA)

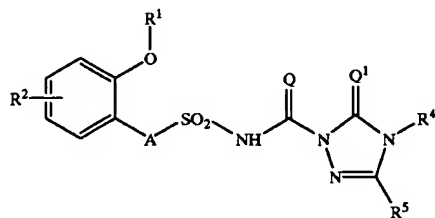


Ex. No.	A	Q	R ¹	(Position-) R ²	Q ¹	R ⁴	R ⁵	Melting point ° C.
1480	—	O	CF ₃	(6-)C ₂ H ₅	O			156
1481	—	O	CF ₃	(6-)C ₂ H ₅	O		CH ₂ OCH ₃	146
1482	—	O	CH ₃	(6-)C ₂ H ₅	O	OCH ₃	SCH ₃	149
1483	—	O	CH ₃	(6-)C ₂ H ₅	O	OCH ₃	SC ₂ H ₅	137
1484	—	O	CH ₃	(6-)C ₂ H ₅	O	OC ₂ H ₅	SCH ₃	115
1485	—	O	CH ₃	(6-)C ₂ H ₅	O	OC ₂ H ₅	SC ₂ H ₅	99
1486	—	O	CH(CH ₃) ₂	(6-)CH ₃	O		CH(CH ₃) ₂	132
1487	—	O	CF ₃	(6-)C ₂ H ₅	O		CH(CH ₃) ₂	108
1488	—	O	CH ₃	(6-)C ₂ H ₅	O		CH(CH ₃) ₂	151
1489	—	O	CH ₃	(6-)OCH ₃	O		CH(CH ₃) ₂	163
1490	—	O	C ₂ H ₅	(6-)OC ₂ H ₅	O		CH(CH ₃) ₂	111
1491	—	O	CH ₃	(6-)C ₂ H ₅	O	CH ₃		118
1492	—	O	CH ₃	(6-)C ₂ H ₅	O			147
1493	—	O	CF ₃ CF ₃ H	(6-)CH ₃	O		OCH ₃	160
1494	—	O	CF ₃ CF ₃ H	(6-)CH ₃	O		OC ₂ H ₅	155
1495	—	O	CF ₃ CF ₃ H	(6-)CH ₃	O		OCH(CH ₃) ₂	166
1496	—	O	CH ₂ CF ₃	(6-)CH ₃	O	CH ₃	OCH ₃	149
1497	—	O	CH ₂ CF ₃	(6-)CH ₃	O	CH ₃	OC ₂ H ₅	230
1498	—	O	CH ₂ CF ₃	(6-)CH ₃	O	CH ₃	OC ₃ H ₇	238
1499	—	O	CH ₂ CF ₃	(6-)CH ₃	O	CH ₃	OCH(CH ₃) ₂	89

TABLE 1A-continued

Examples of the compounds of the formula (IA)

(IA)

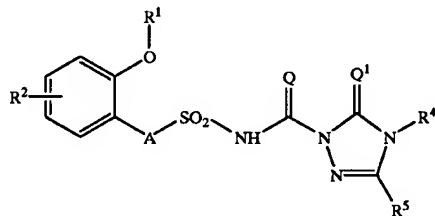


Ex. No.	A	Q	R^1	(Position-) R^2	Q^1	R^4	R^5	Melting point ° C.
1500	—	O	CH_2CF_3	(6-) CH_3	O	CH_3	$O-CH_2-$	134
1501	—	O	CH_2CF_3	(6-) CH_3	O		OCH_3	189
1502	—	O	CH_2CF_3	(6-) CH_3	O		OC_2H_5	129
1503	—	O	CH_2CF_3	(6-) CH_3	O		OC_3H_7	126
1504	—	O	CH_2CF_3	(6-) CH_3	O		$OCH(CH_3)_2$	161
1505	—	O	CH_2CF_3	(6-) CH_3	O		$O-CH_2-$	128
1506	—	O	CH_2CF_3	(6-) CH_3	O	CH_3	C_3H_7	148
1507	—	O	CH_3	(6-) C_2H_5	O	CH_3		144
1508	—	O	CH_3	(6-) C_2H_5	O	CH_3	$CH=CH(CH_3)$	138
1509	—	O	CH_3	(6-) C_2H_5	O	CH_3	$O-CH_2-$	160
1510	—	O	CH_3	(6-) C_2H_5	O		CH_3	168
1511	—	O	CH_3	(6-) C_2H_5	O		C_2H_5	143
1512	—	O	CH_3	(6-) C_2H_5	O		C_3H_7	140
1513	—	O	CH_3	(6-) C_2H_5	O			170
1514	—	O	CH_3	(6-) C_2H_5	O		$CH=CH(CH_3)$	127
1515	—	O	CH_3	(6-) C_2H_5	O		$N(CH_3)_2$	214

TABLE 1A-continued

Examples of the compounds of the formula (IA)

(IA)

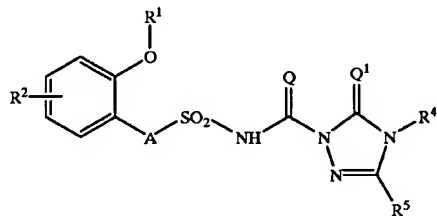


Ex. No.	A	Q	R ¹	(Position-) R ²	Q ¹	R ⁴	R ⁵	Melting point ° C.
1516	—	O	CH ₃	(6-)C ₂ H ₅	O		SCH ₃	167
1517	—	O	CH ₃	(6-)C ₂ H ₅	O		SC ₂ H ₅	137
1518	—	O	CH ₃	(6-)C ₂ H ₅	O		OCH ₃	165
1519	—	O	CH ₃	(6-)C ₂ H ₅	O		OCH ₂ CF ₃	163
1520	—	O	CH ₃	(6-)C ₂ H ₅	O		OC ₃ H ₇	125
1521	—	O	CH ₃	(6-)C ₂ H ₅	O		O—CH ₂ —	133
1522	—	O	CH ₃	(6-)C ₂ H ₅	O	CH ₃	S—	136
1523	—	O	CF ₂ H	(6-)C ₂ H ₅	O	CH ₃	OCH ₃	121
1524	—	O	CF ₂ H	(6-)C ₂ H ₅	O	CH ₃	OC ₂ H ₅	150
1525	—	O	CF ₂ H	(6-)C ₂ H ₅	O	CH ₃	OC ₃ H ₇	119
1526	—	O	CF ₂ H	(6-)C ₂ H ₅	O	CH ₃	OCH(CH ₃) ₂	122
1527	—	O	CF ₂ H	(6-)C ₂ H ₅	O	CH ₃	O—CH ₂ —	146
1528	—	O	CF ₂ H	(6-)C ₂ H ₅	O		OCH ₃	163
1529	—	O	CF ₂ H	(6-)C ₂ H ₅	O		OC ₂ H ₅	140
1530	—	O	CF ₂ H	(6-)C ₂ H ₅	O		OC ₃ H ₇	129
1531	—	O	CF ₂ H	(6-)C ₂ H ₅	O		OCH(CH ₃) ₂	118
1532	—	O	CF ₂ H	(6-)C ₂ H ₅	O		O—CH ₂ —	133
1533	—	O	CH ₂ CH ₂ F	(6-)C ₂ H ₅	O	CH ₃	CH ₃	290

TABLE 1A-continued

Examples of the compounds of the formula (IA)

(IA)

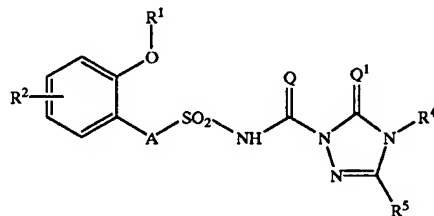


Ex. No.	A	Q	R^1	(Position-) R^2	Q^1	R^4	R^5	Melting point ° C.
1534	—	O	CH_2CH_2F	(6-) C_2H_5	O	CH_3	C_3H_7	274
1535	—	O	CH_2CH_2F	(6-) C_2H_5	O	OC_2H_5	C_2H_5	118
1536	—	O	CH_2CH_2F	(6-) C_2H_5	O	CH_3	$N(CH_3)_2$	170
1537	—	O	CH_2CH_2F	(6-) C_2H_5	O	CH_3	SCH_3	168
1538	—	O	CH_2CH_2F	(6-) C_2H_5	O	CH_3	OCH_3	166
1539	—	O	CH_2CH_2F	(6-) C_2H_5	O	CH_3	OC_2H_5	140
1540	—	O	CH_2CH_2F	(6-) C_2H_5	O	CH_3	OC_3H_7	138
1541	—	O	CH_2CH_2F	(6-) C_2H_5	O	CH_3	$OCH(CH_3)_2$	136
1542	—	O	CH_2CH_2F	(6-) C_2H_5	O	CH_3	$O-CH_2-$	139
1543	—	O	CH_2CH_2F	(6-) C_2H_5	O		OCH_3	156
1544	—	O	CH_2CH_2F	(6-) C_2H_5	O		OC_2H_5	138
1545	—	O	CH_2CH_2F	(6-) C_2H_5	O		OC_3H_7	125
1546	—	O	CH_2CH_2F	(6-) C_2H_5	O		$OCH(CH_3)_2$	140
1547	—	O	CH_2CH_2F	(6-) C_2H_5	O		$O-CH_2-$	158
1548	—	O	CH_3	(6-) C_2H_5	O		C_5H_9	149
1549	—	O	CH_3	(6-) C_2H_5	O	CH_3	C_5H_9	118
1550	—	O	CF_3	(6-) C_3H_7	O	CH_3	OCH_3	94
1551	—	O	CF_3	(6-) C_3H_7	O	CH_3	OC_2H_5	175
1552	—	O	CF_3	(6-) C_3H_7	O	CH_3	OC_3H_7	166
1553	—	O	CF_3	(6-) C_3H_7	O	CH_3	$OCH(CH_3)_2$	164
1554	—	O	CF_3	(6-) C_3H_7	O	CH_3	OCH_2CF_3	221
1555	—	O	CF_3	(6-) C_3H_7	O	CH_3	$O-CH_2-$	189
1556	—	O	CF_3	(6-) C_3H_7	O		OCH_3	182
1557	—	O	CF_3	(6-) C_3H_7	O		OCH_2CH_3	134
1558	—	O	CF_3	(6-) C_3H_7	O		OC_3H_7	207

TABLE 1A-continued

Examples of the compounds of the formula (IA)

(IA)

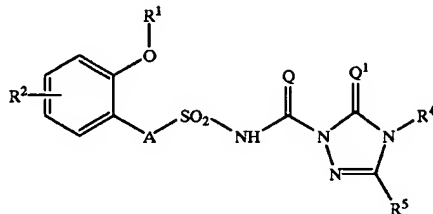


Ex. No.	A	Q	R ¹	(Position-) R ²	Q ¹	R ⁴	R ⁵	Melting point ° C.
1559	—	O	CF ₃	(6-)C ₃ H ₇	O		OCH(CH ₃) ₂	148
1560	—	O	CF ₃	(6-)C ₃ H ₇	O		OCH ₂ CF ₃	204
1561	—	O	CF ₃	(6-)C ₃ H ₇	O		O—CH ₂ —	186
1562	—	O	CH ₃	(6-)C ₂ H ₅	O		CH ₂ CH ₂ OCH ₂	163
1563	—	O	CH ₃	(6-)C ₂ H ₅	O		CH ₂ CH ₂ CH ₂ OCH ₂	191
1564	—	O	CF ₃	(6-)CH ₃	O		OCH ₃	197
1565	—	O	CH ₂ CF ₂ H	(6-)CH ₃	O	CH ₃	OCH ₃	231
1566	—	O	CH ₂ CF ₂ H	(6-)C ₂ H ₅	O	CH ₃	OCH ₃	174
1567	—	O	CH ₂ CF ₂ H	(6-)C ₂ H ₅	O		OCH ₃	160
1568	—	O	CH ₂ CF ₂ H	(6-)C ₂ H ₅	O		OCH(CH ₃) ₂	162
1569	—	O	CH ₂ CF ₂ H	(6-)C ₂ H ₅	O	CH ₃	CH ₃	>250
1570	—	O	CH ₂ CF ₂ H	(6-)CH ₃	O	CH ₃	OCH ₂ CF ₃	117
1571	—	O	CH ₂ CF ₂ H	(6-)CH ₃	O	CH ₃	O—CH ₂ —	128
1572	—	O	CH ₂ CF ₂ H	(6-)CH ₃	O		O—CH ₂ —	124
1573	—	O	CH ₂ CF ₂ H	(6-)CH ₃	O	CH ₃	C ₂ H ₅	137
1574	—	O	CH ₂ CF ₂ H	(6-)CH ₃	O	CH ₃	C ₃ H ₇	139
1575	—	O	CH ₂ CF ₂ H	(6-)CH ₃	O	CH ₃		150
1576	—	O	CH ₂ CF ₂ H	(6-)CH ₃	O	CH ₃	CH ₂ OCH ₃	114
1577	—	O	CH ₂ CF ₂ H	(6-)CH ₃	O	CH ₃	SCH ₃	155
1578	—	O	CH ₂ CF ₂ H	(6-)C ₃ H ₇	O	CH ₃	CH ₃	290
1579	—	O	CH ₂ CF ₂ H	(6-)CH ₃	O	CH ₃	N(CH ₃) ₂	116
1580	—	O	CH ₂ CF ₂ H	(6-)CH ₃	O	OC ₂ H ₅	C ₂ H ₅	125
1581	—	O	CH ₂ CF ₂ H	(6-)CH ₃	O		CH ₃	137

TABLE 1A-continued

Examples of the compounds of the formula (IA)

(IA)

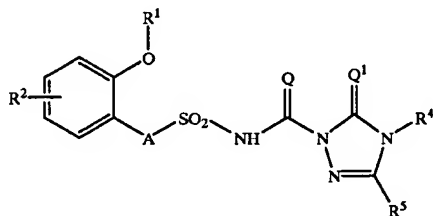


Ex. No.	A	Q	R ¹	(Position-) R ²	Q ¹	R ⁴	R ⁵	Melting point ° C.
1582	—	O	CH ₂ CF ₂ H	(6-)CH ₃	O		OC ₂ H ₅	99
1583	—	O	CH ₂ CF ₂ H	(6-)CH ₃	O		OC ₃ H ₇	130
1584	—	O	CH ₂ CF ₂ H	(6-)C ₂ H ₅	O	CH ₃	CC ₂ H ₅	147
1585	—	O	CH ₂ CF ₂ H	(6-)C ₂ H ₅	O	CH ₃	OCH(CH ₃) ₂	151
1586	—	O	CH ₃	(6-)C ₂ H ₅	O	CH ₃	N(CH ₃) ₂	199
1587	—	O	CF ₃	(6-)C ₂ H ₅	O	CH ₃	CH=CH(CH ₃)	128
1588	—	O	CF ₃	(6-)C ₂ H ₅	O	CH ₃	CH ₂ OCH ₃	101
1589	—	O	CF ₃	(6-)C ₂ H ₅	O		N(CH ₃) ₂	154
1590	—	O	CF ₃	(6-)C ₂ H ₅	O		SC ₂ H ₅	143
1591	—	O	CF ₃	(6-)C ₂ H ₅	O	CH ₃	SC ₂ H ₅	113
1592	—	O	CF ₃	(6-)C ₂ H ₅	O	OC ₂ H ₅	C ₂ H ₅	122
1593	—	O	CF ₃	(6-)C ₂ H ₅	O	OC ₂ H ₅	SC ₂ H ₅	133
1594	—	O	CF ₃	(6-)C ₂ H ₅	O	CH ₃	OC ₆ H ₅	193
1595	—	O	CH ₂ CH ₂ F	(6-)CH ₃	O	CH ₃	C ₂ H ₅	109
1596	—	O	CH ₂ CH ₂ F	(6-)CH ₃	O	CH ₃		157
1597	—	O	CH ₂ CH ₂ F	(6-)CH ₃	O	CH ₃	CH ₂ OCH ₃	106
1598	—	O	CH ₂ CH ₂ F	(6-)CH ₃	O	CH ₃	SCH ₃	135
1599	—	O	CH ₂ CH ₂ F	(6-)CH ₃	O	CH ₃	N(CH ₃) ₂	141
1600	—	O	CH ₂ CH ₂ F	(6-)CH ₃	O	OC ₂ H ₅	C ₂ H ₅	129
1601	—	O	CH ₂ CH ₂ F	(6-)CH ₃	O		CH ₃	169
1602	—	O	CH ₂ CH ₂ F	(6-)CH ₃	O		OC ₂ H ₅	125
1603	—	O	CH ₂ CH ₂ F	(6-)CH ₃	O		OC ₃ H ₇	117
1604	—	O	CH ₂ CH ₂ F	(6-)CH ₃	O	CH ₃	O—CH ₂ —	145
1605	—	O	CH ₂ CH ₂ F	(6-)CH ₃	O		O—CH ₂ —	118
1606	—	O	CH ₂ CH ₂ CH ₂ F	(6-)C ₃ H ₇	O	CH ₃	OCH ₃	127

TABLE 1A-continued

Examples of the compounds of the formula (IA)

(IA)

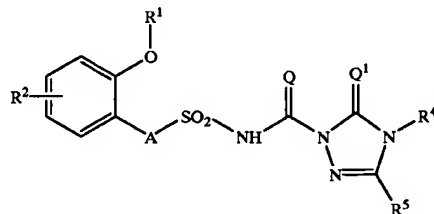


Ex. No.	A	Q	R ¹	(Position-) R ²	Q ¹	R ⁴	R ⁵	Melting point ° C.
1607	—	O	CH ₂ CH ₂ CH ₂ F	(6-)C ₃ H ₇	O	CH ₃	OCH(CH ₃) ₂	155
1608	—	O	CH ₂ CH ₂ CH ₂ F	(6-)C ₃ H ₇	O		OCH ₃	106
1609	—	O	CH ₂ CH ₂ CH ₂ F	(6-)C ₃ H ₇	O		OCH(CH ₃) ₂	138
1610	—	O	CH ₂ CH ₂ F	(6-)C ₂ H ₅	O	CH ₃		157
1611	—	O	CH ₂ CH ₂ F	(6-)C ₂ H ₅	O	CH ₃	CH ₂ OCH ₃	101
1612	—	O	CH ₂ CH ₂ F	(6-)C ₂ H ₅	O		CH ₃	147
1613	—	O	CH ₂ CH ₂ F	(6-)CH ₃	O	CH ₃	C ₃ H ₇	280
1614	—	O	CH ₂ CH ₂ CH ₂ F	(6-)C ₃ H ₇	O	CH ₃	CH ₃	302
1615	—	O	CH ₂ CH ₂ CH ₂ F	(6-)C ₃ H ₇	O	CH ₃	C ₃ H ₇	261
1616	—	O	CH ₂ CH ₂ F	(6-)C ₃ H ₇	O	CH ₃	C ₃ H ₇	108
1617	—	O	CH ₂ CH ₂ F	(6-)C ₃ H ₇	O	CH ₃	OCH ₃	158
1618	—	O	CH ₂ CH ₂ F	(6-)C ₃ H ₇	O	CH ₃	OC ₂ H ₅	152
1619	—	O	CH ₂ CH ₂ F	(6-)C ₃ H ₇	O	CH ₃	OCH(CH ₃) ₂	164
1620	—	O	CH ₂ CH ₂ F	(6-)C ₃ H ₇	O		OCH ₃	137
1621	—	O	CH ₂ CH ₂ F	(6-)C ₃ H ₇	O		OC ₂ H ₅	117
1622	—	O	CH ₂ CH ₂ F	(6-)C ₃ H ₇	O		OCH(CH ₃) ₂	158
1623	—	O	CH ₂ CH ₂ F	(6-)C ₃ H ₇	O	CH ₃	N(CH ₃) ₂	147
1624	—	O	CH ₂ CH ₂ F	(6-)C ₃ H ₇	O	OC ₂ H ₅	C ₂ H ₅	237
1625	—	O	CH ₂ CH ₂ F	(6-)C ₃ H ₇	O	CH ₃	SCH ₃	130
1626	—	O	CHF ₂	(6-)Br	O	CH ₃	OCH ₃	160
1627	—	O	CHF ₂	(6-)Br	O	CH ₃	CH ₃	146
1628	—	O	CHF ₂	(6-)Br	O	CH ₃	C ₃ H ₇	127
1629	—	O	CHF ₂	(6-)Br	O	CH ₃	OC ₂ H ₅	168
1630	—	O	CHF ₂	(6-)Br	O	CH ₃	OCH(CH ₃) ₂	126
1631	—	O	CHF ₂	(6-)Br	O		OCH ₃	145
1632	—	O	CHF ₂	(6-)Br	O		OC ₂ H ₅	125

TABLE 1A-continued

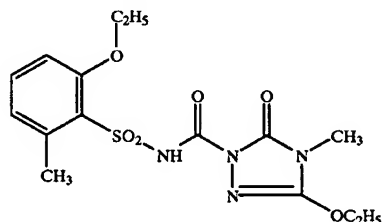
Examples of the compounds of the formula (IA)

(IA)



Ex. No.	A	Q	R ¹	(Position-) R ²	Q ¹	R ⁴	R ⁵	Melting point ° C.
1633	—	O	CHF ₂	(6-)Br	O		OCH(CH ₃) ₂	148
1634	—	O	CHF ₂	(6-)Br	O	CH ₃	N(CH ₃) ₂	173
1635	—	O	CF ₃	(6-)F	O	CH ₃	SCH ₃	
1636	—	O	CF ₃	(6-)F	O	CH ₃	C ₂ H ₅	
1637	—	O	CF ₃	(6-)F	O	CH ₃	OC ₂ H ₅	

The compound listed in Table 1 as Example 9 can be prepared, for example, as follows:



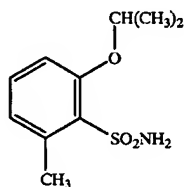
(Process (b))

1.4 g (0.01 mol) of 5-ethoxy-4-methyl-2,4-dihydro-3H-1,2,4-triazol-3-one and 2.4 g (0.01 mol) of 2-ethoxy-6-methyl-phenylsulfonyl isocyanate are stirred at 20° C. For 15 hours in 50 ml of acetonitrile. The solvent is distilled off, the residue is stirred with diethyl ether and the precipitate is filtered off with suction.

3.3 g (85% of theory) of 5-ethoxy-4-methyl-2-(2-ethoxy-6-methyl-phenylsulfonylamino-carbonyl)-2,4-dihydro-3H-1,2,4-triazol-3-one are obtained of melting point 160° C.

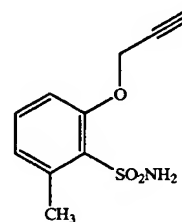
Starting Materials of the Formula (II) or (IIa):

Example (II-1)



64.6 g (0.26 mol) of 2-isopropoxy-6-methyl-benzenesulphochloride are stirred at 20° C. For 12 hours in 350 ml of 25% strength aqueous ammonia solution. The crystalline product is subsequently isolated by filtration with suction. 54 g (90% of theory) of 2-isopropoxy-6-methyl-benzenesulphonamide are obtained of melting point 78° C.

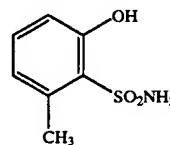
Example (II-2)



A mixture of 1.9 g (10 mmol) of 2-hydroxy-6-methyl-benzenesulphonamide, 1.2 g (10 mmol) of propargyl bromide (in the form of an 80% strength solution in toluene) and 1.4 g (10 mmol) of potassium carbonate is heated under reflux for 2 hours. The mixture is then filtered, the filtrate is concentrated under a water pump vacuum, the residue is digested with petroleum ether and the crystalline product obtained from this digestion is isolated by filtration with suction.

2.1 g (93% of theory) of 6-methyl-2-propargyloxy-benzenesulphonamide are obtained of melting point 129° C.

Example (II-3)



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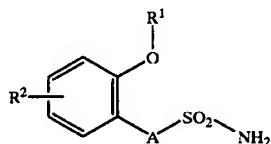
188 ml of a 1-molar solution of boron(III) bromide in methylene chloride are added dropwise at 20° C. with stirring to a solution of 32.3 g (0.15 mol) of 2-ethoxy-6-methylbenzenesulphonamide in 300 ml of methylene chloride, and the reaction mixture is stirred at 20° C. for 30 minutes. Then 300 ml of methanol are added dropwise at from 0° C. to 5° C. (ice cooling). After heating to 20° C., the reaction mixture is concentrated under a water pump vacuum and the residue is stirred with ethyl acetate. The solution obtained is washed with water, dried over sodium sulphate and filtered. The filtrate is concentrated under a water pump vacuum, the residue is crystallized by stirring with petroleum ether, and the crystalline product is isolated by filtration with suction.

20.3 g (72% of theory) of 2-hydroxy-6-methylbenzenesulphonamide are obtained of melting point 126° C.

In analogy to Examples (II-1) to (II-3) and in accordance with the general description of the preparation process according to the invention, it is also possible, for example, to prepare the compounds of the formula (II) or (IIa) listed in Table 2 below,

TABLE 2

Examples of the compounds of the formula (II)

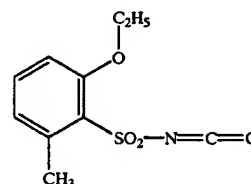


Ex. No.	A	R ¹	(position-) R ²	Melting point (° C.)
II-4	—	C ₂ H ₅	(6-)CH ₃	104
II-5	—	n-C ₃ H ₇	(6-)CH ₃	63
II-6	—	—CH ₂ CH ₂ Cl	(6-)CH ₃	102
II-7	—	CH ₃	(6-)CH ₃	132
II-8	—	—CH ₂ C ₆ H ₅	(6-)CH ₃	131
II-9	—	—CH ₂ COOCH ₃	(6-)CH ₃	90
II-10	—	CH ₃	(6-)C ₃ H ₇ -n	108
II-11	—	C ₂ H ₅	(6-)C ₃ H ₇ -n	80
II-12	—	C ₂ H ₅	(5-)CH ₃	131
II-13	—	CH ₃	(6-)Cl	166
II-14	—	C ₂ H ₅	(6-)Cl	121
II-15	—	H	(6-)Cl	118
II-16	—	i-C ₃ H ₇	(6-)Cl	85
II-17	—	—CH ₂ CH=CH ₂	(6-)Cl	106
II-18	—	—CH ₂ C≡CH	(6-)Cl	181
II-19	—	CF ₃	(5-)Cl	
II-20	—	CHF ₂	(5-)CH ₃	127
II-21	—	CHF ₂	(6-)CH ₃	89
II-22	—	CH ₃	(5-)C(CH ₃) ₃	160
II-23	—	CH ₃	(5-)Cl	
II-24	—	CHF ₂	(4-)CH ₃	153
II-25	—	—CF ₂ CHFCI	(6-)CH ₃	85
II-26	—	C ₂ H ₅	(6-)CH ₂ Cl	

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Starting Materials of the Formula (IV):

Example (IV-1)

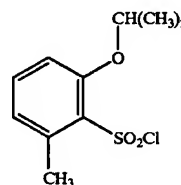


21.5 g (0.1 mol) of 2-ethoxy-6-methylbenzenesulphonamide and 10 g (0.1 mol) of n-butyl isocyanate are heated to boiling in 100 ml of chlorobenzene. At reflux temperature, phosgene is passed in for 4 hours. The clear solution is concentrated under reduced pressure and the residue is subjected to precision distillation. At a pressure of 0.8 bar and an overhead temperature of 135–140° C., 2-ethoxy-6-methyl-phenylsulphonyl isocyanate goes over and solidifies in the receiver.

7.9 g of 2-ethoxy-6-methyl-phenylsulphonyl isocyanate are obtained as a colourless product of melting point 40° C.

Starting Materials of the Formula (VI) or (VIa):

Example (VI-1)



47.8 g (0.29 mol) of 2-isopropoxy-6-methyl-aniline are dissolved in a mixture of 87 ml of 1N hydrochloric acid and 145 ml of conc. hydrochloric acid, and the solution is cooled to -5° C. At -5° C. to 0° C., a solution of 22 g (0.32 mol) of sodium nitrite in 87 ml of water is then added dropwise with stirring and the mixture is stirred at about 0° C. for a further hour. After removal of the nitrite excess with amidosulphonic acid, the diazonium salt solution obtained is added dropwise at -5° C. to 0° C. to a saturated solution of sulphur dioxide in 175 ml of 1,2-dichloro-ethane. After about 30 minutes, 1.7 g of copper(I) chloride and 1.7 g of dodecyl-trimethylammonium bromide are added, and the reaction mixture is allowed to rise to room temperature over the course of about 60 minutes, heated to about 40° C. over a further hour, and stirred at this temperature for about 12 hours. At about 20° C., 14.2 g of a 35% strength hydrogen peroxide solution are then added and the mixture is stirred for about 30 minutes. It is subsequently stirred with 300 ml of methylene chloride, and the organic phase is separated off, washed with water, dried over sodium sulphate and filtered. The solvent is carefully removed from the filtrate by distillation under a water pump vacuum.

65.9 g (90% of theory) of 2-isopropoxy-6-methylbenzenesulphochloride are obtained as a brownish oily residue.

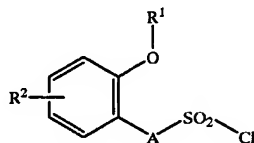
¹H-NMR (CDCl₃, TMS, δ ppm): 1.47 (d, J=6.1 Hz, 2xCH₃); 2.68 (s, CH₃); 4.79 (sept., J=6.1 Hz, 1H); 6.83 (d, J=7.5 Hz, 1H); 6.95 (d, J=8.4 Hz, 1H); 7.45 (pseudo t, J=8.3 Hz, 1H).

In analogy to Example (VI-1) it is also possible, for example, to prepare the compounds of the formula (VI) or (VIa) listed in Table 3 below.

TABLE 3

Examples of the compounds of the formula (VI)

(IV)



Ex. No.	A	R^1	(position-) R^2	Physical data
VI-2	—	CH_3	(6-) CH_3	Fp.: 52° C.
VI-3	—	C_2H_5	(6-) CH_3	1H -NMR ($CDCl_3$, TMS, δ , ppm): 1,55 (t, $J=6,97$ Hz, CH_3), 2,69 (s, CH_3), 4,24 (q, $J=6,97$ Hz, CH_2), 6,87 (d, $J=7,68$ Hz, 1H), 6,95 (d, $J=8,34$ Hz, 1H), 7,46 (pseudo t, $J=8,1$ Hz, 1H)
VI-4	—	$n-C_3H_7$	(6-) CH_3	1H -NMR ($CDCl_3$, TMS, δ , ppm): 1,33 (t, $J=7,38$ Hz, CH_3), 1,95 (m, CH_2), 2,69 (s, CH_3), 4,12 (t, $J=6,3$ Hz, CH_2), 6,86 (d, $J=7,69$ Hz, 1H), 6,94 (d, $J=8,37$ Hz, 1H), 7,46 (pseudo t, $J=7,8$ Hz, 1H)
VI-5	—	$-CH_2CH_2Cl$	(6-) CH_3	1H -NMR ($CDCl_3$, TMS, δ , ppm): 2,71 (s, CH_3), 3,94 (t, $J=6,1$ Hz, CH_2), 4,41 (t, $J=6,1$ Hz, CH_2), 6,96 (t, $J=7,1$ Hz, 2H), 7,5 (t, $J=7,8$ Hz, 1H)
VI-6	—	$-CH_2CH_2OC_2H_5$	(6-) CH_3	1H -NMR ($CDCl_3$, TMS, δ , ppm): 1,23 (t, $J=7$ Hz, CH_3), 2,69 (s, CH_3), 3,65 (q, $J=7$ Hz, CH_2), 3,91 (t, $J=5,16$ Hz, CH_2), 4,30 (t, $J=5,16$ Hz, CH_2), 6,89 (d, $J=7,7$ Hz, 1H), 7,0 (d, $J=8,3$ Hz, 1H), 7,47 (pseudo t, $J=8,1$ Hz, 1H)
VI-7	—	C_2H_5	(5-) CH_3	1H -NMR ($CDCl_3$, TMS, δ , ppm): 1,53 (t, $J=7$ Hz, CH_3), 2,36 (s, CH_3), 4,25 (q, $J=7$ Hz, CH_2), 7,0 (d, $J=8,53$ Hz, 1H), 7,45 (d, $J=8,53$ Hz, $J_2=2,15$ Hz, 1H), 7,75 (d, $J=2,15$ Hz, 1H)
VI-8	—	$n-C_3H_7$	(5-) CH_3	1H -NMR ($CDCl_3$, TMS, δ , ppm): 1,08 (t, $J=7,38$ Hz, CH_3), 1,85 (m, CH_2), 2,23 (s, CH_3), 3,99 (t, $J=6,5$ Hz), 6,74 (d, $J=8,2$ Hz, 1H), 6,92 (m, 1H), 7,34 (d, $J=1,65$ Hz, 1H)
VI-9	—	$i-C_3H_7$	(5-) CH_3	1H -NMR ($CDCl_3$, TMS, δ , ppm): 1,45 (d, $J=6,06$, $2 \times CH_3$), 2,35 (s, CH_3), 4,77 (sept., $J=6,06$ Hz, 1H), 6,99 (d, $J=8,57$ Hz, 1H), 7,43 (dd, $J=8,56$ Hz, 1H, $J_2=2,1$ Hz, 1H), 7,74 (d, $J=2,1$ Hz, 1H)
VI-10	—	C_2H_5	(6-) CH_2Cl	(Oil)
VI-11	—	$-CF_2CH_2Cl$	(6-) CH_3	1H -NMR ($CDCl_3$, TMS, δ , ppm): 2,78 (s, CH_3), 6,46 (td, CH_2Cl), 7,2–7,6 (Ar—H)
VI-12	—	CHF_2	(6-) CH_3	1H -NMR ($CDCl_3$, TMS, δ , ppm): 2,76 (s, CH_3), 6,61 (t, CHF_2), 7,27–7,59 (Ar—H)
VI-13	—	CH_3	(5-) $C(CH_3)_3$	Fp.: 62° C.
VI-14	—	CHF_2	(4-) CH_3	1H -NMR ($CDCl_3$, TMS, δ , ppm): 2,50 (s, CH_3), 6,68 (t, CHF_2), 7,05–7,92 (Ar—H)
VI-15	—	CHF_2	(5-) CH_3	1H -NMR ($CDCl_3$, TMS, δ , ppm): 2,45 (s, CH_3), 6,64 (t, CHF_2), 7,35–7,86 (Ar—H)
VI-16	—	$-CH_2CH_2Cl$	(6-) CH_3	
VI-17	—	$-CH_2CH=CH_2$	(6-) CH_3	
VI-18	—	$-CH_2C \equiv CH$	(6-) CH_3	
VI-19	—	$-CH_2C_6H_5$	(6-) CH_3	